

1 Article

# 2 Ucuùba (*Virola surinamensis*) Fat-based 3 Nanostructured Lipid Carriers for Nail Drug Delivery 4 of Ketoconazole: Development and Optimization 5 Using Box- Behnken Design

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19 **Abstract:** Ucuùba fat is fat obtained from a plant found in South America, mainly in Amazonian  
20 Brazil. Due to its biocompatibility and bioactivity, the Ucuùba fat was used for production of  
21 ketoconazole-loaded nanostructured lipid carriers (NLC) in view of an application for the treatment  
22 of onychomycosis and other persistent fungal infections. The development and optimization of the  
23 Ucuùba fat based NLC were performed using a Box-Behnken design of experiment. The  
24 independent variables were surfactant concentration (% w/v), liquid lipids concentration (% w/v),  
25 solid lipids concentration (% w/v), while the outputs of interest were particle size, polydispersity  
26 index (PDI) and drug encapsulation efficiency (EE). The Ucuùba fat based NLC were produced and  
27 the process optimized determining a predictive mathematical model. Applying the model, two  
28 formulations with the pre-required particle size, *i.e.* 30 and 85 nm, were produced for further  
29 evaluation. The optimized formulations were characterized and showed a particle size in agreement  
30 to the predicted value, *i.e.* 33.6 nm and 74.6 nm, respectively. The optimized formulations were also  
31 characterized using multiple techniques in order to investigate the solid state of drug and excipients  
32 (DSC and XRD), particle morphology (TEM) and interactions between the formulation components  
33 (FTIR). Furthermore, particle size and surface charge of the formulations was studied during a one-  
34 month stability study and did not evidence any significative modification during storage.

35 **Keywords:** Amazonian fat; Ucuùba fat; Box Behnken Design; Solid lipid nanoparticles; Antifungal  
36 therapy; Onychomycosis.

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## 38 1. Introduction

39 Onychomycosis is the most common nail disease of nail plate and nail bed [1–3]. Several  
40 antifungal agents are used in the treatment of onychomycosis, among them ketoconazole, an  
41 imidazole derivative that acts by blocking the synthesis of ergosterol. Oral treatment of  
42 onychomycosis with ketoconazole provides a cure rate of 50 to 70% in fingernail and 15 to 30% in  
43 toenail infections [4,5]. However oral treatment has several disadvantages, such as long duration,  
44 drug interactions, systemic side effects, and high rates of recurrence. In this regard, topical treatment

45 appears advantageous and in fact several topical preparations are on the market containing 2%  
46 ketoconazole in the form of gel, cream or lotions. However, the topical treatment available has the  
47 disadvantage of low permeation of the drug through the nail [4,6,7].

48 Nanostructured lipid carriers (NLC) are colloidal dispersions in which the dispersed phase is  
49 composed of a mixture of both solid and liquid lipids stabilized by an emulsifier [8,9]. Lipid  
50 nanoparticles have been extensively used for topical application, since their lipid matrix can interact  
51 with the lipids of the *stratum corneum*, promoting drug delivery to the skin. Moreover, the occlusive  
52 film formed by lipid nanoparticles on the skin surface significantly increases tissue hydration and  
53 drug skin permeation [10–12]. It has been shown that lipid nanocarriers are also capable of  
54 increasing nail plate hydration enhancing drug penetration [13].

55 The solid lipid selected for production NLC has a pivotal role, because it affects a number of  
56 physicochemical and biopharmaceutical properties of the particles (surface charge, melting point,  
57 drug loading and release properties among others) [14,15]. Plant derived fats are a viable option,  
58 since they are made of essential fatty acids important for maintaining the homeostatic balance of the  
59 skin and its annexes [16,17]. Ucuùba butter (*Virola surinamensis*) is extracted from the seeds of Ucuùba,  
60 a tree found mainly in the Amazon river basin, but present in an area that encompasses all of Central  
61 America and part of South America. Traditional uses of this fat span from the production of candles  
62 to medical soaps, but Ucuùba butter can be found also as a component of moisturizing creams and  
63 shampoos in the Brazilian cosmetic market [16,18]. The main fatty acids present in Ucuùba fat are  
64 myristic and lauric acid, while oleic, palmitoleic, linoleic acids are present at lower concentrations  
65 [19]. In addition, the presence in the leaves, root, bark and seeds of *V. surinamensis* of  
66 phenylpropanoids, propiophenones, lignans, neolignans, polyketides and flavonoids with anti-  
67 oxidant, anti-inflammatory and anti-mycotic action has been reported in other studies and elicited  
68 interest in the pharmaceutical properties of this fat [20–22].

69 In several studies related to the optimization of NLC production it is observed that the choice of  
70 the relative concentration of surfactant and lipids strongly affect the stability of the formulation [23]  
71 . Furthermore, also the physical-chemical characteristics of NLC, such as size, entrapment efficiency  
72 (EE), physical stability, drug release, morphology, are strongly influenced by the choice and amount  
73 of lipids and surfactant [23–25]. Thus, an optimization of NLC production with the aid of design of  
74 experiment could be carried out with the objective of modulating the amount of liquid and solid  
75 lipids and surfactant to obtain the most desirable properties in terms particle size, entrapment  
76 efficiency and stability. Response surface method (RSM) is one of the prevalent approaches in design  
77 of experiments, which involves the use of different type experimental designs to generate polynomial  
78 mathematical relationships mapping of the response over the experimental domain with the aim to  
79 predict the outcome of a process or select the optimal process parameters [26]. Box-Behnken design  
80 (BBD) is one of RSM approaches, which is an independent, rotatable or nearly rotatable, quadratic  
81 design having the process variable combinations at the midpoints of the edges of the process space  
82 and its centre. A significant advantage of the Box-Behnken design is that it is more cost effective if  
83 compared with the other techniques such as Central Composite Design, 3-levels factorial design and  
84 D-optimal design, as it requires fewer experimental runs and less time for process optimisation  
85 [26,27].

86 We hypothesized that Ucuùba fat, on account of its physicochemical and biological properties,  
87 could be a suitable lipid excipient for the production of NLC loaded with ketoconazole. Thus, the  
88 objective of this work was to obtain and optimize ketoconazole-loaded NCL production by high  
89 pressure optimization using Ucuùba fat in oily phase. The Design of Experiment was then applied to  
90 select the optimal conditions for the production of Ucuùba fat-based NLC in terms of encapsulation  
91 efficiency (EE), particle size and polydispersity index (PDI) in view of their use as a nail drug delivery  
92 system.

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## 97 2. Materials and Methods

### 98 2.1. Materials

99 Virgin Ucuùba fat (*V. surinamensis*) was provided by Amazon Oil Industry (Ananindeua, Brazil).  
100 D- $\alpha$ -Tocopheryl polyethylene glycol 1000 succinate (TPGS 1000) was supplied by ISOICHEM  
101 (Gennevilliers, France). Propylene glycol monocaprylate (type II) (Capryol™ 90) acquired Gattefossé  
102 (Saint Priest, France). Ketoconazole was purchased Galena (São Paulo, Brazil). All other reagents  
103 were of analytical purity grade. Ultrapure (0.055  $\mu$ S/ cm, TOC 1 ppb) was obtained with a Purelab  
104 Pulse + Flex Ultrapure water system (Elga Veolia, Milan, Italy).

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### 107 2.2. Ucuùba fat-based NLC preparation

108 Ucuùba fat NLC were prepared using an emulsification, homogenization and solidification  
109 technique. The lipid phase was prepared by melting Ucuùba fat at 10°C above the lipid melting point  
110 and by mixing the liquid lipid with TPGS 1000, Capryol™ 90 and ketoconazole (0.5% w/v). Then, the  
111 aqueous phase, preheated at 70°C, was added gradually into the lipid phase under continuous  
112 magnetic stirring. Subsequently, a pre-emulsion was prepared using (Ultra-Turrax® TP18/10, IKA  
113 Werke GmbH&Co., Staufen, Germany) operated at 10,000 rpm for 1 minute. Finally, the pre-emulsion  
114 was processed using laboratory High Pressure Homogenizer (HPH) (Panda Plus 2000, GEA Niro  
115 Soavi S.p.A., Parma, Italy) at 70°C for 10 successive cycles at the pressure of 500 bar. Nanoparticles  
116 were then left to cool to ambient temperature.

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### 118 2.3. Experimental design

119 A three-level three-factor (3<sup>3</sup>) Box-Behnken design (BBD) was applied for NLC production  
120 optimization process using the statistical package Desing-Expert® Software (version 11, Stat-Ease Inc.,  
121 Minneapolis, MN, USA). To this purpose, we selected three input factors, or critical process  
122 parameters (CPPs), namely: the concentration of the surfactant (TPGS), the concentration of liquid  
123 lipid (Capryol 90) and the concentration of the solid lipid (Ucuùba fat). These input factors were set  
124 at three levels, i.e. low, middle and high values, indicated conventionally by -1, 0 and +1, respectively.  
125 The design, including five repetitions of the central point, included a total of 17 experiments. The  
126 measured responses, or critical quality attributes (CQAs), were particle size ( $Y_1$ ), polydispersity index  
127 (PDI) ( $Y_2$ ) and encapsulation efficiency (EE) ( $Y_3$ ) (see **Table 1**).

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129 For each CQA, the influence of the input factors and their interactions on the responses can be  
130 described for the non-linear quadratic model generated by the design was:

$$Y = A_0 + A_1X_1 + A_2X_2 + A_3X_3 + A_4X_1X_2 + A_5X_2X_3 + A_6X_1X_3 + A_7X_1^2 + A_8X_2^2 + A_9X_3^2, \quad (1)$$

131 in which  $Y$  is the predicted response,  $A_0$  is intercept,  $A_1$ - $A_9$  are the regression coefficients values.  
132  $X_1$ ,  $X_2$  and  $X_3$  are codes for independent variables, concentration of the surfactant, concentration of  
133 liquid lipid and the concentration of lipid solid, respectively. The terms ( $X_1X_2$ ,  $X_2X_3$  and  $X_1X_3$ ) and ( $X_i^2$ ,  
134 where  $i=1,2$  and 3) represent the factors interactions and quadratic terms respectively.

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147**Table 1** Input factors and their levels for Box-Behnken design

Input factors	Levels		
	-1	0	+1
Surfactant Concentration (% w/v, X <sub>1</sub> )	3	4.5	6
Liquid Lipid Concentration (% w/v, X <sub>2</sub> )	2	2.5	3
Solid Lipid Concentration (% w/v, X <sub>3</sub> )	7	7.5	8

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#### 2.4. Ucuùba fat-based NLC Characterization

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##### 2.4.1. Particle Size Distribution

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##### 2.4.2 Zeta Potential

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##### 2.4.2 Encapsulation Efficiency (EE)

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Ucuùba fat-based NLC were characterized by dynamic light scattering using a Zetasizer Nano ZS (Malvern Pananalytical, Malvern, UK) applying Non-Invasive Back-Scatter (scattering angle 173°) at the temperature of 25°C. NLC were diluted (1:100) with ultrapure water prior to measurements. Optimized NLC were at different days (0, 1, 7, 15 and 30 days). Average particle size (Z-average) and polydispersity index (PDI) obtained through the cumulants analysis of scattering data by the instrument software were measured in triplicate and results were expressed as mean and standard deviation.

The zeta potential, representative of particle surface electrostatic charge, was determined by electrophoretic mobility (Zetasizer Nano ZS, Malvern, Pananalytical, Malvern, UK). The samples were diluted in 10 mM NaCl (1:500). Helmholtz-Smoluchowski equation was employed for calculation of zeta potential by the instrument software.

The content of ketoconazole in nanoparticles was determined by the high-performance liquid chromatography (HPLC). The analysis was carried out on a Shimadzu chromatography system consisting of an autosampler (Model 542, ESA Biosciences, Chelmsford, MA, USA), a pump (LC-10AS, Shimadzu, Kyoto, Japan) and an UV-VIS detector (SPD-10 A, Shimadzu, Kyoto, Japan). A reverse phase C18 column (YMC-Pack ODS-AQ, 250 x 4.6 mm, particle size 5µm, pore size 120 Å, YMC Co. Ltd., Kyoto, Japan) was employed for chromatographic separation in combination with a mobile phase consisting of a 75:25 (v/v) mixture of 0.5% w/v triethanolamine in methanol and 0.2% w/v ammonium acetate (pH 5). HPLC analyses were carried out at a flow rate 0.8 mL/min, detection wavelength 240 nm at 25°C. Sample injection volume was 20µL and sample elution time 10 minutes.

Linearity of response in calibration curves for ketoconazole was verified in the range of 1–3 µg/mL ( $r^2=0.999$ ). Limit of detection and limit of quantification were 0.18 and 0.56 µg/mL, respectively.

The encapsulation efficiency (EE) of nanoparticles was determined by an indirect method, that is the amount of ketoconazole precipitated or present in agglomerates was quantified and subtracted from the total amount of drug present in the whole NLC dispersion. Firstly, a precise volume of Ucuùba fat-based NLC preparation was dispersed in the mobile phase in order to extract the drug and quantitate the total drug present in the formulation by HPLC analysis. The second step consisted in separating and determine the precipitated drug from the encapsulated drug. In order to separate the drug present as crystals or particle agglomerates, the formulation was centrifuged at 9,500×g at 25°C for 15 min (Model D3024, Scilogex, USA). The supernatant was discarded without troubling the sedimented drug present as agglomerate or precipitated drug. The pellet was then solubilized in

188 mobile phase and analysed by HPLC. The EE were calculated using Equation (1), also considering  
189 the poor aqueous solubility of ketoconazole [28].

$$190 \quad EE (\%) = \frac{(Total\ drug - Precipitated\ drug)}{Total\ drug} \times 100 \quad (1)$$

### 192 2.4.3. Morphology

193 Transmission Electron Microscopy (TEM) analysis of the NCL produced was carried out using  
194 a JEM 2200-FS microscope (JEOL Ltd., Tokyo, Japan) operated at 80 kV. For sample preparation, a  
195 drop of the suspension was deposited on formvar/carbon coated copper grids (300 mesh, Electron  
196 Microscopy Sciences, Holfield, PA, USA), after 60 seconds the excess was gently dried with filter  
197 paper and grid was stained using a drop of a 2% w/v solution of uranyl acetate (Sigma Aldrich, St.  
198 Louis, MO, USA) for 120 seconds. The staining solution was gently eliminated with filter paper and  
199 the grid rapidly dipped in particle free ultrapure water to further eliminate loosely bound material  
200 and excess staining residuals. The images were processed with Digital Micrograph software (Gatan  
201 Inc., Pleasanton, Ca, USA).

### 203 2.4.4. Differential scanning calorimetry (DSC)

204 Differential scanning calorimetry was performed using a Mettler-Toledo DSC823e equipment  
205 (Schwerzenbach, Switzerland). All the samples NLC, ketoconazole and excipients (mixtures Capryol  
206 90 and Ucuùba fat 3:1, 1:1 and 1:2) were prepared introducing 1- 5 mg of sample in aluminium pans  
207 sealed and double pierced. Thermal scans were recorded from 15 to 100°C for the lipid mixtures  
208 and from 15 to 200°C for the NLC samples at a heating rate of 10°C/ min under dry nitrogen purge  
209 (80 mL/ min).

210 The samples of the NLC were air dried for 4h at room temperature for DSC analysis. Lipid  
211 mixtures were magnetically stirred at 200 rpm for 1h at 70°C for complete homogenization between  
212 lipids. The degree of crystallinity of mixtures of virgin Ucuùba fat and Capryol 90 was determined  
213 by calculating the crystallinity index (CI) from the heat of fusion using Equation (2) [29]:

$$214 \quad CI(\%) = \left( \frac{\Delta H_{mix}}{\Delta H_{PL}} \right) \cdot f_{mix} \cdot 100 \quad (2)$$

215 where  $\Delta H_{mix}$  is the enthalpy of fusion of the lipid mixture,  $\Delta H_{PL}$  is the fusion enthalpy of the pure  
216 lipid, and  $f_{mix}$  is the weight fraction of the solid lipid in the mixture.

### 217 2.4.5 X- ray diffraction (XRD)

218 X-ray diffraction (XRD) of the NLC formulation and the Ucuùba fat was performed using a  
219 MiniFlex X-Ray Diffractometer (Rigaku, Japan) using Cu  $K\alpha$  radiation ( $\lambda = 1.5418 \text{ \AA}$ ) generated with  
220 30 kV. About 200 mg of air dried NLC formulation and Ucuùba fat were loaded on the sample holder  
221 until it was completely full and then pressed with a glass slide in order to obtain a flat and  
222 homogeneous surface. The samples were analyzed from a starting angle  $\theta$  of 2° to an end angle of  
223 35° with a scanning rate of 1.5 ° min<sup>-1</sup> (step size 0.5°).

### 224 2.4.6 Infrared Spectroscopy

225 The absorption spectra of the excipients, *i.e.* Capryol 90, Ucuùba fat and ketoconazole, and NLC  
226 formulations, were analyzed using a Shimadzu spectrophotometer (IR Prestige 21, Kyoto, Japan) in  
227 the range 4,000-1,000 cm<sup>-1</sup>. The samples were dripped in tablets of potassium bromide and spectra  
228 recorded using 32 scans and 2 cm<sup>-1</sup> resolution at room temperature [30].

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238 **3. Results**239 *3.1. Design of Experiment*

240 In this study, a Box-Behnken design was applied in order to evaluate the interactions and the  
 241 main quadratic effects of selected at different levels on three critical quality attributes, *i.e.* NLC  
 242 particle size, polydispersity index and encapsulation efficiency. **Table 2** the Box- Behnken design  
 243 results are summarized.

244 The mean particle size (response  $Y_1$ ) ranged from 23.94 nm (Experiment 16) to 90.13 nm  
 245 (Experiment 08) depending on variables level selected during production in **Table 2**. PDI values  
 246 (response  $Y_2$ ) ranged from 0.248 (Experiment 7) to 0.558 (Experiment 2) indication that there were  
 247 formulations with distributions from mostly unimodal to multimodal size distributions. The  
 248 encapsulation efficiency (EE, response  $Y_3$ ) varied from 93.91% (Experiment 1) to 99.66% (Experiment  
 249 5) (see **Table 2**). The ratio between maximum and minimum values for responses  $Y_1$ ,  $Y_2$  and  $Y_3$  was  
 250 found to be 3.76, 2.25 and 1.06, indicating no requirement of power transformation of values.

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252 **Table 2.** Results of the experiments of the Box-Behnken design

Experiment #	Surfactant (% w/v)	Liquid Lipid (% w/v)	Solid Lipid (% w/v)	Size (nm)	PDI <sup>1</sup>	Ketoconazole EE <sup>2</sup> (%)
1	4.5	2	8	65.9	0.506	93.91
2	3	2.5	8	79.4	0.248	98.71
3	6	2.5	7	26.6	0.432	96.05
4	4.5	3	8	68.0	0.413	97.16
5	4.5	3	7	45.1	0.492	99.66
6	4.5	2.5	7.5	45.4	0.36	97.75
7	3	2	7.5	82.1	0.291	94.67
8	3	3	7.5	90.1	0.311	97.26
9	4.5	2.5	7.5	47.6	0.399	97.67
10	6	2.5	8	39.8	0.533	97.99
11	6	2	7.5	66.8	0.558	96.38
12	4.5	2.5	7.5	43.3	0.449	98.56
13	4.5	2	7	54.0	0.481	97.95
14	4.5	2.5	7.5	48.1	0.329	99.50
15	3	2.5	7	69.8	0.263	96.55
16	6	3	7.5	23.9	0.288	98.29
17	4.5	2.5	7.5	63.0	0.461	96.57

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<sup>1</sup> Polydispersity Index

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<sup>2</sup> Encapsulation Efficiency

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256 It was observed that the best fitted model was quadratic for all considered responses. The results of  
 257 the analysis of variance (ANOVA) performed to investigate the models for the three responses  
 258 (particle size,  $Y_1$ ; PDI,  $Y_2$ ; encapsulation efficiency,  $Y_3$ ) are shown in **Table 3**. The significance of the  
 259 regression coefficients for the input factors was evaluated by the correspondent p-value (<0.05)  
 260 calculated by ANOVA (see Table 3). The p-values were used as tool to check the significance of each  
 261 coefficient for understand the pattern of the mutual interactions between the selected process  
 262 parameters. The smaller the magnitude of the p-value, the more significant is the corresponding  
 263 coefficient. A p-value below 0.05 was considered an indication of a significant contribution of the  
 264 factor (see **Table 3**). The fit of the model was evaluated by coefficient of determination ( $R^2$ ).

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**Table 3.** ANOVA results obtained for the regression model obtained from the experimental design for the prediction particle size, PDI and EE.

Factors	Particle Size (Y <sub>1</sub> )			PDI (Y <sub>2</sub> )			EE (Y <sub>3</sub> )		
	Coefficient Estimate	F-value	p-value	Coefficient Estimate	F-value	p-value	Coefficient Estimate	F-value	p-value
Intercept	49.49	11.56	0.0020*	0.3996	4.26	0.0346*	98.1	0.64	0.7392
X <sub>1</sub>	-20.54	66.07	0.0001*	0.0873	17.58	0.0041*	0.1900	0.10	0.7649
X <sub>2</sub>	-5.21	4.24	0.0784	-0.0415	3.98	0.0864	1.18	3.74	0.0942
X <sub>3</sub>	7.19	8.10	0.02*	0.0040	0.04	0.8530	-0.3050	0.25	0.6330
X <sub>1</sub> X <sub>2</sub>	-12.73	12.69	0.0092*	-0.0725	6.07	0.0433*	-0.1700	0.04	0.8497
X <sub>1</sub> X <sub>3</sub>	0.8825	0.06	0.8120	0.0290	0.97	0.3573	-0.0550	0.00	0.9510
X <sub>2</sub> X <sub>3</sub>	2.76	0.59	0.4660	-0.0260	0.78	0.4063	0.3850	0.20	0.6695
X <sub>1</sub> <sup>2</sup>	5.96	2.93	0.1307	-0.0708	6.09	0.0430*	-0.6025	0.51	0.4976
X <sub>2</sub> <sup>2</sup>	10.29	8.73	0.0212*	0.0332	1.34	0.2851	-0.7575	0.81	0.3984
X <sub>3</sub> <sup>2</sup>	-1.56	0.20	0.6682	0.0332	1.96	0.2038	-0.0825	0.01	0.9247
<b>Model</b>									
<b>R<sup>2</sup></b>		0.9369			0.8456			0.4513	
<b>F-value</b>		11.48			4.33			0.6347	

267 \* Significant effect of the factor on individual responses.

268 Abbreviations: ANOVA, analysis of variance; PDI, polydispersity index; EE, encapsulations efficiency; X<sub>1</sub> surfactant concentration; X<sub>2</sub> liquid lipid concentration; X<sub>3</sub>

269 solid lipid concentration; F-value, ratio of the mean regression sum of squares divided by the mean error sum of squares; R<sup>2</sup>, coefficient of determination.

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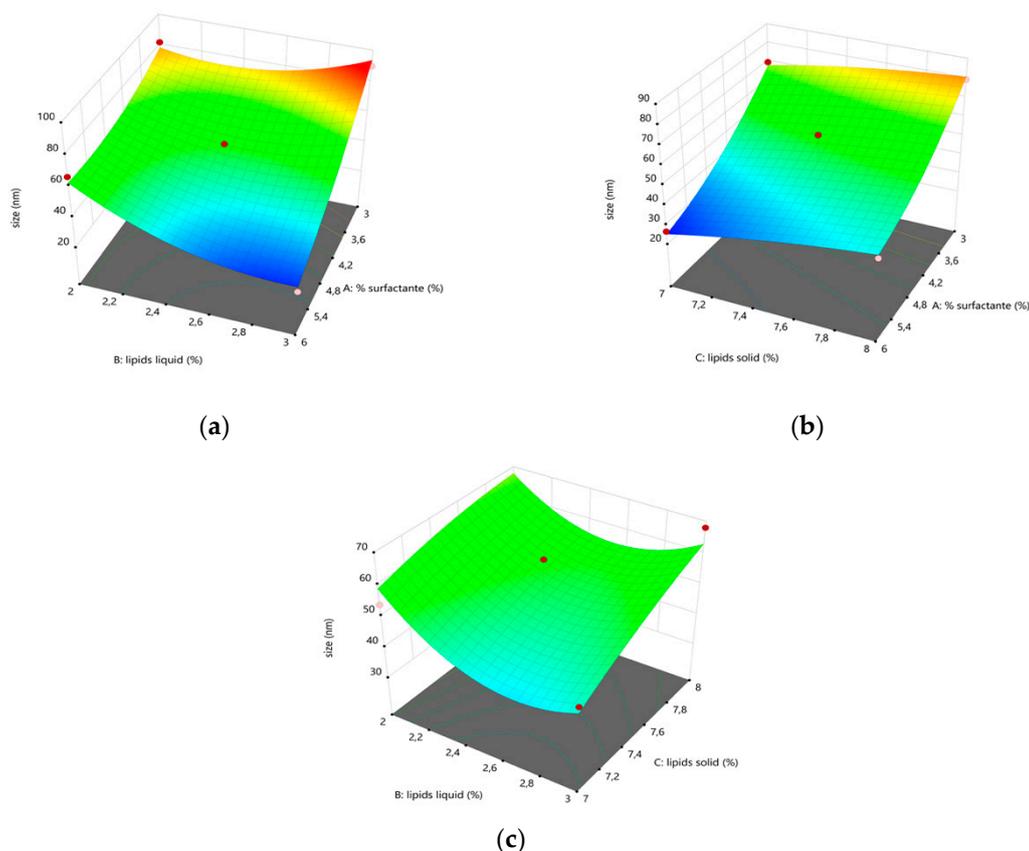
### 3.1.1 Particle size

In the case of particle size, the value of coefficient of determination ( $R^2 = 0.9369$ ) of the quadratic model indicated that only 6.31% of the total variations was not explained by the model (see **Table 3**). The quadratic model correlating input factors to particle size resulted in the polynomial Equation (3) and allowed to design three response surface graphs (Figure 1) for the optimization lipid nanoparticle size.

$$Y_1 = 49.49 - 20.54X_1 - 5.21X_2 + 7.19X_3 - 12.73X_1X_2 + 0.88X_1X_3 + 2.76X_2X_3 + 5.96X_1^2 + 10.29X_2^2 - 1.56X_3^2 \quad (3)$$

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The concentration of surfactant ( $X_1$ ), liquid lipid ( $X_2$ ), solid lipid ( $X_3$ ) and interaction between surfactant and liquid lipid ( $X_1X_2$ ) and were the factors significantly affecting ( $p < 0.05$ ) the particle size ( $Y_1$ ) according to ANOVA (see **Table 3**). A positive coefficient value before a factor in the regression equation indicates that the response increases along with the factor and *vice versa*. The coefficient values show that particle size is strongly dependent from concentrations of TPGS. In fact, as shown in Figure 2a, when maintaining the Ucuùba fat (solid lipid) concentration constant, lipid nanoparticles particle size decreased when the concentrations of Capryol 90 (liquid lipid) and TPGS (surfactant) increased (Figure 1a).



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**Figure 1.** Three-dimensional response surface plots showing the effect of (a) the concentration of liquid lipid and surfactant, (b) the concentration of surfactant and solid lipid and (c) the concentration of liquid lipid and solid lipid on NLC particle size, respectively.

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Similarly, blocking the concentration of Capryol 90 and varying the concentrations of TPGS and Ucuùba fat, it was observed a decrease of lipid nanoparticles particle size with the increase of the

294 surfactant concentration and the decrease of the Amazonian fat concentration (Figure 1b). On the  
 295 other hand, when the concentration of the surfactant was maintained constant and the concentration  
 296 of lipids was varied (Figure 1c), no significant variations in terms of particle size were observed.  
 297

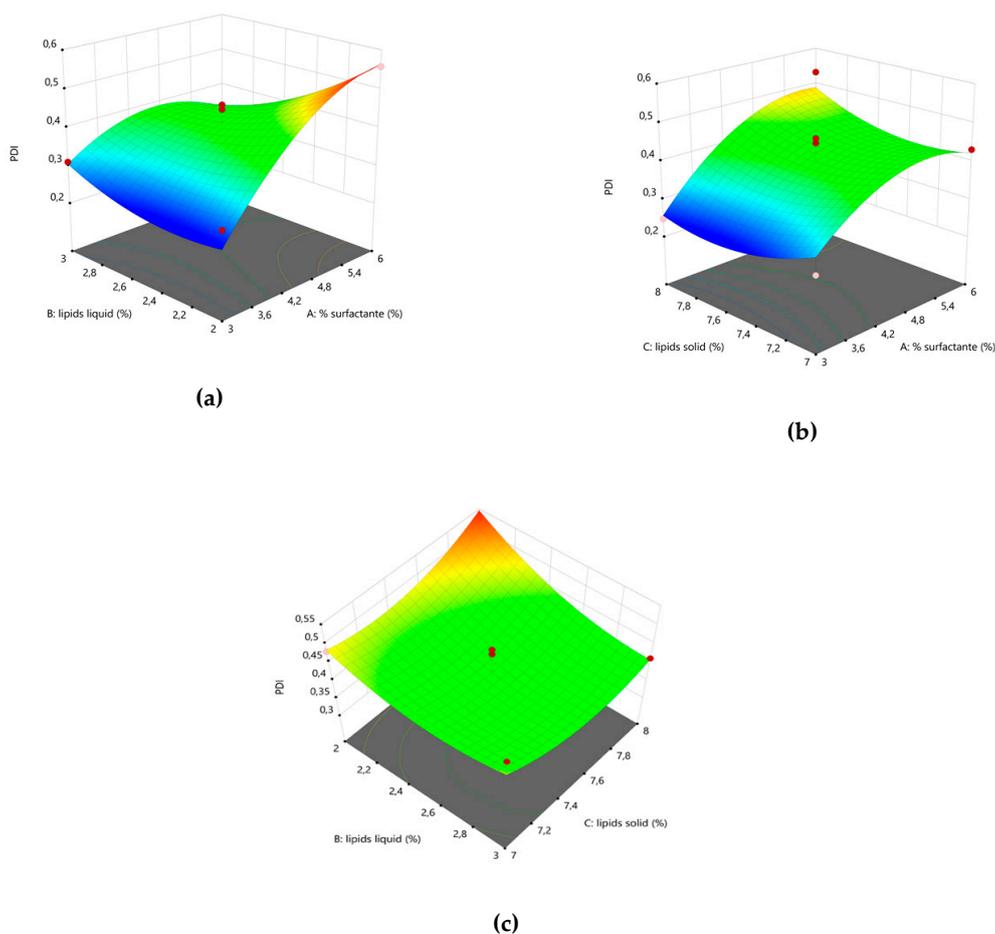
### 298 3.1.2. Polydispersity Index (PDI)

299 For PDI, the value of coefficient of determination of the quadratic was  $R^2=0.8477$  indicating that  
 300 15.23% of the total variations was not explained by the model, an higher when compared with the  
 301 6.31% the particle size related model (see all the values in Table 3).  
 302

303 The effect of the selected factors on PDI was described by the quadratic Equation (4).

$$Y_2 = 0.400 + 0.087X_1 - 0.041X_2 + 0.004X_3 - 0.073X_1X_2 + 0.029X_1X_3 - 0.026X_2X_3 - 0.071X_1^2 + 0.033X_2^2 - 0.040X_3^2 \quad (4)$$

304 According to ANOVA the significant factor was the concentration of the surfactant and the  
 305 combination of surfactant and liquid lipid concentrations. Figure 2 shows the response surface plots  
 306 correlating the PDI to the factors investigated.



307

308

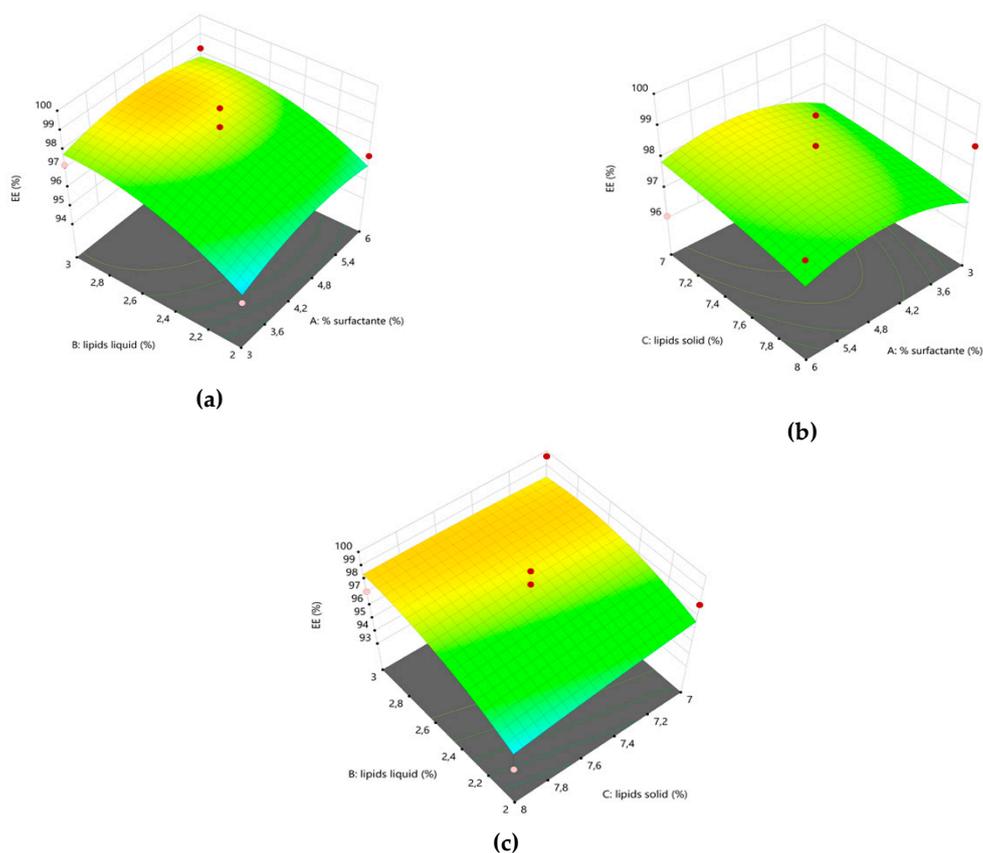
309 **Figure 2.** Three-dimensional response surface plots showing the effect of (a) the concentration of  
 310 liquid lipid and surfactant, (b) the concentration of surfactant and solid lipid and (c) the concentration  
 311 of liquid lipid and solid lipid on NLC polydispersity index (PDI), respectively.

312 Analysing the response surfaces for PDI, again it appears that this nanoparticle attribute was  
 313 mainly influenced by surfactant concentration (Figure 2a and 2b). However, differently from particle  
 314 size, PDI steeply increased when the TPGS concentration was increased. The Ucuùba fat and Capryol  
 315 90 concentrations (Figure 2c) appeared to be less relevant for PDI values compared to their effect  
 316 combined with surfactant variations, even if a tendency to an increase of the PDI when both the solid  
 317 lipid to liquid lipid concentrations increase is evident. In any case, the response surface plot  
 318 representing Ucuùba fat and Capryol 90 interaction, keeping constant TPGS concentration, resulted  
 319 overall in a not significant PDI variation.

320

### 321 3.1.3. Ketoconazole Encapsulation Efficiency

322 The model correlating ketoconazole encapsulation efficiency to the selected factors was found  
 323 to be not significant ( $R^2=0.4513$ , p-value 0.7392). Then, within the present design space it was not  
 324 possible to achieve a significant different in terms of entrapment efficiency. It was possible to  
 325 evidence that ketoconazole at the selected concentration of 0.5% w/v could be efficiently loaded into  
 326 nanostructured lipid carriers obtained using TPGS surfactant, Capryol 90 as liquid lipid, and Ucuùba  
 327 fat as solid lipid within the range of concentration tested (Figure 3).

328  
329

330 **Figure 3.** Three-dimensional response surface plots showing the effect of the concentration of liquid lipid  
 331 and surfactant (a), the concentration of surfactant and solid lipid (b) and the concentration of liquid lipid  
 332 and solid lipid (c) on NLC ketoconazole encapsulation efficiency (EE), respectively.

### 333 3.1.4. Formulation optimization and model validation

334 A desirability function was applied to the models obtained to identify the process parameters  
 335 required to prepare optimized formulations. The optimal formulation was based on the criteria of  
 336 maximum entrapment efficiency, minimum PDI and specific particle size. The actual values of the  
 337 dependent variables at the optimal combination of factors suggested by the software were assessed

338 to confirm the validity of the calculated optimal factors by expected responses. In particular two  
 339 optimized formulations were prepared using the model, one with smaller (30 nm, F 30) and a second  
 340 with larger (85 nm, F 85) required particle size, respectively. Other constraints were minimum PDI  
 341 and maximum entrapment efficiency. The composition of the optimized formulations is presented in  
 342 Table 4. Formulation F 85 showed the highest residual value in terms of size (10.4 nm) and PDI (0.114),  
 343 this value indicates how much the experimental value is far from the predicted value. The overall  
 344 desirability value for these formulations (obtained using a desirability function that assigns numbers  
 345 between 0 and 1 to the possible values each response with 0 representing a completely undesirable  
 346 value and 1 representing the ideal response value) was 0.86 for F85 and was 0.78 for F 30. The good  
 347 correlation between the actual and predicted results indicates that Box–Behnken design along with  
 348 desirability function could be successfully used to optimize the nanoparticles manufacturing process.  
 349

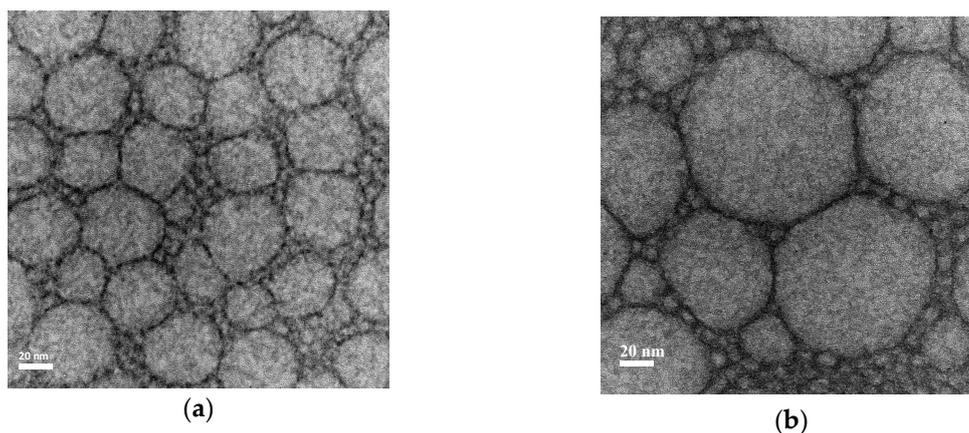
350 **Table 4.** Composition of the optimized Ucuùba fat-based NLC formulations F 30 and F 85 along with  
 351 the predicted, observed and residual values of the responses

Input Factors	F 30			F 85		
Surfactant (% w/v)	6			3		
Liquid Lipid (% w/v)	3			2.75		
Solid Lipid (% w/v)	7.63			7.68		
Responses	Predicted	Observed	Residual*	Predicted	Observed	Residual*
Size (nm)	30.0	33.6	-3.59	85.0	74.6	10.4
PDI	0.339	0.255	0.084	0.257	0.143	0.114
EE (%)	97.85	98.20	-0.35	97.67	98.70	-1.03

352 \*Residual = Predicted – Observed

353

354 Figure 4 shows the TEM images of the NLC optimized formulations F30 and F85.



355 **Figure 4.** Representative TEM images of the optimized Ucuùba fat-based NLC formulations (a) F 30 and  
 356 (b) F 85.

357 The lipid nanoparticles displayed a globular shape, characterized by a spherical homogeneous  
 358 lipid core surround by a surfactant corona. It can be appreciated from the images that particle size  
 359 distribution appears quite uniform, as indicated by the low polydispersity index obtained with DLS  
 360 measurements.

361

## 362 3.2. Stability study of Ucuùba fat-based NLC prepared according to the optimized conditions

363 Ucuùba fat-based NLC prepared according to the optimized conditions were characterized  
 364 immediately after preparation and up to 30 days of storage at room temperature. Results are shown  
 365 in **Table 5**.

366

367 **Table 5.** Physico- chemical characterization of optimized Ucuùba fat-based NLC during 30 days  
 368 storage at room temperature

Time (days)	F30			F85		
	Z- average (nm)	PDI	$\zeta$ potential (mV)	Z- average (nm)	PDI	$\zeta$ potential (mV)
0	33.6 ± 0.2	0.26 ± 0.01	-15.2 ± 0.8	74.6 ± 0.3	0.14 ± 0.00	-24.5 ± 0.6
1	33.7 ± 0.1	0.20 ± 0.01	-16.4 ± 1.3	74.1 ± 0.2	0.13 ± 0.02	-21.3 ± 1.1
7	34.4 ± 0.3	0.19 ± 0.00	-13.3 ± 1.1	72.9 ± 0.3	0.16 ± 0.04	-20.2 ± 6.5
15	33.3 ± 0.2	0.22 ± 0.01	-12.1 ± 1.3	71.1 ± 0.1	0.15 ± 0.01	-17.4 ± 0.6
30	33.8 ± 0.2	0.20 ± 0.01	-15.2 ± 0.8	73 ± 0.2	0.15 ± 0.01	-20.2 ± 1.5

369

370 The average particle size did not change significantly during the 30 days of observation.  
 371 Furthermore, the PDI of the optimized Ucuùba fat-based NLC were found to be below the values  
 372 0.16 and 0.26, respectively for F85 and F30, indicating a narrow particle size distribution and absence  
 373 of agglomerates for up to 30 days storage. All formulations exhibited negative zeta potential which  
 374 were around -15 mV and -25 mV, respectively for F30 and F85, and did not show significant changes  
 375 during the 30 days storage ( $p > 0.05$ ).

376

## 377 3.2.2 DSC

378 Differential Scanning Calorimetry is widely used for investigation the solid state of drugs and  
 379 excipients, such as lipids. In particular, thanks to the melting point identification and melting  
 380 enthalpies determination, this technique is interesting to highlight modifications of crystalline  
 381 components of the formulations, complementing the XRD data. The Ucuùba fat evidenced a melting  
 382 point of 43 °C (see Figure 5 and **Table 6**) while the blends of Ucuùba fat with Capryol 90 showed  
 383 lower melting points (see Figure 5a and **Table 6**). At the same time a reduction of the melting enthalpy  
 384 was evident.

385

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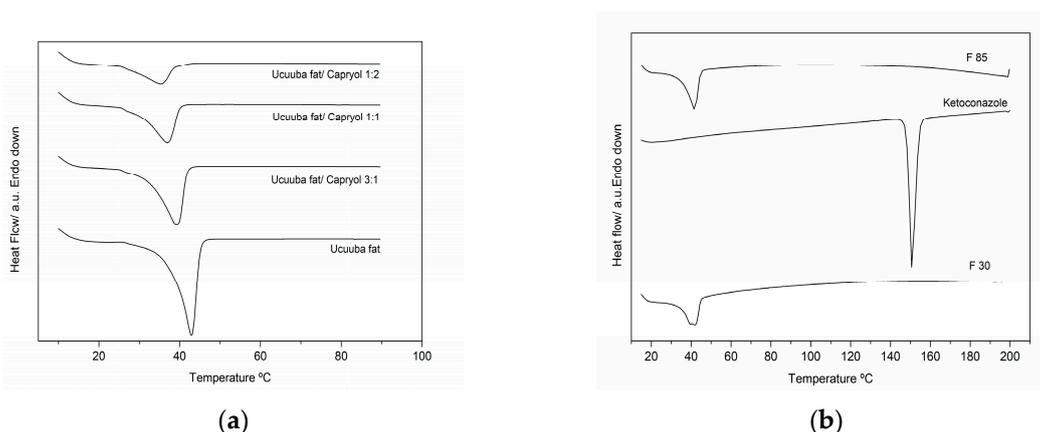
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391



392 **Figure 5.** DSC traces of Ucuuba fat/Capryol 90 mixtures (a) and of the optimized Ucuuba fat-based NLC  
 393 formulations F 30 and F 85 along with ketoconazole raw material (b).

394 Table 6 shows the value CI which decreases with the increases of the amount Capryol 90. The  
 395 data evidence that mixture with Capryol 90 influenced the order of Ucuuba fat crystalline lattice  
 396 contributing to increase the lattice disorder causing the decrease crystalline index and a shift in the  
 397 melting point to lower value. This last characteristic is desirable for the production of NCL, since a  
 398 solid fat with less ordered lattice, reduces the probability of stability issues related to the expulsion  
 399 of the entrapped active compound during storage [28].

400 **Table 6.** Melting temperatures, enthalpy and crystallinity index obtained by DSC analysis of the  
 401 Ucuuba fat alone and in mixture with Capryol 90  
 402

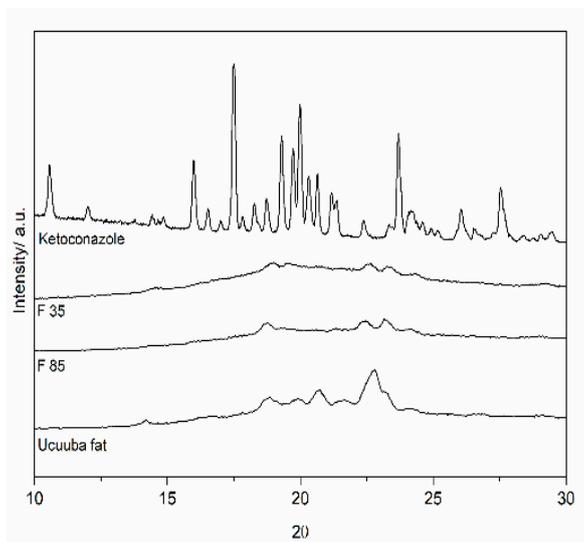
Ucuuba fat / Capryol 90 weight ratio	T <sub>onset</sub> (°C)	T <sub>endset</sub> (°C)	Peak (°C)	Enthalpy (mJ)	CI (%)
1:0	37.76	45.11	42.99	562.72	100
3:1	31.96	41.77	39.79	493.38	65.75
1:1	28.90	40.00	36.98	452.26	40.18
1:2	26.61	38.91	35.24	260.32	15.16

403 The DSC results presented in Figure 5b are related to ketoconazole and the two optimized  
 404 formulations F30 and F85 after drying. It can be observed that only one endothermic peak appeared  
 405 in the DSC traces obtained for both NLC formulations F30 and F85. For formulation F30 a single  
 406 melting peak at 41.86°C and melting enthalpy of -89.13 J/g were observed during the heating process,  
 407 attributed to lipid matrix solid-to-liquid transition. Similarly, also for F85 a single melting peak was  
 408 evident at 41.59°C, corresponding to an enthalpy of -54.59 J/g were also related to the lipid matrix  
 409 melting. Considering that ketoconazole showed an endothermic peak of fusion at 150.76°C, it can be  
 410 inferred that ketoconazole is molecularly dispersed in the lipid matrix and no detectable drug crystals  
 411 are formed during the production of the nanostructured lipid carriers.  
 412

### 413 3.2.3 XRD 414

415 Ucuuba fat is a mixture of mono-, di- and tri- triglycerides (trimyrustin being the main component)  
 416 characterized by a complex polymorphism that can be evidenced by X-ray diffraction. The  
 417 diffractograms of the Ucuuba fat and the dried NLC formulations F 30 an F 85 are shown in Figure  
 418 6. The diffraction pattern Ucuuba fat appears similar a diffraction pattern F 30 and F 85 (presence of

419 peaks at 2 theta 18.8°, 22.6° and 23.5°), but the diffractograms of the NLC formulations show more  
 420 broad and less defined peak indicating a solid state more amorphous than the virgin Ucuùba fat. This  
 421 was expected also considering the presence in the NLC formulations of the liquid lipid Capryol 90,  
 422 that as shown by DSC data, reduced the Amazonian fat organization.  
 423

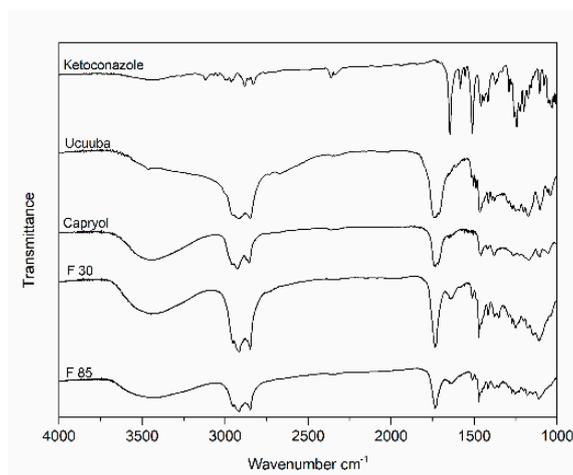


424  
 425 **Figure 6.** The X-ray diffraction spectra of Ucuùba fat and optimized Ucuùba fat-based NLC formulations F 30  
 426 and F 85.  
 427

428 At the same time it can be appreciated that of the numerous peaks characterizing the diffraction  
 429 pattern of ketoconazole crystalline powder (see Figure 6) none appears in the diffractogram obtained  
 430 for ketoconazole-loaded NLC formulations F30 and F85, possibly indicating the ketoconazole is  
 431 completely dispersed in the lipid matrix, even if due to the relatively low drug loading the presence  
 432 of crystalline drug would be difficult to detect.  
 433

#### 434 3.2.4 FTIR

435 Infrared spectroscopy (FTIR) is a valuable qualitative tool for characterization of the interaction of  
 436 the components of pharmaceutical systems. FTIR analysis were performed for the main components  
 437 of the nanostructured lipid carriers, *i.e.* Capryol 90, Ucuùba fat and ketoconazole, and the two  
 438 Ucuùba fat-based NLC formulations (F 30 and F 85). The spectra are presented in Figure 7 and it  
 439 appears clear that the FT-IR spectra of Ucuùba fat, Capryol 90 and NLC samples presented striking  
 440 similarities and were almost superimposable. In these spectra, the high intensity peak at 1745 cm<sup>-1</sup>  
 441 corresponds to the vibration carbonyl groups [30]. The vibration of the C-O bond of ester bonds  
 442 present in triacylglycerols is related to the sequence of peaks identified at 1232, 1165, 1118 and 1097  
 443 cm<sup>-1</sup> [31]. The high intensity bands observed at 2922 and 2850 cm<sup>-1</sup> are characteristic of the C-H  
 444 stretching vibrations of lipids and correspond to olefinic (CH=CH) double bonds and methyl and  
 445 methylene groups of fatty acid chains, respectively (Figure 7) [32]. Overall these bands, that are  
 446 characteristic of the lipid nature of Ucuùba fat and Capryol 90, are also strongly represented in the  
 447 FTIR spectra of the NLC formulations in which the lipid matrix is the defining component.



448  
449 **Figure 7.** FT-IR spectra of ketoconazole, Ucuùba fat, Capryol 90 and of optimized Ucuùba fat-based NLC  
450 formulations F 30 and F 85.  
451

452 The FT-IR spectrum of pure ketoconazole shows at least three characteristic peaks at 1647, 1510 and  
453 1242  $\text{cm}^{-1}$ , corresponding to the carbonyl (C=O), aromatic ring and C-N piperazine bonds stretching,  
454 respectively [33]. The spectrum ketoconazole in the spectra of formulations is almost completely  
455 disappeared, suggesting an efficient drug entrapment into the solid lipid nanoparticle matrix.  
456 However, the drug loading is demonstrated by the presence of weak bands between 1600 and 1650  
457 and around 1250  $\text{cm}^{-1}$ , not so evident in the FTIR spectra of lipid excipients.  
458

459

#### 460 4. Discussion

461 The main purpose of this study was the development and optimization Ucuùba fat-based NLC, using  
462 drug model the ketoconazole, for treatment onychomycosis. The pharmaceutical sector, since the  
463 introduction at starting of the new millennium of the Quality-by-Design approach, has started using  
464 systematically design of experiment as a statistical tool to maximize process knowledge and  
465 efficiently optimize product quality characteristics [34,35]. This approach has been applied to the  
466 manufacturing process of several dosage forms, including as tablets [36], pellets [37], semisolid  
467 formulations [38] and spray-dried powders to be used in dry powder inhalers [39].

468 However, DoE appears especially suitable for the development of innovative and complex dosage  
469 forms such as nanoparticles. In fact, scale-up manufacturing issues, lack of reproducibility with tight  
470 analytic standards and poor process control have been indicated as some of the critical underlying  
471 problems that hampered in the past the transfer of nanomedicines to the clinic [40]. At the same time,  
472 the potential of nanoformulations and in specific of lipid nanovectors for their inclusion for oral  
473 [41,42] or transmucosal drug products has not yet been fully tapped in [43–45].

474 For these reasons, in this study high-pressure homogenization, a robust and easy to scale up method,  
475 was selected for the manufacturing of nanostructured lipid carriers [46] and a Box-Behnken design  
476 (BBD) was applied to understand and optimize of Ucuùba fat-based NLC manufacturing process.  
477 The BBD allows to obtain a mathematical model describing as closely as possible effects of the factors  
478 and their levels on the critical quality attributes of the product, over the whole of the experimental  
479 domain, and also to predict the response inside domain [47].

480 The high-pressure homogenization manufacturing process led to a homogeneous, clear and  
481 transparent yellow-brownish dispersions, characteristic color of the Ucuùba fat. Particle size ( $Y_1$ )  
482 analysis confirmed that it was possible to obtain dispersions below 100 nm over all the experimental  
483 domain. Particle size was strongly influenced by TPGS and Ucuùba fat concentrations. The liquid  
484 lipid Capryol 90 influenced the particle size only in combination with TPGS ( $X_1X_2$ ,  $p=0.0092$ ) probably  
485 because it acts as co-surfactant in this formulation.

486 In particular, TPGS is a vitamin E derivative and a nonionic surfactant able to stabilize nanoemulsions  
487 by lowering interfacial tension with steric repulsion between nanoparticles. As a consequence, bigger  
488 amounts of surfactant result in smaller particle size [48]. Interestingly, TPGS has recently been shown  
489 to release by enzymatic cleavage of ester bonds tocopherol and tocopherol succinate [49]. Alpha-  
490 tocopherol has been demonstrated to be able to increase amphotericin B activity against *Candida*  
491 *albicans* [50]. Therefore, the surfactant increase could not only contribute to particle size decrease, but  
492 high TPGS concentrations could result in an enhanced antifungal activity of ketoconazole.

493 PDI ( $Y_2$ ) values ranged from 0.248 to 0.558. This parameter reflects homogeneity of nanoparticle  
494 populations, values lower than 0.3 indicating relatively narrow particle size distribution and values  
495 even lower associated with monodisperse particles are generally desirable. Higher values are on the  
496 contrary associated to polydisperse particle size distribution [51]. In our DoE, formulations with high  
497 surfactant concentration showed PDI values above 0.4 and multimodal size distribution. The  
498 correlation among surfactant concentration and increasing PDI value could be explained considering  
499 that an excess of TPGS tends to self-assemble into micelles. Furthermore, it is possible that during the  
500 Ucuùba fat-based NLC preparation with high pressure homogenizer the temperature increase could  
501 have decreased the TPGS critical micelle concentration (CMC) and favoured the production of  
502 different micelle populations leading to PDI increase [29].

503 Concerning ketoconazole encapsulation efficiency was found to range from 93.91 to 99.66 %. The  
504 encapsulation efficiency in this work was found not to depend in a statistically significant manner  
505 from the process parameters explored. Probably this result was obtained because the percentage  
506 ketoconazole was maintained constant in all formulations (0.5% w/v ketoconazole) and Ucuùba fat-  
507 based NLC offered sufficient lipid matrix to fully solubilize and encapsulate ketoconazole inside  
508 nanoparticles. The lipid phase NLC is the mixture of the solid and liquid lipids, in this work Ucuùba  
509 fat was the choice for the solid lipid and Capryol is the liquid lipid. This peculiarity of NLC is the  
510 reason for its higher encapsulation efficiency and better storage stability when compared to solid  
511 lipid nanoparticles (SLN) [52]. In particular, the DSC thermograms of Ucuùba fat was found to be  
512 affected by the mixing with Capryol and it was observed that increasing the ratio of Capryol a  
513 broadening of the melting peak of the solid fat was observed, possibly indicating the formation of  
514 several polymorphic forms. Similarly, [53] studied the effect of the addition of  $\alpha$ -tocopherol on the  
515 crystalline organization of Compritol® ATO 888, for NLC production, and observed that the addition  
516 of 30%  $\alpha$ -tocopherol decreases the melting point of Compritol from 71° C to 50.3° C. The authors  
517 explained this melting point decrease to the appearance of less organized polymorphic arrangements  
518 in the solid lipid due to the addition of  $\alpha$ -tocopherol.

519 In order to study lipid matrix crystalline organization, X-ray diffraction is a widely used technique  
520 also for the investigation of SLN and NLC [54]. The lipid polymorphism is a property that allows the  
521 triglycerides to exist in the form of several crystalline structures in different molecular packages [55].  
522 The three most common types of packaging of triglycerides are hexagonal, orthorhombic and  
523 triclinic, which are usually referred to as polymorphic forms  $\alpha$ ,  $\beta'$  and  $\beta$ , respectively. The type of  
524 crystalline form depends on the molecular structure and composition of the lipids, as well as the  
525 environmental conditions during storage (cooling rate, for example). For the Ucuùba fat based- NLC,  
526 the sharp peaks at were at 18.9°, 22.6° and 23.2°, which correspond to short spacings at 0.46 nm, 0.393  
527 nm and 0.382 nm indicative of  $\beta$  e  $\beta'$ , respectively [55,56]. The measurements of crystallinity and lipid  
528 modification are strongly connected with drug release and drug incorporation [55].

529 Zeta potential values are often used to predict the stability of nanoparticles. In fact, this parameter  
530 reflects the charge of the surface of nanoparticles in suspensions and high zeta potential values  
531 indicate high repulsive forces between particles, preventing them from forming large particle size  
532 agglomerates [10,51,57,58]. In general, nanomaterials with zeta potential above 30 mV are considered  
533 highly stable [59,60]. However, in our DoE the zeta potential absolute values were found to be smaller  
534 than 30 mV. However, these systems which contain a steric stabilizer, such as the non-ionic and  
535 PEGylated TPGS and as a consequence, the absorption of the steric stabilizer on nanoparticle surface  
536 decreases the zeta potential due the shift of the shear plane of the electrical double layer existing

537 around each particles [10] but it provides a steric contribution to particle stability due the presence  
538 of the hydrophilic polymer chains [43].  
539

## 540 5. Conclusions

541 In this study, nanostructured lipid carriers formulated using an Amazonian fat were manufactured  
542 using a high-pressure homogenizer and the process was studied and optimized using a design of  
543 experiment approach. Results obtained allowed to build a predictive mathematical model showing  
544 that among the critical process parameters studied surfactant and Ucuùba fat concentrations were  
545 the most significant factors for the determination of the nanoparticles average size and particle size  
546 distribution in terms of polydispersity index. The encapsulation efficiency was found to be above  
547 90% for all the experimental design space and factors selected did not significantly affect its value in  
548 this model.

549 Using the model obtained with the design of experiment, two optimized NLC formulations, one with  
550 low (30 nm) and high particle size (85 nm) and both characterized by low PDI and high encapsulation  
551 efficiency. The optimized nanoparticles were successfully obtained and were stable during the 30  
552 days, in which there was no significant change in the values of PDI, zeta potential and particle size.  
553 The analysis of the XDR and DSC analysis were excellent tools for the characterization of the crystal  
554 order the formulation, showed that formulations the decrease the crystal order formulations when  
555 compared Ucuùba fat. The total absence of the ketoconazole signals in the DSC thermograms and  
556 XDR spectra, suggested an optimal drug encapsulation and no sign of drug expulsion.

557 The optimized formulations obtained will be further investigated for their topical application against  
558 onychomycosis.

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560 F.R.; investigation, R.R.P.; M.T.; writing—original draft preparation, R.R.P.; writing—review and  
561 editing, R.R.P., F.S. and R.B.; supervision, C.P., F.S.; funding acquisition, R.B., P.S. and J.O.C.S.J.

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564 **Conflicts of Interest:** The authors declare no conflict of interest.

565

566

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