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# New Cassane Diterpenoids from *Caesalpinia sappan* and Their Antiplasmodial Activity

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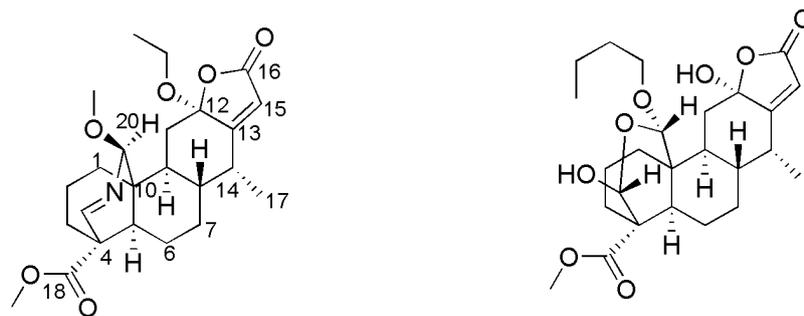
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**Abstract:** One new cassane diterpene possessing an unusual *N* bridge between C-19 and C-20 named caesalsappanin R (**1**), as well as another new diterpene caesalsappanin S (**2**), were isolated from the seeds of *Caesalpinia sappan* with methanol extract. Their structures were determined by spectroscopic analysis and examined alongside existing data from prior studies. Their biological activities were profiled by their antiplasmodial activity.

**Keywords:** *Caesalpinia sappan*; cassane diterpenes; *N* bridge; antiplasmodial activity

## 1. Introduction

*Caesalpinia sappan* has been a part of traditional Chinese herbal medicine and is widely used in the treatment of dysmenorrhea, blood stagnation, and tetanus. Previous phytochemical investigations indicated that this genus contains an abundant source of cassane diterpenes with different structure types, and most of them showed in vitro or in vivo pharmacological impacts such as antiproliferative [1–3], antiplasmodial [4–5], antibacterial [6], antihelminthic, and antineoplastic activity [7]. As a continuation of our project towards new bioactive diterpenes discovery from the genus *Caesalpinia* [2–3,8], we examined the chemical constituents of *C. sappan* and obtained two new cassane diterpenes, designated caesalsappanin R (**1**) and caesalsappanin S (**2**) (Figure 1). Compound **1** is a rather unusual cassane diterpenoid lactone-type skeleton, consisting of an *N* bridge between C-19 and C-20. In this paper, we detail the separation and structural determination of the novel agents and the examination of their antiplasmodial activity.



**Figure 1.** The structures of compounds 1–2.

## 2. Results and Discussion

### 2.1. Purification of Compounds 1–2

The seeds of *C. sappan* were extracted with MeOH three times. The two cassane-type diterpenoids were isolated and purified via silica gel chromatography, Sephadex LH-20 gel chromatography and semi-HPLC.

### 2.2. Structure Elucidation of Compounds 1–2

Compound **1** was acquired as a white shapeless powder. The HRESIMS spectrum demonstrated a quasi-molecular ion at  $m/z$  454.2199 (Calcd for  $C_{24}H_{33}NO_6Na$ , 454.2206), which in connection with the NMR data, confirmed that the molecular formula was  $C_{24}H_{33}NO_6$ . The IR and UV spectra revealed absorptions for an amidogen ( $3190\text{ cm}^{-1}$ ), a carbonyl ( $1735\text{ cm}^{-1}$ ), and an  $\alpha,\beta$ -unsaturated butenolide unit ( $210\text{ nm}$ ;  $1749\text{ cm}^{-1}$ ) [2]. The  $^1\text{H}$  and  $^{13}\text{C}$  APT NMR spectra (Table 1) displayed the olefinic proton signal at  $\delta_{\text{H}}$  5.86 (H-15, s) and four downfield-shifted carbon signals at  $\delta_{\text{C}}$  107.4 (C-12), 171.0 (C-13), 115.9 (C-15), and 179.9 (C-16), which also confirmed the presence of the  $\alpha,\beta$ -unsaturated butenolide ring. Additionally, the  $^1\text{H}$  NMR spectrum exhibited signals for a methyl at  $\delta_{\text{H}}$  1.14 (d,  $J = 7.2\text{ Hz}$ , H<sub>3</sub>-17), two methoxys at  $\delta_{\text{H}}$  3.74 (s, 18-OCH<sub>3</sub>) and 3.72 (s, 20-OCH<sub>3</sub>), an ethoxy group at  $\delta_{\text{H}}$  3.30, 3.58 (m, OCH<sub>2</sub>CH<sub>3</sub>) and 1.21 (d,  $J = 7.2\text{ Hz}$ , OCH<sub>2</sub>CH<sub>3</sub>), a nitrogen oxymethylene proton at  $\delta_{\text{H}}$  5.07 (d,  $J = 2.4\text{ Hz}$ ), and a nitrogen alkenyl at  $\delta_{\text{H}}$  7.53 (s). Except for the methoxy ( $\delta_{\text{C}}$  52.2, 57.1) and ethoxy ( $\delta_{\text{C}}$  59.3, 15.0) substituents, the  $^{13}\text{C}$  APT NMR spectrum showed 20 carbons including one methyl ( $\delta_{\text{C}}$  12.2), six methylenes ( $\delta_{\text{C}}$  19.3, 25.6, 29.1, 30.2, 33.1, and 37.2), seven methines ( $\delta_{\text{C}}$  37.0, 42.2, 43.4, 47.0, 91.2, 115.9, and 169.9), and six quaternary carbons ( $\delta_{\text{C}}$  44.1, 49.8, 107.4, 169.9, 171.0, and 175.3). The HSQC spectrum displayed all of the proton signals assigned to the corresponding carbons through direct  $^1\text{H}$  and  $^{13}\text{C}$  correlations. The overall  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopic data confirmed that **1** is an oxynitride diterpene possessing a fused butenolide unit [9,10], and its entire structure was connected, as confirmed using HSQC, HMBC, and  $^1\text{H}$ - $^1\text{H}$ -COSY spectra (Figure 2). The nitrogen oxymethylene proton at  $\delta_{\text{H}}$  5.07 (d,  $J = 2.4\text{ Hz}$ , H-20) showed long-range correlations with carbons at  $\delta_{\text{C}}$  30.2 (C-1), 49.8 (C-10), 162.6 (C-19), and 57.1 (20-OCH<sub>3</sub>), which suggested that C-1, C-10, C-19, and -OCH<sub>3</sub> were connected through the nitrogen oxymethylene carbon C-20. The quaternary carbon C-4 ( $\delta_{\text{C}}$  44.1) was connected to C-3 ( $\delta_{\text{C}}$  33.1), C-5 ( $\delta_{\text{C}}$  47.0), C-18 ( $\delta_{\text{C}}$  175.3), and C-19 ( $\delta_{\text{C}}$  162.6) due to the HMBC correlations of H-19, H<sub>2</sub>-3, and H-5 to C-4 and C-18. Moreover, the nitrogen bridge between C-19 and C-20 was confirmed by the downfield chemical shifts of C-19 ( $\delta_{\text{C}}$  162.6) and C-20 ( $\delta_{\text{C}}$  91.2) together with the HMBC correlations between H-19 and C-20. Finally, the  $\alpha,\beta$ -unsaturated butenolide moiety was connected to C-11 and C-14 based on the HMBC correlations from H<sub>2</sub>-11 to C-12 ( $\delta_{\text{C}}$  107.4) and H-14 to C-13 ( $\delta_{\text{C}}$  171.0). The proton H<sub>3</sub>-17 ( $\delta_{\text{H}}$  1.14, d,  $J = 7.2\text{ Hz}$ ) showed long-range correlations with carbons C-14 ( $\delta_{\text{C}}$  37.0), which indicated that the methyl group of C-17 was connected to C-14. The methoxyl and ethoxyl groups were attached to C-18 and C-12, respectively, based on the HMBC correlations between  $\delta_{\text{H}}$  3.74 (s, -OCH<sub>3</sub>) and  $\delta_{\text{C}}$  175.3 (C-18),  $\delta_{\text{H}}$  3.30, 3.58 (m, OCH<sub>2</sub>CH<sub>3</sub>) and  $\delta_{\text{C}}$  107.4 (C-12). The NOESY experiment established the relative configuration of compound **1**, the correlations of H-20 ( $\delta_{\text{H}}$  5.07)/H-1 $\alpha$  ( $\delta_{\text{H}}$  1.69–1.72), H<sub>3</sub>-17 ( $\delta_{\text{H}}$  1.14)/H-9 ( $\delta_{\text{H}}$  1.78), OCH<sub>2</sub>CH<sub>3</sub>-12 ( $\delta_{\text{H}}$  3.30)/H<sub>3</sub>-17 ( $\delta_{\text{H}}$  1.14)

showed that the hydroxyl group was  $\beta$ -oriented at C-20, and the methyl group at C-14 and the ethoxy group at C-12 were all  $\alpha$ -oriented. The same carbon skeleton with the *trans/anti/trans* system of three six-membered rings A, B, and C, and the oriented proton at C-8 was  $\beta$ -axial and the oriented protons at C-5/C-9 was  $\alpha$ -axial, which are well established on all cassane diterpenes isolated so far from the genus *Caesalpinia* [3,8,11]. Considering the biosynthetic relationship and comparing with the literature of cassane diterpenoids [12], the absolute configurations of the chiral carbons were determined to be 4*S*, 5*R*, 8*S*, 9*S*, 10*S*, 12*S*, 14*R* in **1** and are shown in Figure 2. Therefore, the structure of **1** was determined and it was named caesalsappanin R (Figure 1). Compound **1** is representative of a new cassane diterpenoid lactone-type skeleton with an *N* bridge between C-19 and C-20.

**Table 1.** NMR spectral data of **1–2** (CDCl<sub>3</sub>, 600 and 150 MHz).

No.	<b>1</b>		<b>2</b>		<b>Caesalsappanin H</b>
	$\delta_c$ , Type	$\delta_H$ (J in Hz)	$\delta_c$ , Type	$\delta_H$ (J in Hz)	$\delta_c$ , Type
1	30.2 CH <sub>2</sub>	1.69–1.72 (m) 2.17–2.21 (m)	37.7 CH <sub>2</sub>	1.28–1.30 (m) 2.07–2.09 (m)	37.8 CH <sub>2</sub>
2	19.3 CH <sub>2</sub>	1.38–1.41 (m) 2.59–2.63 (m)	20.6 CH <sub>2</sub>	1.58–1.60 (m) 2.23–2.25 (m)	20.6 CH <sub>2</sub>
3	33.1 CH <sub>2</sub>	1.28–1.32 (m) 1.89–1.93 (m)	28.5 CH <sub>2</sub>	1.82–1.83 (m) 2.25–2.30 (m)	28.6 CH <sub>2</sub>
4	44.1 C		50.3 C		50.4 C
5	47.0 CH	1.73 (m)	47.2 CH	1.68–1.71 (m)	47.2 CH
6	25.6 CH <sub>2</sub>	1.18–1.20 (m) 1.39–1.42 (m)	24.2 CH <sub>2</sub>	1.19–1.21 (m) 2.00–2.02 (m)	24.2 CH <sub>2</sub>
7	29.1 CH <sub>2</sub>	1.69–1.72 (m) 2.19–2.23 (m)	29.5 CH <sub>2</sub>	1.25–1.28(m) 1.60–1.62 (m)	29.5 CH <sub>2</sub>
8	43.4 CH	1.49 (m)	41.5 CH	2.19–2.21 (m)	41.1 CH
9	42.2 CH	1.78 (m)	41.3 CH	1.51–1.53 (m)	41.3 CH
10	49.8 C		38.6 C		38.7 C
11	37.2 CH <sub>2</sub>	1.68–1.70 (m) 2.75 (dd, 12.0,2.4)	38.0 CH <sub>2</sub>	1.36–1.38 (m) 2.51–2.53 (m)	38.1 CH <sub>2</sub>
12	107.4 C		105.5 C		105.9 C
13	171.0 C		173.4 C		173.7 C
14	37.0 CH	2.99 (qd, 7.2, 2.4)	37.1 CH	2.91 (qd, 7.2, 2.4)	37.1 CH
15	115.9 CH	5.86 (s)	113.5 CH	5.69 (s)	113.8 CH
16	169.9 C		170.7 C		170.7 C
17	12.2 CH <sub>3</sub>	1.14 (d, 7.2)	12.0 CH <sub>3</sub>	1.13 (d, 7.2)	12.1 CH <sub>3</sub>
18	175.3 C		175.6 C		175.5 C
19	162.6 CH	7.53, s	90.1 CH	5.60 (s)	90.1 CH
20	91.2 CH	5.07 (d, 2.4)	104.2 CH	4.49 (s)	105.4 CH
		3.30 (m)			
	59.3, CH <sub>2</sub>	3.58 (m)			
OCH <sub>2</sub> CH <sub>3</sub> -12	15.0, CH <sub>3</sub>	1.21 (t, 7.2)			
OCH <sub>3</sub> -18	52.2	3.74 (s)	52.0	3.71 (s)	51.7
OCH <sub>3</sub> -20	57.1	3.72 (s)			55.7
OCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> -20			67.8	3.22 (ddd, 9.6, 6.0, 3.0) 3.80 (ddd, 9.6, 6.0, 3.0)	
OCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> -20			31.9	1.32 (m) 1.47 (m)	
OCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> -20			20.0	1.28 (m) 1.43 (m)	
OCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> -20			13.7	1.87 (t, 7.2)	



spectrometer (Thermo Fisher Scientific, Boston, MA, USA). Column–chromatography (CC) was performed using silica gel (100–200 and 300–400 mesh, Qingdao Marine Chemical Plant, Qingdao, China), Sephadex LH-20 (Pharmacia, Uppsala, Sweden). Precoated silica gel GF<sub>254</sub> plates (Zhi Fu Huang Wu Pilot Plant of Silica Gel Development, Yantai, China) were used for TLC. All solvent used was of analytical grade (Beijing Chemical Plant, Beijing, China).

### 3.2. Plant Material

The seeds of *C. sappan* were collected from Nanning, Guangxi Province, People's Republic of China, in April 2013, and identified by Professor Jing Quan Yuan of the Department of Pharmaceutical Chemistry, Guangxi Botanical Garden of Medicinal Plants. A voucher specimen (NO. 13418) was deposited at the Guangxi Botanical Garden of Medical Plants.

### 3.3. Isolation and Purification of Compounds 1–2

The air-dried seeds of *C. sappan* (5.0 kg) were extracted with MeOH (3 × 40 L, 3 h each) at room temperature. Removal of the MeOH under reduced pressure yielded a MeOH-soluble extract (1267 g). The residue was suspended to H<sub>2</sub>O (3 L) and partitioned with petroleum ether (3 × 3 L), CH<sub>2</sub>Cl<sub>2</sub> (3 × 3 L), EtOAc (3 × 3 L), and n-BuOH (3 × 3 L), successively. The EtOAc fraction (164 g) was subjected to CC (column–chromatography) over silica gel (100–200 mesh, 15 × 60 cm) eluting with a stepwise gradient of CH<sub>2</sub>Cl<sub>2</sub>-MeOH (from 1:0 to 0:1, 100:0, 90:1, 70:1, 50:1, 30:1, 20:1, 10:1, 5:1, 2:1, 1:1, 0:1, *v/v*) to afford fractions A-G. Fraction E (3.1 g) was subjected to chromatographed repeatedly over silica gel CC eluting with CH<sub>2</sub>Cl<sub>2</sub>-MeOH (50:0, 40:1, 30:1, 20:1, 10:1, *v/v*) to obtained sub-fractions Fractions E1-E5. Fraction E3 was separated using silica gel CC eluting with CH<sub>2</sub>Cl<sub>2</sub>-MeOH (40:1, 30:1, 0:100, *v/v*) to obtained sub-fractions I-III. Sub-fraction II was purified by semi-HPLC of MeOH-H<sub>2</sub>O (55:45, *v/v*) as the mobile phase to yield compound **1** (6.3 mg, 0.000146%, *t<sub>R</sub>* = 28.4 min). Fraction E2 was separated using silica gel CC eluting with CH<sub>2</sub>Cl<sub>2</sub>-MeOH (50:1, *v/v*), yielding compound **2** (8.6 mg, 0.000172%).

### 3.4. Characterization of Compounds 1–2

caesalsappanin R (**1**), White powder (MeOH); [ $\alpha$ ]<sub>D</sub><sup>20</sup> -24.2 (*c* = 0.05, MeOH); UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 210 (3.86) nm; IR (film)  $\nu_{\max}$  3190, 1745, 1735 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C-NMR data (CDCl<sub>3</sub>), see (Table 1); HR-ESI-MS *m/z* 454.2199 [M + Na]<sup>+</sup>. (Calcd for. 454.2206 C<sub>24</sub>H<sub>33</sub>NO<sub>6</sub>Na).

caesalsappanin S (**2**), White powder (MeOH); [ $\alpha$ ]<sub>D</sub><sup>20</sup> -47.3 (*c* = 0.1, MeOH); UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 213 (3.94) nm; IR (film)  $\nu_{\max}$  3450, 1730 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C-NMR data (CDCl<sub>3</sub>), see (Table 1); HR-ESI-MS *m/z* 487.2332 [M + Na]<sup>+</sup>. (Calcd for. 487.2308 C<sub>24</sub>H<sub>34</sub>O<sub>8</sub>Na).

### 3.5. Antiplasmodial Assays of Compounds 1–2

Antiplasmodial activity in vitro was determined by means of the microculture radioisotope technique based on the method described by Desjardins [13]. The parasite *P. falciparum* (K1, multidrug-resistant strain) was cultured continuously according to the method of Trager and Jensen [14]. Three preparations were used for each experiment. The determination of IC<sub>50</sub> values against the erythrocytic stages of *P. falciparum* was carried out in duplicate using the [<sup>3</sup>H]-hypoxanthine incorporation assay [15]. Laboratory colonies of mosquito larvae/pupae (*Culex quinquefasciatus* Say, Diptera, Culicidae) were used for the larvicidal/pupicidal activity. Twenty-five numbers of first to fourth instars larvae and pupae were introduced into 500 mL glass beaker containing 249 mL of de-chlorinated water and 1 mL of desired concentrations of ethanolic leaf extract were added. Larval food was given for the test larvae. At each tested concentration two to five trials were made and each trial consisted of five replicates. The control was setup by mixing 1 mL of acetone with 249 mL of dechlorinated water. The larvae and pupae were exposed to dechlorinated water without acetone served as control. The control mortalities were corrected by using Abbott's formula [16,17]. The LC<sub>50</sub> were calculated from toxicity data by using probit analysis [18]. Chloroquine was included as a standard for comparison. Data are presented as means ± SEM.

Statistical analyses were done by means of the Student's *t*-test. A *P* value of less than 0.05 was considered a significant difference.

#### 4. Conclusions

In conclusion, two new cassane-type diterpenoids (**1** and **2**) were isolated and characterized by spectrometric analysis (1 and 2D NMR, HRESIMS). Compound **1** exhibited active antiplasmodial activity in vitro with IC<sub>50</sub> at 3.60 μM. In addition, the compounds that we had reported also showed antiplasmodial activities; caesalsappanins A, G, H, and I displayed antiplasmodial activities with IC<sub>50</sub> values of 7.4, 0.78, 0.52, and 2.5 μM, respectively [3]. Therefore, we believe that this plant is an important source for the diverse structure of cassane-type diterpenoids and should be further investigated for the antiplasmodial activity.

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**Conflicts of Interest:** There is no conflict of interest associated with the authors of this paper.

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**Sample Availability:** Samples of the compounds ..... are available from the authors.