

1 *Article*

2 Optimization of the CO₂ Liquefaction Process - 3 Performance Study with Varying Ambient 4 Temperature

5 Steven Jackson* and Eivind Brodal

6 UiT- Norges Arktiske Universitetet

7 * Correspondence: steve.jackson@uit.no

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9 In CCS projects, the transportation of CO₂ by ship can be an attractive alternative to transportation
10 using a pipeline, particularly when the distance between source and disposal location is large.
11 However, the energy consumption of the liquefaction process can be significant, making the
12 selection of an energy-efficient design an important factor in the minimization of operating costs.
13 Since the liquefaction process operates at low temperature, its energy consumption will vary with
14 ambient temperature, which could be a factor that influences the trade-off point between pipelines
15 and shipping in different geographic locations. A consistent set of data showing the relationship
16 between energy consumption and cooling temperature is therefore potentially useful to CCS system
17 modelling. This study compares the performance of a wide range of CO₂ liquefaction schemes. It
18 applies a methodical approach to the optimization of process operating parameters and studies
19 performance across a range of operating temperatures. A set of data for the minimum energy
20 consumption cases is presented. The main findings are that open-cycle CO₂ processes often offer
21 minimum energy consumption; NH₃ based schemes often offer better performance at higher
22 ambient temperatures; and that for the cooling temperature range 15 to 50 °C, the energy
23 consumption for the best performing liquefaction process rises by around 40%.

24 **Keywords:** CO₂; liquefaction; CCS; optimization; ambient temperature.

25

26 **1. Introduction**

27 The transportation of CO₂ by sea on a small scale has been a commercial practice in Europe for
28 several years, where ships are used to transport food-quality CO₂ from production plants to coastal
29 distribution terminals [1]. The current commercial vessel sizes vary between 1000 and 1500 m³ and
30 the transport pressure is in the range 14–20 bara [2]. Although the use of shipping in CCS projects
31 would require a considerable scale-up in transport capacity, there are no technical barriers and
32 shipping has long been identified as potential option for the long-distance transport of CO₂. The IPCC
33 special report on CCS [1], for example, identifies shipping as the lowest cost option for distances over
34 1700 km. Other studies such as Mallon et al. [3] and Jakobsen et al. [4] have found that the trade-off
35 distance for shipping could be much lower. As a result, shipping of CO₂ is a standard feature in the
36 modelling of CCS transportation networks and a number of studies have been made into the technical
37 and commercial aspects of the liquefaction processes required [2,5-12].

38 When transporting CO₂ in ships, the energy consumption of the liquefaction process is
39 significant, implying high operating costs. As a result, many studies have considered the design of
40 the liquefaction process, particularly with a focus on reducing energy consumption. Hegerland [2]
41 states that “In principle, there are two process alternatives” and goes on to suggest that when low
42 temperature cooling water is available, CO₂ should be used directly as a cooling medium, but above
43 some trade-off temperature, an in-direct ammonia (NH₃) refrigeration process becomes the best
44 option. Subsequent studies have considered the selection of the optimal process flow scheme for the

45 liquefaction of CO₂ in more detail. Although the majority of work is focused on either open cycle CO₂
46 processes or closed cycle NH₃ refrigeration processes [6,7,10,13,14], others have also studied more
47 novel approaches such as the use of absorption refrigeration [15], cascade refrigeration [16], and the
48 application of turbo expanders [17]. Some have also compared a broad range of schemes.
49 Alabdulkarem et al. [16], for example, compares simple refrigeration schemes, cascade refrigeration
50 schemes, and absorption refrigeration schemes using waste heat. Most studies, however, focus on the
51 performance of one or two schemes and one set of operating conditions.

52 In addition to the selection of the process flow scheme, the chosen CO₂ transport pressure
53 represents an important operating parameter for the liquefaction process. Hegerland [2] states that
54 "To reduce investment costs of storage and ship tanks, it is required to operate as close to the triple
55 point of 5.17 bara and -56.6 °C as practically feasible." Aspelund et al. [6] and Lee, et al. [14] looked
56 at 6.5 bara transportation pressure based on the design of current commercial CO₂ transportation by
57 ship and also follow the assumption that the larger vessels used for CCS would operate at lower
58 pressures. Decarre et al. [7] compared liquefaction at 7 bara and 15 bara, finding that transportation
59 at 15 bara offers both lowest cost and lowest energy consumption. Seo et al., over the course of two
60 papers [10] and [18], also studied the optimum liquefaction pressure conditions finding that the
61 overall cost was lowest for 15 bara cases. More generally, both Seo et al. [18], Alabdulkarem et al. [16]
62 and Jackson et al. [13] found that the optimum liquefaction pressure for the transportation of CO₂ by
63 pipeline to be around 50 bara, which is well above the practical limits for ship based transport.

64 No study has been found that considers, in detail, how the minimum energy consumption of the
65 CO₂ liquefaction process varies with cooling temperature, or identifies the trade-off point between
66 NH₃ and CO₂ based systems. The aim of this study is, therefore, to make a review of the energy
67 consumption associated with the different process flow scheme examples available in the literature,
68 compare process performance at transport pressures up to 15 bara, and present a consistent set of
69 data for the variation in energy consumption of CO₂ liquefaction across a range of ambient
70 temperatures. The broader aim is to generate data that can be used in the modelling a CCS
71 transportation systems.

72 2. Materials and Methods

73 The study method is set out below in two parts: first, a general description of the study approach
74 is made, which breaks the methods employed into five steps; secondly, a more detailed description is
75 presented for each of the five steps.

76 *General description*

77 To begin with, a survey of previous studies was made to help identify the full range of possible
78 process flow schemes available for the liquefaction process. Based on a review of the studies
79 mentioned in the introduction [6,7,10,13-17], three were selected as the basis for further work that
80 covered all of the principle flow scheme alternatives: Alabdulkarem et al. [16], Seo, et al. [10], and Øi,
81 et al. [17]. From these three studies, four 'base' flow schemes were then selected.

82 In the first phase of the modelling work, each of the four 'base' schemes were modelled using
83 the parameters from the study to which they belong. This exercise provided both a verification of the
84 modelling approach and the correct interpretation of the design intent of each scheme. The
85 performance of each scheme was then compared using a new common set of 'base' parameters to
86 allow an unambiguous comparison of the performance of each process. Importantly, the operating
87 parameters for each of the process schemes were optimized with the 'base' parameters for minimum
88 energy consumption. After this, an additional set of process flow schemes was developed based on
89 the most promising features from the schemes already modelled and finally, the performance of all
90 schemes was compared to identify the best performing schemes to be use in the final phase of
91 modelling work.

92 In the final phase of work, the performance of the best performing schemes was investigated
93 over a range of ambient temperature conditions. At each temperature condition the operating

94 parameters of each scheme were optimized to provide an accurate reflection of how performance
 95 varies with ambient temperature.

96 The method described above is summarized below as a five-step process. More detail on each of
 97 the individual steps is then presented below under separate sub-headings.

- 98 1. Selection of 'base' flow schemes and parameters
- 99 2. Validation of the modelling basis
- 100 3. Optimization and comparison of the 'base' schemes
- 101 4. Development and selection of the study basis
- 102 5. Performance variation with cooling temperature

103 Selection of 'base' flow schemes and parameters

104 The 'base' schemes selected for this study were: a CO₂-NH₃ closed-loop refrigerant cascade
 105 scheme as Case 1 from Alabdulkarem et al. [16]; an NH₃ based refrigeration scheme as Case 2 from
 106 Seo et al. [10] and as Case 3 from Øi et al. [17]; and an open-cycle CO₂ scheme as Case 4, also from Øi
 107 et al. [17]. A process flow diagram (PFD) for each of these schemes is presented in Figure 1.

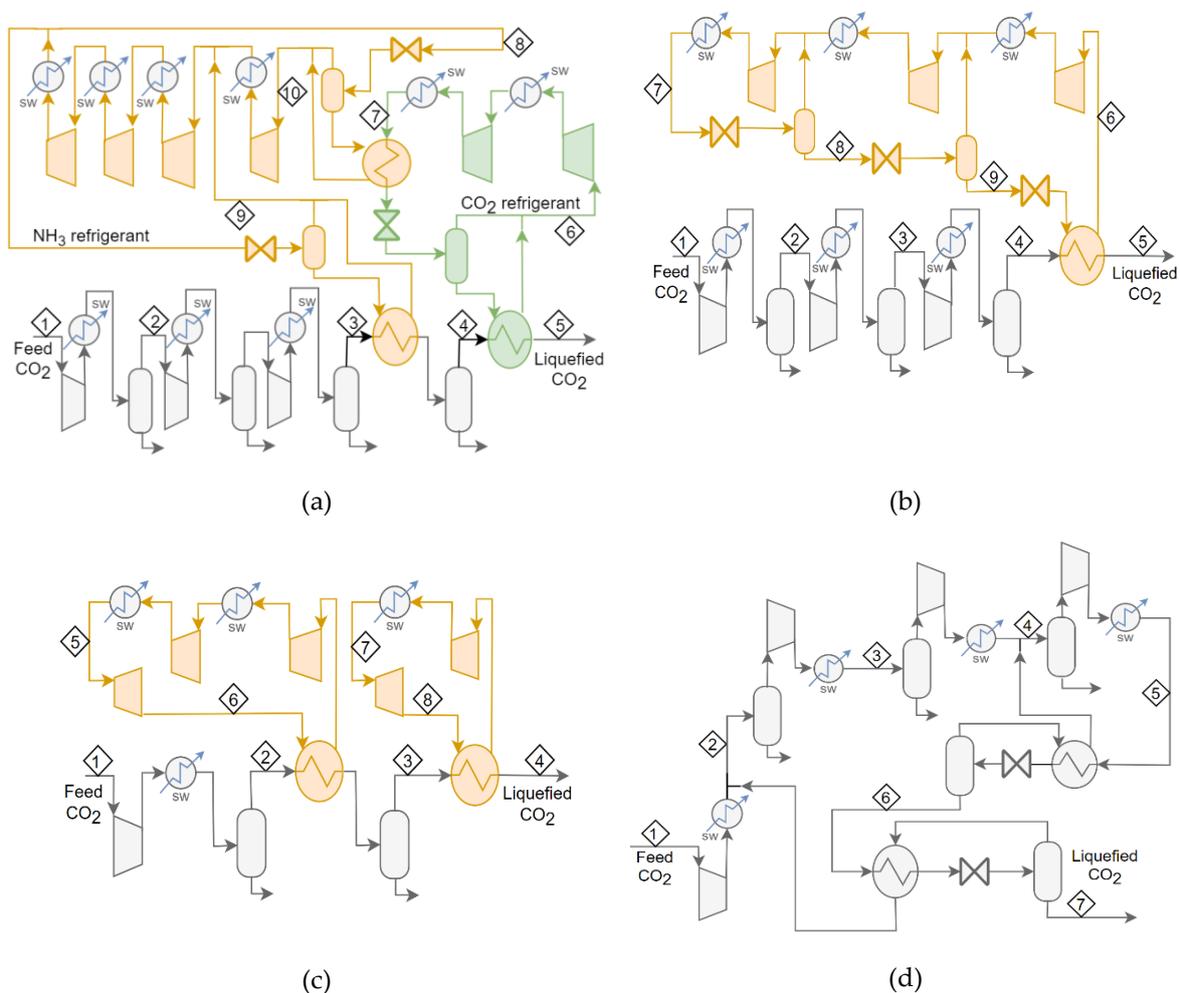


Figure 1. Process flow diagrams for the 'base' cases for the CO₂ liquefaction schemes (a) shows Case 1, (b) Case 2, (c) Case 3 and (d) Case 4.

108 Although each of these schemes was developed by the original study authors to achieve low
 109 energy consumption, each was developed based on a different set of modelling assumptions. A
 110 summary of the parameters used in each case is made below in Table 1.

111 To select a process scheme as the basis for this work, it was determined that the performance of
 112 each of the four 'base' case schemes (Cases 1 to 4) should be compared on a consistent basis. This
 113 basis was selected based on a review of the range of the original operating parameters used in Cases
 114 1 to 4. A review of the earlier work carried out by Jackson et al. [19] was also used to ensure
 115 compatibility with other related study work. The resulting 'base' parameters are also summarized in
 116 Table 1.

117 **Table 1.** Summary of modelling parameters.

	Case 1	Case 2	Cases 3 & 4	'base'
Exchanger pressure drop, ΔP , (bara)	0.10	0.20	0.50 ²	0.30 ¹
Min. exchanger approach, ΔT_m , (°C)	3	3	5	5
Compressor efficiency (%)	80	75	85	85
Compressor stage limit	90 °C	Pr < 3	150 °C ¹	150 °C
Expander efficiency (%)	-	-	90	90
Cooling utility temp, ΔT_c , (°C)	35	35	15	25
CO ₂ feed pressure (bara)	1.8	1.8	0.20	0.10
CO ₂ feed temp (°C)	38 ¹	40	15 ¹	30
CO ₂ product pressure (bara)	8.0	15	7.0	15
CO ₂ temperature (°C)	-45	-28	-50	-28

118 ¹assumed, ²pressure drop for the refrigerant side of evaporator is zero bara.

119 Omitted from Table 1 is a summary of the CO₂ compositions used as the basis in each of the
 120 cases. This is partly for the sake of brevity and partly because the composition of the CO₂ stream used
 121 is not a focus of this study. For validation purposes, the CO₂ feed stream compositions used in Cases
 122 1 to 4 corresponded to the original study basis; in all subsequent work a pure stream of CO₂ was used
 123 as the basis.

124 *Process modelling and validation*

125 A process model for each of the schemes was developed using Aspen HYSYS [20]. Operating
 126 conditions and energy flows were calculated using the Peng Robinson (PR) equation of state. Earlier
 127 studies have confirmed that PR generally provides reasonable accuracy in predicting the relevant
 128 properties for pure CO₂ apart from the region immediately around the critical point [21,22].

129 To validate the modelling approach the flow schemes shown in Figure 1 were recreated with the
 130 parameters used in their original development. Where modelling parameters could not be
 131 determined directly by reference to the published data, assumptions were made. In Case 1 the
 132 temperature out of the NH₃/ CO₂ exchanger was unknown and was, therefore, selected in this work
 133 to provide minimum overall power. In the study data published for Case 2, the compressor stage
 134 pressure ratios are not reported and, therefore, in this work a constant pressure ratio (equal to 2.1)
 135 was assumed. In Case 3, the number of compressor stages used in the LP ammonia refrigeration is
 136 unclear and in this work two compression stages are used to limit maximum temperature to 150 °C.

137 The detailed modelling results for each of the four validation cases are presented in Appendix A
 138 and a summary of the reported and modelled energy consumption is presented below in Table 2,
 139 which shows good agreement between the reported and modelled values.

140 **Table 2.** Summary of the reported and modelled power consumptions for the 'base' flow schemes.

	Case 1	Case 2	Case 3	Case 4
Modelled (kWh/ton)	106	101	79.9	86,6
Reported (kWh/ton)	104	102	80.3	87,0
Difference (%)	1.9	> 1	> 1	> 1

141 It is worth noting that in Case 1, the energy consumption associated with pumping the CO₂
 142 product up to 150 bara is included in the model validation work although this is not needed for
 143 liquefaction at low pressure.

144 *Optimization and comparison of the 'base' schemes*

145 To ensure a consistent basis for the comparison of the four 'base' flow schemes, the fixed set of
 146 'base' parameters shown in Table 1 were used in each of the cases. To ensure that the comparison
 147 was a fair one, the variable operating parameters for each case were optimized to achieve the
 148 minimum energy consumption for each case.

149 Implications of the 'base' parameters

150 Implementation of the 'base' parameters shown in Table 1 in the 'base' flow schemes for Cases
 151 1 to 4 (illustrated in Figure 1) has an impact on the flow scheme design in some cases. In Case 1, a
 152 cooling temperature of 25 °C and a maximum temperature of 150 °C requires one fewer stages in each
 153 of the three compressors compared to the 'base' flow scheme illustrated in Figure 1(a). In Case 2, the
 154 CO₂ feed compressor also required one fewer stages for the 8.9 bara case compared to the scheme
 155 shown in Figure 1, but the NH₃ compressor remained unchanged compared to the original design.
 156 In Case 3, due to a lower CO₂ feed pressure, an extra stage is needed in the CO₂ feed compressor. The
 157 low-pressure NH₃ refrigeration compressors associated with Case 3 also require an additional stage
 158 to meet the 150 °C limit, but the high-pressure compressor is unchanged. In Case 4, the CO₂ feed
 159 compressor also needs an additional stage, but the scheme is otherwise unchanged. Each of these
 160 modifications was implemented in the models developed for the 'base' case flow schemes when
 161 making the comparisons of process performance for each Case in this part of the study.

162 Optimisation

163 The approach to optimization used in this study was to conduct the optimization outside of
 164 HYSYS using a link to MATLAB [23]. This link allowed the two-way transfer of process parameters
 165 between MATLAB and HYSYS and the implementation of the optimization algorithms directly in
 166 MATLAB.

167 The optimization algorithms used in MATLAB were *fminsearch* and GA. *fminsearch* uses a
 168 simplex algorithm that is suitable for unconstrained, multi-variable, non-linear optimization
 169 problems. A benefit of this method is that it will usually quickly converge to a solution, a downside
 170 is that in some cases a local minimum may be obtained. A drawback specific to the application of this
 171 algorithm in this study is that an unconstrained search can also lead HYSYS to non-viable solutions
 172 (e.g., where temperature crossing is inevitable in heat exchangers), which pauses the HYSYS solver
 173 and requires a time-consuming re-set of the optimization process. The GA algorithm can solve
 174 smooth or non-smooth optimization problems with or without constraints. It is a stochastic,
 175 population-based, algorithm, which is generally slower than *fminsearch* to reach a solution, but more
 176 reliable in solving for the global minimum.

177 The objective of the optimization work was to minimize energy consumption, which can be
 178 expressed as the sum of all compressor and expander stage energy consumptions:

$$\min f(P_i) = \sum W_{comp} - \sum W_{exp} \text{ for } i = 1, \dots, n, \quad (1)$$

179 where P_i are the compressor discharge pressure levels that are optimized, W_{comp} are the
 180 compression stage energy flows (CO₂ and NH₃ compressor stages are not differentiated here), W_{exp}
 181 is the expander stage energy flows (only relevant where an expander forms part of the process
 182 scheme), and n is the number of variable pressure specifications, which varies from case-to-case, as
 183 described in more detail below.

184 In Case 1, the condensing pressure in the CO₂ refrigeration loop is set by the NH₃ refrigeration
 185 process and the evaporating pressure is set by the liquefaction process; the maximum pressure in the
 186 NH₃ refrigeration loop is set by the condensing temperature; and the discharge pressure from the

187 first NH₃ compressor stage is set by the liquefaction process. This leaves the two remaining pressure
 188 levels in the NH₃ refrigeration compressor and the inter-stage pressure for the CO₂ feed compressor
 189 as variables that can be optimized, giving $n = 3$.

190 In Case 2, the pressure level used in the second and third stages of the NH₃ compressor are
 191 optimization variables along with the inter-stage pressure of the CO₂ feed compressor, also giving
 192 $n = 3$.

193 In Case 3, the inter-stage pressure of the low pressure NH₃ compressor is an optimization
 194 variable along with the inter-stage pressure of the CO₂ feed compressor, making $n = 2$.

195 In Case 4, the discharge pressure for each of the CO₂ compression stages is an optimization
 196 variable apart from the stage that occurs at the liquefaction pressure level, making $n = 4$.

197 When using the GA routine, the compressor stage outlet temperature was set as a constraint in
 198 the optimization algorithm for all Cases:

$$T_i = f(P_i) \leq 150 \text{ }^\circ\text{C for } i = 1, \dots, n, \quad (2)$$

199 where T_i is the temperature at each of the compressor discharge pressure levels that can be
 200 optimized. When *fminsearch* was used, T_i was checked manually and constrained, when necessary,
 201 by applying an appropriate constraint to an individual stage pressure level.

202 Development and selection of the study basis

203 The performance comparisons made using the 'base' parameters shown in Table 1 allowed the
 204 best performing processes to be identified from Cases 1 to 4 using the methods described above. The
 205 comparisons also helped – in some cases – to identify potential improvements to the flow schemes.
 206 In particular, the use of expanders in place of valves in Case 4 was identified as a potential
 207 improvement along with the possibility of an additional stage of cooling in Cases 2 and 4.

208 The application of these features in the flow schemes for Cases 1 to 4 leads to the development
 209 of five new cases. The new cases are called Case 4b, which is Case 4 with expanders replacing valves;
 210 Cases 5a and 5b, which are developed from Case 2 by adding an extra cooling stage (5a with valves
 211 and 4b with expanders); and Case 6, which is developed from Case 4 by adding an additional stage
 212 and valves or expanders. Figures 2 and 3 present the flow scheme for Cases 5 and 6.

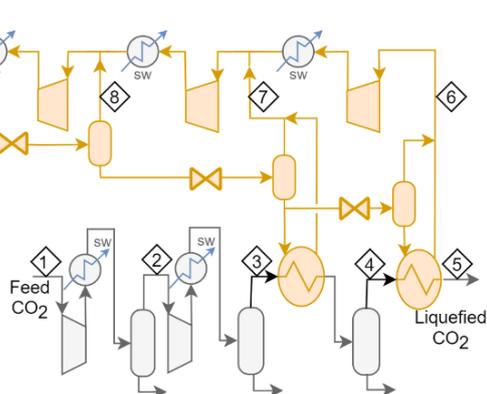


Figure 2. Flow scheme for Case 5a. In Case 5b the each of the three letdown valves is replaced with an expander.

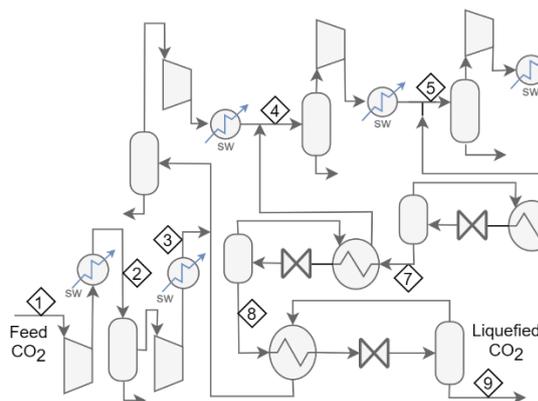


Figure 3. Flow scheme for Case 6a. In Case 6b each of the three letdown valves is replaced with an expander.

213 Each of these new cases were optimized following the same procedure as Cases 1 to 4. The
 214 number of optimization variables, n , for Cases 5a and b are each one less than Case 2 because the
 215 pressure level in the second stage of the NH₃ compressor in Cases 5a and b is set by the temperature
 216 in the CO₂ cooler (i.e. just above 0 °C). This gives $n = 2$ for Cases 5a and 5b. Cases 6a and b have the
 217 same number of optimization variables as Case 4 ($n = 4$). Table 4 presents the results of the
 218 optimization work.

219 Once the optimization process was complete, the best performing schemes were selected for the
 220 final part of the modelling work: the investigation of the impact of ambient temperature on energy
 221 consumption.

222 *Performance variation with cooling temperature*

223 The method used to determine the Cases that would be studied in the final part of the modelling
 224 work was to firstly identify the best performing schemes and then second to consider what might
 225 represent the natural 'next-best' process alternatives taking into consideration of the complexity of
 226 each of the best performing cases. The selected cases were then optimized for temperatures in the
 227 range 15 to 50 °C aftercooler temperature.

228 **3. Results**

229 The main results of this study are presented below.

230 *3.1. Process modelling and validation*

231 The main results of the validation work, which are presented in the Method part of this paper,
 232 show that there is good agreement between the reference studies and the modelling work conducted
 233 here. As a supplement to this, the detailed modelling results for the validation cases are presented in
 234 Appendix A as Tables A1 to A4. The stream numbering used corresponds to that shown in Figure 1.

235 *Performance Comparisons*

236 The energy consumption associated with each of the 'base' process schemes (Cases 1 to 4) is
 237 presented below in Table 3. In all cases, the 'base' operating parameters are used (see Table 1) and
 238 the values of all variable operating parameters optimized to minimize energy consumption.

239 **Table 3.** Summary of optimized power consumption for the base process schemes.

Case	Case 1		Case 2		Case 3		Case 4	
Refrigerant	CO ₂ & NH ₃		NH ₃		NH ₃		CO ₂ -open	
Cycle	cascade		3-stage		dual		2-stage	
Cooling method	valves		valves		expanders		valves	
Cooling stages	2-stage		1-stage		2-stage		n/a	
Pressure (bara)	8.9	15	15	7.0	15	7.0	15	
Energy (kWh/ton)	105	96.0	92.6	103	93.3	106	95.2	

240 The results show that the 15 bara cases offer reduced power consumption in all of the 'base'
 241 process schemes. They also show that a 3-stage NH₃ process has the potential to outperform a 2-stage
 242 expander based process and that a 2-stage open CO₂ cycle can also offer low energy consumption.
 243 Based on these findings, Cases 4b, 5a, 5b, 6a and 6b were developed. The performance of the
 244 optimized versions of these schemes is presented below in Table 4.

245
 246

Table 4. Summary of optimized power consumption for the new process schemes.

Case	Case 4b	Case 5a	Case 5b	Case 6a	Case 6b
Refrigerant	CO ₂ -open	NH ₃	NH ₃	CO ₂ -open	CO ₂ -open
Cycle	2-stage	3-stage	3-stage	3-stage	3-stage
Cooling method	expanders	valves	expanders	valves	expanders
Cooling stages	n/a	2-stage	2-stage	n/a	n/a
Pressure (bara)	15	15	15	15	15
Energy (kWh/ton)	89.5	91.8	90.8	93.0	88.7

247 Table 4 shows that Case 6b offers lowest overall energy consumption when compared at 25 °C
 248 cooling temperature and that the use of expanders in the open CO₂ refrigeration cycle offers a

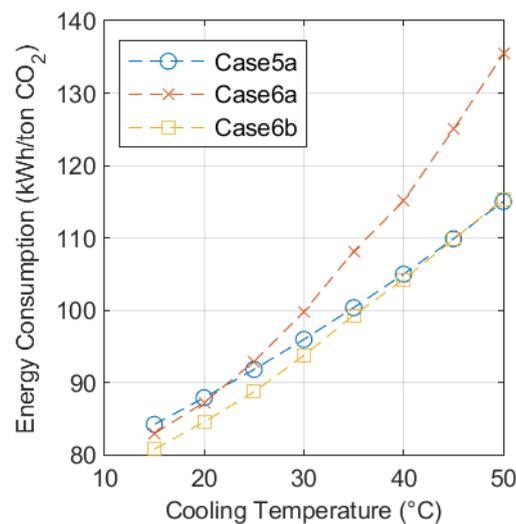
249 significant benefit over the use of valves. In the case of the NH₃ refrigeration process, the benefit of
250 using expanders is not as pronounced.

251 Based the results shown on Tables 3 and 4 three cases were selected for study in the next part of
252 the work: Case 6b, because it offers the lowest overall power; Case 6a because it offers a more
253 conventional alternative to Case 6b; and Case 5a to provide a comparison with a NH₃ based process.
254 Case 5b was not selected because the addition of expanders offered little reduction in energy
255 consumption compared to Case 5a. The detailed modelling results for the cases shown in Tables 3
256 and 4 are presented in Appendix A.

257 *Performance variation with cooling temperature*

258 Figure 4 shows how the energy consumption for Cases 5a, 6a and 6b varies with ambient
259 temperature. The smoothness of the three curves gives an indication of the level of consistency
260 achieved in the optimization process. The results presented in Figure 4 illustrates the significant
261 impact of cooling temperature on process performance and also, how at higher temperatures the
262 performance of Case 5a improves relative to Cases 6a and 6b.

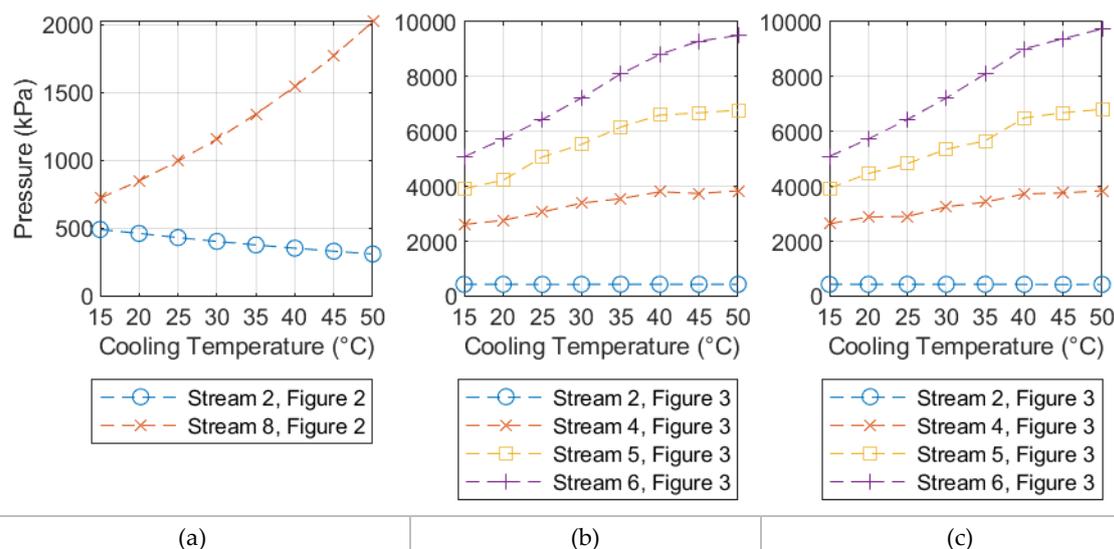
263 Figure 5 shows the variation in the operating pressure parameters for Cases 5a, 6a and 6b. Again,
264 the smoothness of the two curves gives an indication of the level of consistency achieved in the
265 optimization process. In both cases, the inter-stage pressure of the CO₂ feed compressor drops as the
266 cooling temperature increases. This is a result of the 150 °C temperature limits, which represents the
267 optimum condition for inter-stage pressure in all the cases shown here.



268

269

Figure 4. Variation in Energy Consumption for Cases 5a, 6a and 6b with Cooling Temperature.



270

271 **Figure 5.** Variation in the Optimization Parameters for Cases 5a (a), 6a (b) and 6b (c).272 **4. Discussion**

273 Table 2 shows that the modelling approach used in this study is validated by comparison to the
 274 results for other similar studies. Additionally, the smoothness of the curves presented in Figures 4
 275 and 5 show that the optimization of the process parameters for Cases 5a, 6a and 6b has been made
 276 on a consistent basis giving confidence in the trends illustrated in the results.

277 The variation of performance with ambient temperature presented in Figure 4 shows that Case
 278 6a is the process that consumes the least energy across most of the temperature range studied. There
 279 is, however, a trade-off point around 45 °C, where Case 5a becomes the lowest energy consuming
 280 case. Case 6a does not offer the minimum energy consumption in any cases, but it does outperform
 281 Case 5a when the cooling temperature is below 20°C. These trends support the original assertion of
 282 Hegerland [2] that open-cycle CO₂ based systems outperform NH₃ based refrigeration at low
 283 temperatures.

284 It is worth noting that the comparison between Case 5a and 6b is not entirely a fair one since the
 285 machinery required in Case 6b is significantly more complicated than that required in Case 5a. The
 286 potential benefit of a turbo expander based process would ultimately be determined by the optimum
 287 balance between operating and capital costs, which in-turn will vary between projects. Also worth
 288 noting is that Case 4b offers similar performance to Case 6b with a small reduction in process
 289 complexity. Although the life-cycle costing of these schemes falls outside the scope of this study, the
 290 results provided here suggest that all of the Cases 4b, 5a, 6a and 6b would merit consideration in a
 291 techno-economic study of the CO₂ liquefaction process.

292 Cases 5a and 6a could be considered 'conventional' and are also similar in the level of process
 293 complexity involved. Case 6a does, however, have a potentially significant advantage over Case 5a
 294 due to the possibility of condensing water out of the CO₂ stream at pressure levels above the final
 295 liquefaction pressure. This will reduce, and possibly eliminate, the requirement for an additional
 296 dehydration step in the process. Therefore, in Case 5a an additional pressure-drop associated with a
 297 dehydration unit located prior to the liquefaction step may be necessary, representing a small
 298 additional energy penalty for this processes. Again, the level of dehydration required and the energy
 299 penalty this represents would be expected to vary on a project-by-project basis.

300 **5. Conclusions**

301 As suggested by Hegerland [2] the energy consumption of open CO₂ based refrigeration
 302 processes is found in this study to be lower than NH₃ based refrigeration alternatives when ambient
 303 temperature is low. The trade-off temperature varies with the complexity of the CO₂ based system:
 304 simple systems only offering lower energy consumption in locations with very low ambient

305 temperatures; more complex, turbo-expander based, processes offering improved performance over
306 a much wider temperature range.

307 The results presented in Figure 4 show that ambient temperature strongly influences the energy
308 consumption for all of the liquefaction processes studies here. Overall, the rise in energy consumption
309 over the range 15 to 50°C is around 40%. This variation in energy consumption could be expected to
310 be important to the accurate development of CCS system models and the determination of the trade-
311 off point between shipping and pipelines where the potential sources on disposal locations lie in
312 locations with different ambient temperature conditions.

313 The data developed here is suitable for use as the basis for modelling CCS transport systems
314 where the contribution of CO₂ liquefaction processes is an important part of the overall process
315 energy consumption. The data presented in Figure 4 is suitable to form the basis of such studies.

316 **Author Contributions:** Conceptualization, Jackson. S.; methodology, Jackson. S. and Brodal. E.; validation,
317 Jackson. S.; formal analysis, Jackson. S.; investigation, Jackson. S.; resources, Jackson. S.; data curation, Jackson.
318 S.; writing—original draft preparation, Jackson. S.; writing—review and editing, Jackson. S. and Brodal. E.;
319 supervision, Brodal. E.

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322 Appendix A

323 Detailed modelling results for the validation cases

324 **Table A1.** Mass balance for the Case 1 verification model, numbering as Figure 1a.

Case 1	1	2	3	4	5	6	7	8	9	10
Temperature (°C)	38.0	38.0	38.0	1.00	-45.1	38	-48.1	38	-2.7	-16
Pressure (bara)	1.8	3.1	8.2	8.1	8.0	23.9	7.1	14.6	3.8	2.2
Flow (ton/h)	80.5	73.1	72.7	72.5	72.5	103.0	103	31.3	2.8	31.3

325 **Table A2.** Mass balance for the Case 2 verification model, numbering as Figure 1b.

Case 2	1	2	3	4	5	6	7	8	9
Temperature (°C)	35	35	35	35	-28	-36.7	35	7.80	-15.2
Pressure (bara)	1.8	3.6	7.3	15	15	0.83	13.4	5.6	2.3
Flow (ton/h)	114	114	114	114	114	32.2	39.2	35.0	32.2

326 **Table A3.** Mass balance for the Case 3 verification model, numbering as Figure 1c.

Case 3	1	2	3	4	5	6	7	8
Temperature (°C)	20	15	1.0	-50	15	-55	15	-4.0
Pressure (bara)	2.0	8.0	7.5	7.0	7.2	0.30	7.2	3.6
Flow (ton/h)	125	122	122	122	40.8	40.8	1.36	1.36

327 **Table A4.** Mass balance for the Case 4 verification model, numbering as Figure 1d.

Case 4	1	2	3	4	5	6	7
Temperature (°C)	20	10	15	14.4	15	3.7	-48.5
Pressure (bara)	2.0	6.5	11.5	37.5	70.0	38.0	7.0
Flow (ton/h)	125	177	177	200	200	176	122

328 *Detailed modelling results for optimised base cases*329 **Table A5.** Mass balance Case 1,15 bara liquefaction pressure, Stream Numbering from Figure 1.

Case 1 (15 bara)	1	2	3	4	5	6	7	8	9	10
Temperature (°C)	25,0	25,0	25,0	1,0	-27,7	-30,7	25,0	25,0	-2,0	-10,0
Pressure (bara)	1,00	4,01	15,6	15,3	15,0	13,6	28,7	9,95	3,94	2,87
Flow (ton/h)	100	100	100	100	100	130	130	33,0	1,96	33,0

330 **Table A6.** Mass balance Case 2,15 bara liquefaction pressure, Stream Numbering from Figure 1.

Case 1 (15 bara)	1	2	3	4	5	6	7	8	9
Temperature (°C)	25,0	25,0	-	25,0	-27,7	-32,7	25,0	13,0	-1,48
Pressure (bara)	1,00	4,01	-	15,30	15,00	1,03	9,95	6,74	4,01
Flow (ton/h)	100	100	-	100	100	28,7	31,8	30,3	28,7

331 **Table A7.** Mass balance Case 3, 15 bara liquefaction pressure, Stream Numbering from Figure 1.

Case 1 (15 bara)	1	2	3	4	5	6	7	8
Temperature (°C)	25,0	25,0	1,0	-27,7	25,0	-3,96	25,0	-32,7
Pressure (bara)	1,00	15,6	15,3	15,0	9,95	3,65	9,95	1,03
Flow (ton/h)	100	100	100	100	1,96	1,96	29,2	29,2

332 **Table A8.** Mass balance Case 4, 15 bara liquefaction pressure, Stream Numbering from Figure 1.

Case 1 (15 bara)	1	2	3	4	5	6	7
Temperature (°C)	25,0	19,3	25,0	23,6	25,0	9,44	-27,7
Pressure (bara)	1,00	14,7	27,8	43,9	65,0	44,2	15,0
Flow (ton/h)	100	138	138	191	191	138	99,3

333 *Detailed modelling results for new cases*334 **Table A9.** Mass balance for Case 4b, Stream Numbering from Figure 1.

Case 4b	1	2	3	4	5	6	7
Temperature (°C)	25,0	19,8	25,0	23,7	25,0	10,2	-27,7
Pressure (bara)	1,00	14,7	28,0	44,7	65,0	45,0	15,0
Flow (ton/h)	100	135	135	182	182	134	100

335 **Table A10.** Mass balance for Case 5a, Stream Numbering from Figure 2.

Case 5a	1	2	3	4	5	6	7	8	9
Temperature (°C)	25,0	25,0	25,0	1,00	-27,7	-32,7	2,11	12,9	25,0
Pressure (bara)	1,00	4,01	15,6	15,3	15,0	1,03	3,94	6,73	9,95
Flow (ton/h)	100	100	100	100	100	26,8	3,39	1,51	31,7

336 **Table A11.** Mass balance for Case 5b, Stream Numbering from Figure 2.

Case 5b	1	2	3	4	5	6	7	8	9
Temperature (°C)	25,0	25,0	25,0	1,00	-27,7	-32,7	1,89	15,0	25,0
Pressure (bara)	1,00	4,01	15,6	15,3	15,0	1,03	3,93	7,22	9,95
Flow (ton/h)	100	100	100	100	100	26,6	3,57	1,25	31,4

337 **Table A12.** Mass balance for Case 6a, Stream Numbering from Figure 3.

Case 6a	1	2	3	4	5	6	7	8
Temperature (°C)	25,0	19,2	21,5	23,7	25,0	13,4	-5,5	-27,7
Pressure (bara)	1,00	14,7	29,3	48,5	64,4	48,8	29,6	15,0
Flow (ton/h)	100	119	151	201	201	152	121	102

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Table A13. Mass balance for Case 6b, Stream Numbering from Figure 3.

Case 6b	1	2	3	4	5	6	7	8
Temperature (°C)	25,0	19,6	21,6	23,8	25,0	13,2	-5,9	-27,7
Pressure (bara)	1,00	14,7	29,0	48,3	64,4	48,6	29,3	15,0
Flow (ton/h)	100	118	146	192	192	146	118	100

339 **References**

- 340 1. Metz, B.; Davidson, O.; De Coninck, H.; Loos, M.; Meyer, L. *Carbon Dioxide Capture and Storage*; IPCC,
341 2005: *IPCC Special Report on Carbon Dioxide Capture and Storage*; Cambridge University Press: 2005.
- 342 2. Hegerland, G.; Jørgensen, T.; Pande, J.O. Liquefaction and handling of large amounts of CO₂ for EOR.
343 In *Proceedings of the 7th International Conference on Greenhouse Gas Control Technologies*, Elsevier Science
344 Ltd: 2005; pp. 2541-2544.
- 345 3. Mallon, W.; Buit, L.; van Wingerden, J.; Lemmens, H.; Eldrup, N.H. Costs of CO₂ Transportation
346 Infrastructures. *Energy Procedia* **2013**, *37*, 2969-2980, doi:10.1016/j.egypro.2013.06.183.
- 347 4. Jakobsen, J.; Roussanaly, S.; Anantharaman, R. A techno-economic case study of CO₂ capture,
348 transport and storage chain from a cement plant in Norway. *Journal of Cleaner Production* **2017**, *144*,
349 523-539, doi:10.1016/j.jclepro.2016.12.120.
- 350 5. Mitsubishi Heavy Industries Ltd. *Ship Transport Of CO₂*; PH4/30; July 2004, 2004.
- 351 6. Aspelund, A.; Mølnevik, M.J.; De Koeijer, G. Ship Transport of CO₂: Technical Solutions and Analysis
352 of Costs, Energy Utilization, Exergy Efficiency and CO₂ Emissions. *Chemical Engineering Research and*
353 *Design* **2006**, *84*, 847-855, doi:10.1205/cherd.5147.
- 354 7. Decarre, S.; Berthiaud, J.; Butin, N.; Guillaume-Combecave, J. CO₂ maritime transportation.
355 *International Journal of Greenhouse Gas Control* **2010**, *4*, 857-864, doi:10.1016/j.ijggc.2010.05.005.
- 356 8. Roussanaly, S.; Bureau-Cauchois, G.; Husebye, J. Costs benchmark of CO₂ transport technologies for a
357 group of various size industries. *International Journal of Greenhouse Gas Control* **2013**, *12*, 341-350,
358 doi:10.1016/j.ijggc.2012.05.008.
- 359 9. Knoope, M.M.J.; Ramírez, A.; Faaij, A.P.C. Investing in CO₂ transport infrastructure under uncertainty
360 : A comparison between ships and pipelines. *International Journal of Greenhouse Gas Control* **2015**, *41*,
361 174-5836.
- 362 10. Seo, Y.; Huh, C.; Lee, S.; Chang, D. Comparison of CO₂ liquefaction pressures for ship-based carbon
363 capture and storage (CCS) chain. *International Journal of Greenhouse Gas Control* **2016**, *52*, 1-12,
364 doi:10.1016/j.ijggc.2016.06.011.
- 365 11. Jordal, K.; Aspelund, A. Gas conditioning—The interface between CO₂ capture and transport.
366 *International Journal of Greenhouse Gas Control* **2007**, *1*, 343-354.
- 367 12. Roussanaly, S.; Brunsvold, A.L.; Hognes, E.S. Benchmarking of CO₂ transport technologies: Part II –
368 Offshore pipeline and shipping to an offshore site. *International Journal of Greenhouse Gas Control* **2014**,
369 *28*, 283-299, doi:10.1016/j.ijggc.2014.06.019.
- 370 13. Jackson, S.; Brodal, E. A comparison of the energy consumption for CO₂ compression process
371 alternatives. In *Proceedings of Earth and Environmental Science*, Barcelona.
- 372 14. Lee, U.; Yang, S.; Jeong, Y.; Lim, Y.; Lee, C.S.; Han, C. Carbon Dioxide Liquefaction Process for Ship
373 Transportation. *Ind. Eng. Chem. Res.* **2012**, *51*, 15122-15131, doi:10.1021/ie300431z.
- 374 15. Duan, L.; Chen, X.; Yang, Y. Study on a novel process for CO₂ compression and liquefaction
375 integrated with the refrigeration process. *International Journal of Energy Research* **2013**, *37*, 1453-1464.

- 376 16. Alabdulkarem, A.; Hwang, Y.H.; Radermacher, R. Development of CO₂ liquefaction cycles for CO₂
377 sequestration. *Applied Thermal Engineering* **2012**, *33-34*, 144-156,
378 doi:10.1016/j.applthermaleng.2011.09.027.
- 379 17. Øi, L.E.; Eldrup, N.H.; Adhikari, U.; Bentsen, M.H.; Badalge, J.C.L.; Yang, S. Simulation and Cost
380 Comparison of CO₂ Liquefaction. **2016**.
- 381 18. Seo, Y.; You, H.; Lee, S.; Huh, C.; Chang, D. Evaluation of CO₂ liquefaction processes for ship-based
382 carbon capture and storage (CCS) in terms of life cycle cost (LCC) considering availability. 2015; Vol.
383 35, pp 1-12.
- 384 19. Jackson, S.; Brodal, E. Optimization of the Energy Consumption of a Carbon Capture and
385 Sequestration Related Carbon Dioxide Compression Processes. **2019**, *12*, doi:10.3390/en12091603.
- 386 20. AspenTech *HYSYS*, 9; 2016.
- 387 21. Mazzoccoli, M.; Bosio, B.; Arato, E.; Brandani, S. Comparison of equations-of-state with P-p-T
388 experimental data of binary mixtures rich in CO₂ under the conditions of pipeline transport. *The*
389 *journal of supercritical fluids* **2014**, 474-490.
- 390 22. Wilhelmsena, Ø.; Skaugena, G.; Jørstadb, O.; Hailong, L. Evaluation of SPUNG# and other Equations
391 of State for use in Carbon Capture and Storage modelling. *Energy Procedia* **23** **2012**, 236-245.
- 392 23. The MathWorks, I. *MATLAB*, 2018a; Natick, Massachusetts, United States.
- 393