

Article

Not peer-reviewed version

New Insight into Electric Force in Metal and the Quadratic Electrical Resistivity Law of Metals at Low Temperatures

[Vilius Palenskis](#) *

Posted Date: 19 March 2024

doi: 10.20944/preprints202403.1083.v1

Keywords: effective density of randomly moving (RM) electrons; density of states (DOS); mean free electron path; effective electric force; electronic defects; effective scattering cross-section; quadratic electrical resistivity law at low temperatures



Preprints.org is a free multidiscipline platform providing preprint service that is dedicated to making early versions of research outputs permanently available and citable. Preprints posted at Preprints.org appear in Web of Science, Crossref, Google Scholar, Scilit, Europe PMC.

Copyright: This is an open access article distributed under the Creative Commons Attribution License which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

Article

New Insight into Electric Force in Metal and the Quadratic Electrical Resistivity Law of Metals at Low Temperatures

Vilius Palenskis

Faculty of Physics, Vilnius University, Saulėtekio Av. 9, 10222 Vilnius, Lithuania; vilius.palenskis@ff.vu.lt

Abstract: Considering that Einstein's relation between the diffusion coefficient and the drift mobility of the free randomly moving charge carriers in homogeneous materials including metals is always valid, it is shown that the effective electric force acting on free electrons in metal depends on the ratio between the kinetic free electron energy at the Fermi surface to the classical particle energy $3kT/2$. The electrical resistivity of elemental metals dependence on very low temperatures has the quadratic term, which has been explained by electron-electron scattering. In this paper, it is shown that the quadratic term of the electrical resistivity at low temperatures is caused by scattering of the free RM electrons by electronic defects due to linear effective free electron scattering cross-section dependence on temperature, but not by electron-electron scattering.

Keywords: effective density of randomly moving (RM) electrons; density of states (DOS); mean free electron path; effective electric force; electronic defects; effective scattering cross-section; quadratic electrical resistivity law at low temperatures

1. Introduction

An investigation of the electron transport in metals began over a hundred years ago when the free electrons were postulated by Drude [1] to explain the conductivity of metals. At the same time, the properties of the electrons in metals have been analyzed by using the classical statistical methods of the gases. It was assumed that the density of the free electrons is equal to the density of atoms, and the mean free path of the electrons can be determined from comparison with an experiment. The Drude theory failed to explain why the electron in a pure metal can freely travel over a hundred atomic distances without being scattered. Later a mental imagination that all metal valence electrons become delocalized and are shared by all atoms in the solid forming a “sea” of negative charge [2,3], and are no longer associated with a given atom [4]. The nuclei with their core electrons form ions, which are immersed in this sea of valence electrons. Such ungrounded imagination about free electron transport in metals has continued for nearly the last century.

Usually, metal atoms are arranged in regular three-dimensional periodic lattices. Hard spheres of metal atoms are packed in such crystal structures: face-centered cubic (FCC), body-centered cubic (BCC), hexagonal close-packed (HCP), orthorhombic (ORC), and tetragonal (TET). The dependences of the metal atoms to various lattice structures and their distribution on the periodic table are presented in Figure 1 [5,6].

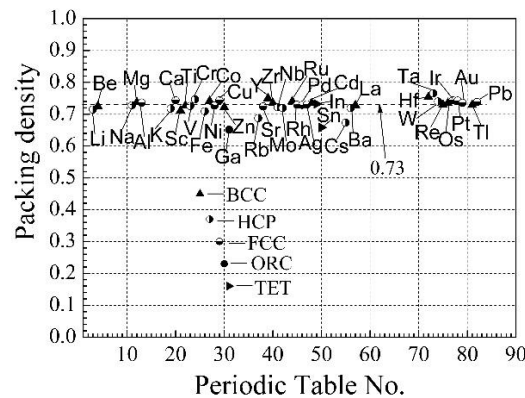


Figure 1. Packing density of various elemental metal distribution on the periodic table. Here lattices: BCC – body-centered cubic; HCP – hexagonal close-packed; FCC – face-centered cubic; ORC – orthorhombic; TET – tetragonal. The dashed line (0.73) corresponds to the ideal close-packed lattice.

As can be seen from Figure 1, almost all metal atom packing density is close to the ideal close-packing value of 0.73, and it does not depend on the lattice structure, only for Ga, Rb, Cs, and Sn the packing density is smaller than the ideal close packing density value.

The interaction of the valence electrons with the ion cores produces the main contribution to the binding energy and lowering of the energy of the valence electrons in metal compared with that of the free atom causes the metallic binding [7–9]. The total number of bonds in metal is considerably large, and not all valence electron bonds have the same binding energy.

Some difficulties in the explanation of the electron transport in metals by classical statistics have been understood by quantum mechanics: the Pauli exclusion principle and Fermi-Dirac statistics. According to quantum mechanics the free electrons as the Bloch waves can freely move in the ideal periodic lattice of the metal without any scattering by lattice ions [3,7–10]. The scattering of the free electrons can be only in spots where there are distortions of the periodicity of the potential energy of the ideal lattice structure.

The electrons in an ideal periodicity crystal are arranged in energy bands (allowed zones) separated by forbidden regions in energy (energy gaps). Each primitive cell exactly contributes only one independent value of the wave vector \mathbf{k} to each energy band [10–12]. The valence electrons occupy the lowest energy levels in the conduction band below the Fermi level energy, and their distribution on energy E is described by Fermi distribution function $f(E)$. Though the valence electron wave functions overlap with that of neighbor atoms [6–8], they remain associated with native ion cores. Only valence electrons, which have energies close to the Fermi level energy due to lattice atom vibrations can be released, and can randomly move in metal [13,14]. The effective density n_{eff} of the free randomly moving (RM) electrons in metals and other materials has been defined in [15] as

$$n_{\text{eff}} = \int_0^{\infty} g(E)f(E)[1 - f(E)]dE \quad (1)$$

where $g(E)$ is the density of states (DOS) in the conduction band. It let to show that the effective density n_{eff} of the free RM electrons in metals can be described as $n_{\text{eff}} = g(E_F)kT$, where $g(E_F) = g(E)$ at $E=E_F$. It also lets us show that the Einstein's relation between the diffusion coefficient D and the drift mobility μ_{drift} of the free RM charge carriers $D/\mu_{\text{drift}} = kT/q$ is valid in all cases with one type of charge carriers. In [16], there is at the first time presented the probability density function of the free RM electron distribution on energy: $p(E) = f(E)[1 - f(E)]/kT$, which meets all requirements of the probability theory [17,18].

Though the value of the n_{eff} reaches only a few per cents of the total density n of the valence electrons of metals at room temperature, the total density of the valence electrons n has been traditionally used for description of the electrical conductivity σ in metals and superconductors in the normal state by $\sigma = q^2 n \tau / m^*$ where q , τ , and m^* are, respectively, the electron charge, the

average electron relaxation time, and its effective mass. From a comparison of the Drude and Sommerfeld conductivity formulae, it has been stated that there is only a difference in the definition of the relaxation time of the free electrons [10]. The Drude conductivity formula can be used for descriptions of the conductivity of weakly doped semiconductors, but it is completely not applicable to describe the conductivity of metals and other materials with degenerate electron gas.

The resistivity of metals linear increase with a temperature above Debye's temperature was usually explained by charge carriers scattering due to lattice atom vibrations: due to an increase of the electron scattering cross-section [6,9–12,19–22]. As shown in [13,14], the electron scattering cross-section at a temperature above the Debye temperature does not depend on temperature, and the lattice vibrations play another role: they produce the free RM electrons, and at the same time they produce the electronic defects (weakly screened ions, which lost valence electrons). The effective density of the electronic defects $N_{\text{eff}} = n_{\text{eff}}$. The rest part $N - N_{\text{eff}}$ of atoms which have the valence electrons with energy lower than the Fermi level energy does not have sufficient thermal vibration energy to excite their valence electrons. As it is pointed in [13], the rest part $n - n_{\text{eff}}$ of the valence electrons, which energy is lower by a few kT than Fermi level energy, due to the Pauli exclusion principle and Fermi-Dirac statistics can not change their energy, and they cannot scatter the free RM electrons.

2. Effective Drift Velocity of the Free RM Electrons, and Effective Electric Force Acting to Free Electrons in Metals

The current density \mathbf{j} in homogeneous material is described as

$$\mathbf{j}_x = \sigma \mathbf{E}_x = q n_{\text{eff}} \mathbf{v}_{x \text{ drift}} = q n_{\text{eff}} \mu_{\text{drift}} \mathbf{E}_x, \quad (2)$$

where σ is the conductivity of the material, E_x is the external applied electric field strength in the x -direction, n_{eff} is the effective density of the free randomly moving electron density, $\mathbf{v}_{x \text{ drift}}$ is their drift velocity in the electric field, and μ_{drift} is the free electron drift mobility.

Considering that Einstein's relation between the diffusion coefficient D and the drift mobility μ_{drift} is valid for free RM electrons in metal, one can express the drift mobility of the free RM electrons in the following way:

$$\mu_{\text{drift}} = \frac{qD}{kT} = \frac{q \langle v^2 \rangle \langle \tau \rangle}{3kT} = \frac{q \langle \tau \rangle}{m^*} \cdot \frac{\langle E \rangle}{(3/2)kT} = \frac{q \langle \tau \rangle}{m^*} \cdot \alpha_\varepsilon, \quad (3)$$

where the parameter $\alpha_\varepsilon = \langle E \rangle / [(3/2)kT]$ shows how many times the average free RM electron kinetic energy exceeds the average classical particle energy $(3/2)kT$. Therefore, the drift velocity of the free RM electron can be described as

$$\mathbf{v}_{x \text{ drift}} = \frac{\langle \tau \rangle}{m^*} \cdot \alpha_\varepsilon q \mathbf{E}_x = \frac{\langle \tau \rangle}{m^*} \mathbf{F}_{x \text{ eff}}, \quad (4)$$

where $\mathbf{F}_{x \text{ eff}}$ is the effective electric force acting on the free RM electron in the material. In the case of semiconductors with non-degenerate electron gas, $\alpha_\varepsilon = 1$, and

$$\mathbf{F}_{x \text{ eff}} = q \mathbf{E}_x, \quad (5)$$

i. e., the effective electric force is described by the usual relation. Though the average drift velocity contribution to the direct current in materials without degenerate electron gas with and without accounting the electron distribution on energy gives the same result, the electrons with higher velocities give relatively higher contribution to the total current. In the case of metals, the effective electric force acting on the free RM electron is equal to

$$\mathbf{F}_{x \text{ eff}} = \alpha_\varepsilon q \mathbf{E}_x = \frac{E_F}{(3/2)kT} q \mathbf{E}_x. \quad (6)$$

Impulse of force \mathbf{I} acting on the free electrons during their scattering by electronic defects

$$\mathbf{I} = \mathbf{F}_{\text{eff}} \Delta t = \alpha_\varepsilon q \mathbf{E}_{\text{def}} \Delta t, \quad (7)$$

where Δt is the acting time, E_{def} is the electric field strength produced by electronic defect. On the other hand, $\mathbf{F}_{\text{eff}} = \Delta \mathbf{p} / \Delta t = \hbar \Delta \mathbf{k} / \Delta t$ (here \mathbf{p} is the momentum, and \mathbf{k} is the wave vector of the electron. If the average energy of the free electron during scattering change is about $1.64kT$ [14], the wave vector due to parameter α_ε experiences a very large exchange of the direction.

Thus, the free electron motion in material must be described by such general equation:

$$m \frac{d\mathbf{v}_{\text{drift}}}{dt} = q\alpha_\varepsilon \mathbf{E} - \frac{m\mathbf{v}_{\text{drift}}}{\langle \tau \rangle}, \quad (8)$$

which is valid for homogeneous materials at any degree of degeneration of electron gas. In stationary state $d\mathbf{v}_{\text{drift}}/dt=0$, and $\mathbf{v}_{\text{drift}}$ is described by Eq. (4).

Similar, the free electron motion in electric and magnetic fields can be described as

$$m \frac{d\mathbf{v}_{\text{drift}}}{dt} = q(\alpha_\varepsilon \mathbf{E} + \mathbf{v} \times \mathbf{B}) - \frac{m\mathbf{v}_{\text{drift}}}{\langle \tau \rangle}, \quad (9)$$

where \mathbf{B} is the magnetic flux density. For static electric and magnetic fields, this equation can be presented as

$$q(\alpha_\varepsilon \mathbf{E} + \mathbf{v} \times \mathbf{B}) = \frac{m\mathbf{v}_{\text{drift}}}{\langle \tau \rangle}. \quad (10)$$

The solution of this equation for the Hall coefficient R_H gives the well-known result [14]:

$$R_H \sigma_0 = \frac{q\langle \tau \rangle}{m^*} r_H = \mu_H, \quad (11)$$

where σ_0 is the material conductivity at $\mathbf{B}=0$, r_H is the Hall factor, and μ_H is the Hall mobility. So, in the general case $\mu_{\text{drift}} = \mu_H \alpha_\varepsilon / r_H$, for non-degenerate electron gas $\mu_{\text{drift}} = \mu_H / r_H$, and for metals and materials with highly degenerate electron gas $\mu_{\text{drift}} = \mu_H \alpha_\varepsilon$, and can many times exceed the Hall mobility.

The fact that the free electron mean path is a hundred times larger than the atomic distance, especially at low temperatures, where the mean free path is over tens millimeters, which shows that the displacement of atoms from their equilibrium positions due to thermal vibrations weakly distorts the potential periodicity in the lattice. The free RM electrons are scattered only at electronic defects with absorption and emission of phonons, usually, a single phonon takes place, because multiple-phonon process contribution is scarcely probable. It is related to the fact that at thermal equilibrium conditions, the lattice atom vibration energy is small. On the other hand, the electron energy distribution is limited by the Pauli exclusion principle and the Fermi-Dirac statistics. The statement that in a perfect crystal below the Debye temperature, the scattering angle becomes smaller and smaller and that it causes the resistivity T^5 -law at low temperatures, is ungrounded. As it follows from Eq. (7), the free electron wave vector during scattering at electronic defect change $\Delta \mathbf{k}$ is very remarkable due to multiplier α_ε , and after each collision with the electronic defect, the free electron wave vector is completely randomized.

3. The Study of the Electrical Resistivity Law AT^2 of Metals at very Low Temperatures

The idea that electron-electron scattering contributes to electrical and thermal resistivities of metals was proposed by Landau and Pomeranchuk [23], but about 20 years the resistivity dependence $\rho_{ee}(T) = A_{ee}T^2$ has not been observed in simple metal. On the base of the conservation of the quasi-momentum and energy of free electrons, it has been proposed that the electron-electron scattering rate changes as T^2 , and as a consequence it must give the law T^2 of the electrical resistivity at very low temperatures. According to [9], the electron-electron scattering rate $\tau_{ee}^{-1}(\varepsilon)$ can be described as

$$\frac{1}{\tau_{ee}} = A_0 \frac{1}{\hbar} \frac{(kT)^2}{E_F}, \quad (12)$$

where A_0 is the non-dimensional quantity. On the base of this relation, it is believed that relaxation time changes as $1/T^2$, and that it can take part only at very low temperatures in very pure metals.

From it also follows that temperature dependence of resistivity $\rho(T)$ for a three-dimensional metal due to electron-electron scattering can be described as $\rho(T) = AT^2$.

A detailed analysis of the electron-electron scattering and its contribution to the electrical resistivity for conducting materials is presented in [24]. The problem of electron-electron scattering in metals also have been solved by using the Boltzmann kinetic equation [25]. Recently [26], it has been shown that Eq. (12) is correct with $A_0 = N_{\text{imp}}/4\pi^2$, where N_{imp} is the number of impurities (or scattering centers) in a given sample. The effects of electron-electron scattering in metals also have been investigated theoretically in both a simple tight-binding model and in a transition metal tungsten by calculating the electron scattering rate at the Fermi surface [27].

Considering that the contribution of the term AT^2 to resistivity at very low temperatures is very small, it was needed the extremely precise measurement technique, better than p.p.m. relative accuracy in the measurement of the low-temperature electrical resistivity, which appeared only a half-century ago [24,28–30]. High-resolution measurements of the electrical resistivity of simple metals at low temperatures have yielded new data that were in contradiction to the generally accepted theories. It has been found that the electrical resistivity does not follow the expected AT^5 law for any metal at very low temperatures.

A detailed analysis of the electrical resistivity of the alkali metals at low temperatures is presented in [24,31,32]. It is shown that for potassium at low temperatures, the phonon drag can play an important role leading to an exponential decrease with temperature [33,34]. It has been believed that the experimental parameter A_{exp} would be the same for all samples of the same metal, but the values A_{exp} are different about ten times. For lithium, it has been expected that values A_{exp} would be similar to K, but the observed values are more than an order larger [24,35,36].

The electrical resistivity of noble metals at low temperatures has been investigated by many authors [37–40]. It has been shown that the electrical resistivity of copper and silver at low temperatures has approximate T^4 behavior, and the term T^2 appears only below 2 K [37]. From the measurement of the electrical resistivity of high-quality copper whiskers taking account of the surface scattering at different magnetic fields, it has been estimated the parameter A for the electrical resistivity of bulk copper [38]. Temperature dependence of the scattering rate in copper also has been investigated by radio frequency size effect to extract the contribution of electron-electron collisions [39]. The observed T^4 behavior for the low-temperature electrical resistivity of Ag is related to the simultaneous presence of both electron-phonon and electron-electron scattering [40]. To observe the term T^2 in noble metals it is needed very low temperatures and very pure metals.

Aluminum is a simple polyvalent metal, which electron-electron scattering has been intensively investigated [41–44]. In [41], it has been presented that the electrical resistivity of aluminum at low temperatures has the term AT^2 . The surface scattering contribution for electrical resistivity of high-purity aluminum samples in the thickness range from 0.1 μm to 7 mm has been investigated in [42]. It is shown that surface scattering of the conduction electrons affects the temperature dependence of the resistivity for samples less than 0.5 mm in thickness. The electrical resistivity term AT^2 is attributed to electron-electron scattering, and it is shown that coefficient A practically does not depend on the residual resistivity value [43]. The influence of the electron-phonon interaction on electron-electron scattering is described within the framework of the Landau Fermi-liquid theory, and electron-electron scattering contribution to the low-temperature resistivity of aluminum can be enhanced by a factor of ~ 20 due to electron-phonon interaction [44].

Due to that transition group metals differ from simple metals, which have partially filled d - and s -band electrons it has been believed that for transition metals the major contribution to resistivity $\rho_{\text{ee}}(T)$ should be caused by highly mobile s -electrons being scattered by the “heavy” d -electrons, i. e. the so-called sd -scattering. It is observed that coefficient A for transition metals is often one or two orders of magnitude larger than for simple metals. Quadratic temperature dependence for $\rho_{\text{exp}}(T)$ for many transition metals has been demonstrated at low temperatures [45–50], but interpretations are different. In [45], it is shown that there is some correlation between coefficient A_{exp} and the electronic heat capacity parameter γ . The electron-electron scattering in tungsten has been investigated by using radio-frequency size-effect for electron scattering rate estimation [46]. The term

AT^2 has been found over the temperature range 1.5–5 K for molybdenum, it was concluded that the quadratic term in the temperature dependence of the scattering rate in molybdenum is due to the electron-electron scattering [47]. The electrical resistivity of a single crystal of vanadium at low temperatures has been analyzed in terms of electron-impurity, electron-electron, and electron-phonon scatterings [48]. They have obtained the term AT^2 with coefficient $A \approx (1.6 - 0.2) \cdot 10^{-11} \Omega \text{cm/K}^2$. The effects of electron-electron scattering in vanadium have been investigated theoretically, and the calculated scattering rates agree with experimental measurement results [49]. The electron-electron scattering in high-purity single crystals of molybdenum has been measured at low temperatures [50], it is shown that when approaching an ideal metallic crystal structure, there have been observed the deviations from the Ohm's law.

The traditional explanation [10] for the large magnitude of A_{exp} due to sd -scattering in transition metals is based on the prepositions that the effective mass of the d -electron is very high, implying that its velocity v_d is very low compared with the velocity v_s of the s -electron. These arguments are very questionable because the drift mobility and drift velocity of metals (see Eqs. (2) and (3)) do not depend on the electron mass. All electrons at the Fermi surface have the same Fermi level energy and are randomly moving with the same Fermi velocity. Often in many cases there is confused the large density of states at the Fermi surface with the effective mass of electrons.

There we want to point attention to the fact that conservation of the quasi-momentum and energy of free electrons due to electron-electron scattering do not produce energy losses, and consequently, this process is elastic and has no affect on material resistivity. It is confirmed by the thermal noise relation with the material resistivity [51], and by the fluctuation-dissipation theorem [52,53]. Therefore, almost for all elemental metals the electrical resistivity at low temperatures has the term AT^2 , but its explanation by electron-electron scattering is doubtful.

In [13,14], it has been shown that the lattice atom vibrations produce not only the effective density of free randomly moving electrons $n_{\text{eff}} = g(E_F)kT$ (here $g(E_F)$ is the density of states at the Fermi level energy), but also the same density of electronic defects (not completely screened ions) is produced $N_{\text{eff}} = n_{\text{eff}}$. In Figure 2, there are demonstrated the electronic defect dependence on temperature for aluminum, and for transition group metal tungsten. As can be seen, the effective density of the electronic defects even at temperature 1 K is higher than 10^{18} cm^{-3} , and these electronic defects cannot be, in principle, decreased as it can be done with the residual defects.

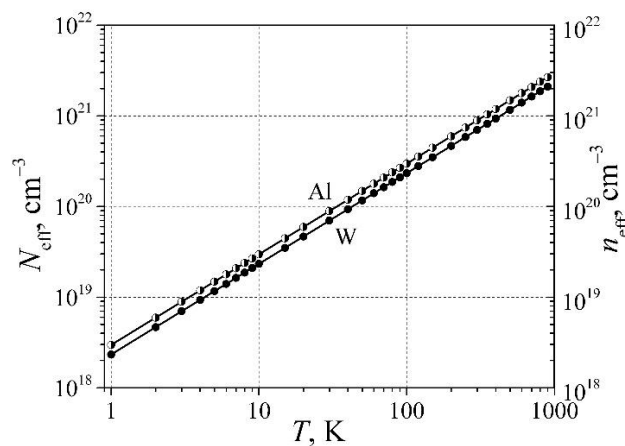


Figure 2. The electronic defect density N_{eff} (left scale), and free RM electron density n_{eff} (right scale) dependences on the temperature for metals Al and W.

As it is shown in [13,14], the resistivity of the elemental metals dependence on temperature can be described as

$$\rho(T) = \rho_0 + \rho(T_0) \cdot (T/T_0) \cdot \eta_{\text{ph}}(T/\Theta), \quad (13)$$

where Θ is the Debye temperature, $\eta_{\text{ph}}(T/\Theta)$ is the phonon mediation factor accounting the free RM electron scattering by electronic defects:

$$\eta_{\text{ph}}(T/\Theta) = (T/\Theta)^4 \int_0^{\Theta/T} \frac{4x^5 dx}{(e^x - 1)(1 - e^{-x})} \quad (14)$$

The resistivity $\rho(T_0)$ at $T=T_0$ is described as

$$\rho(T_0) = \frac{1}{q^2 g(E_F) D} = \frac{3}{q^2 g(E_F) l_F v_F} = \frac{3 \sigma_{\text{eff}0} k T_0}{q^2 v_F} \quad (15)$$

here we used $l_F = 1/(\sigma_{\text{eff}0} N_{\text{eff}}) = 1/[\sigma_{\text{eff}0} g(E_F) k T_0]$, where $\sigma_{\text{eff}0}$ is the effective scattering cross-section estimated at temperature T_0 . The value of the temperature T_0 must be taken from the linear resistivity dependence range on temperature. Then the effective scattering cross-section is estimated as

$$\sigma_{\text{eff}}(T) = \sigma_{\text{eff}0} \eta_{\text{ph}}(T/\Theta). \quad (16)$$

For further study of the resistivity dependence on temperature, we shall investigate the elemental metals from various sites of the periodic tables: the alkali group metal Li, the noble metal Cu, the second group IIB metal Zn, the third group IIIA metal Al, and the transition group metals Mo and W. The resistivity dependences on temperature for these metals are presented in Figure 3: dots are the experimental results, and the solid lines are calculated by Eq. (13). As can be seen, the resistivity $\rho(T)$ of metals has three clearly expressed parts: i) at low-temperature so-named residual resistivity caused by impurity and structural imperfections of metal, ii) so-named phonon-assisted part with $\rho \sim T^5$, which at higher temperatures crosses to iii) the linear resistivity dependence on temperature.

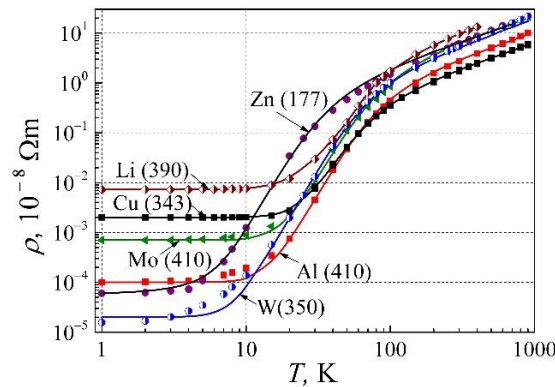


Figure 3. Resistivity dependence on temperature of elemental metals Al, Cu, Li, Mo, W, and Zn. Dots are experimental data from Handbook [54], and solid lines are calculated by Eq. (13); the number near the chemical symbol is the Debye's temperature of this metal used for calculations.

Even though the entire description of the resistivity of metals by Eq. (13) is sufficiently good, at temperatures below 20 K there is a small inadequacy for metals Al, Zn, Mo, and W. Considering the earlier presented experimental results of the electrical resistivity term AT^2 , we shall see further, what causes such electrical resistivity dependence on very low temperatures.

Considering the so-named law $\rho = AT^2$ at low temperatures, we shall carefully investigate the resistivity dependence on low temperatures on the sample of Al. For this aim, we shall find what characteristics cause such resistivity $(\rho(T) - \rho_0)$ dependence on temperature. In Figure 4 there are presented the resistivity $\rho(T)$ and $\rho(T) - \rho_0$ dependences on low temperatures for aluminum.

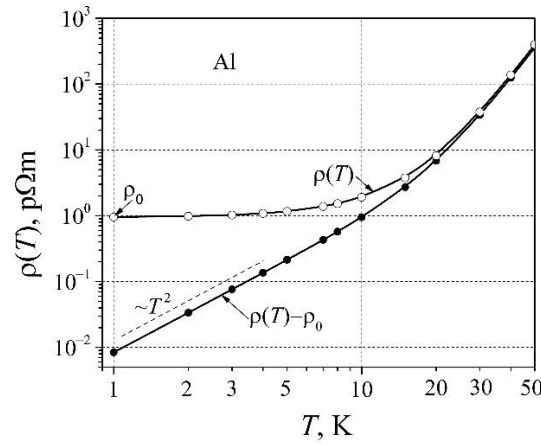


Figure 4. The aluminum electrical resistivity dependence on temperature at low temperatures.

As can be seen, the resistivity $(\rho(T) - \rho_0)$ without residual resistivity ρ_0 has a well-expressed dependence on temperature term AT^2 , where proportionality coefficient $A \approx 6.5 \text{ f}\Omega\text{m K}^{-2}$. Accounting that the resistivity of metals is described as $\rho = (1/\sigma) = 1/[q^2 g(E_F)D]$, where D is the diffusion coefficient of free RM charge carriers. The diffusion coefficient of free RM charge carrier dependence on temperature is shown in Figure 5a. As can be seen, the diffusion coefficient at very low temperatures changes as T^{-2} , and then crosses to the proportionality T^{-5} as the temperature is over 20 K.

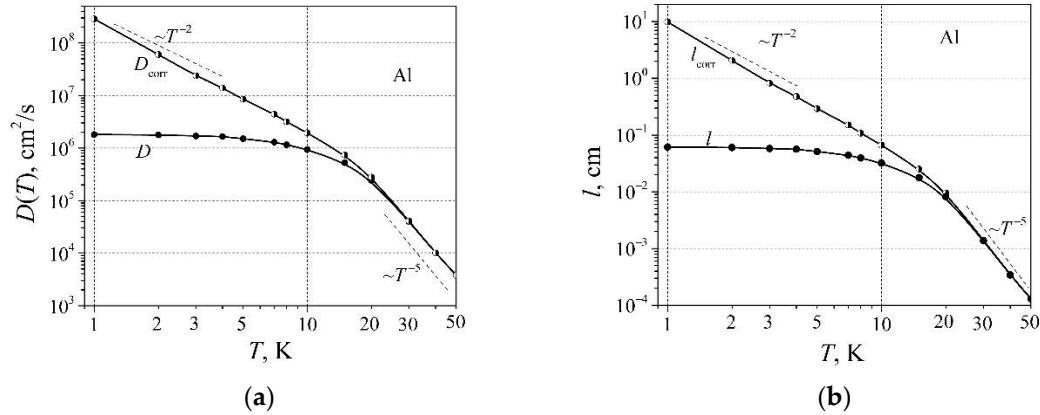


Figure 5. Diffusion coefficient (a) and mean free path (b) of free RM charge carrier dependence at low temperatures for Al. $D = 1/[\rho(T) q^2 g(E_F)]$, $D_{\text{corr}} = 1/[(\rho(T) - \rho_0) q^2 g(E_F)]$, and $l = 3D/v_F$, $l_{\text{corr}} = 3D_{\text{corr}}/v_F$.

The mean free path of charge carriers has been estimated as $l = 3D/v_F$, and $l_{\text{corr}} = 3D_{\text{corr}}/v_F$, which are presented in Figure 5b. As can be seen, the mean free path l_{corr} at very low temperatures changes as T^{-2} , and exceeds several centimeters. It seems that precise resistivity measurements should be done with a very large sample that the transverse dimensions which should greatly exceed the mean path of the charge carriers in order to vanish the charge carrier scattering from the sample boundaries.

The relaxation time of charge carriers has been evaluated as $\tau = l/v_F$ and $\tau_{\text{corr}} = l_{\text{corr}}/v_F$, which are presented in Figure 6a. It is believed, that the relaxation time τ_{corr} changes with temperature as T^{-2} at low temperatures. Now let's estimate the effective scattering cross-section of free RM electrons by electronic defects:

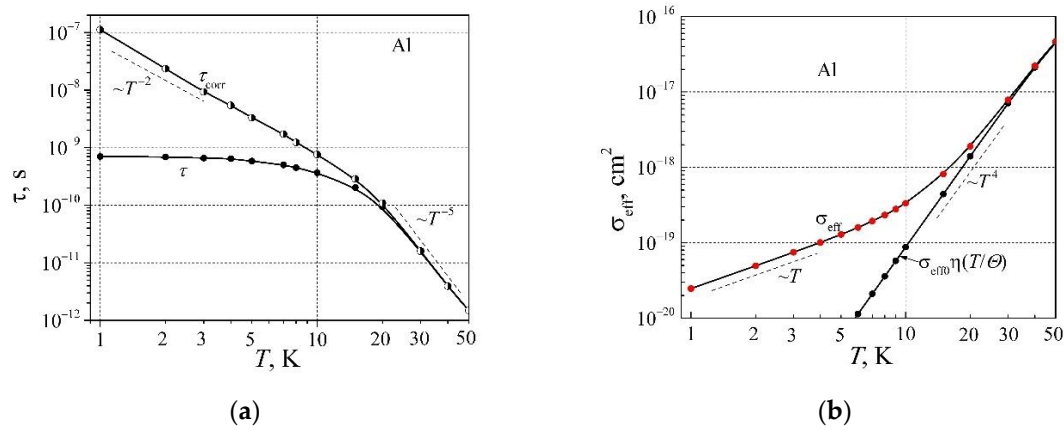


Figure 6. The relaxation time (a) and the effective scattering cross-section (b) dependences on the temperature at low temperatures for aluminum (the σ_{eff} is described by Eq. (17), and $\sigma_{\text{eff}0}\eta_{\text{ph}}(T/\Theta)$ by Eq. (16)).

$$\sigma_{\text{eff}} = \frac{1}{l_{\text{corr}}N_{\text{eff}}} = \frac{1}{l_{\text{corr}}g(E_F)kT}, \quad (17)$$

where N_{eff} is the electronic defect density. The effective scattering cross-section dependence on temperature is presented in Figure 6b. As can be seen, the effective scattering cross-section at very low temperatures is proportional to temperature T , and at higher temperatures coincides with $\sigma_{\text{eff}0}\eta_{\text{ph}}(T/\Theta)$, and is proportional to T^4 . It seems that the resistivity term AT^2 at low temperatures, the solution key is related to the linear effective scattering cross-section dependence on temperature.

In Figure 7, there are presented the heat capacity dependence on temperature for conductors (Pt and W), and pure Si, which at low temperatures is an insulator. The heat capacity of the insulator is described by the Debye's model, for which the heat capacity at low temperatures increases as T^3 , or the thermal energy (quantity of heat) increases as T^4 . In the case of metals, the heat capacity at low temperatures changes slower, approximately linearly, and thermal energy changes as T^2 . As can be seen, to increase the metal of the one mole mass from 1 K to 5 K, it is needed a larger amount of heat, then it is needed for the insulator. This difference is related to the free electrons random moving, and due to their scattering by electronic defects. In equilibrium, the thermal energies between free electrons and lattice atoms are changing by the interaction of free electrons with the electronic defects.

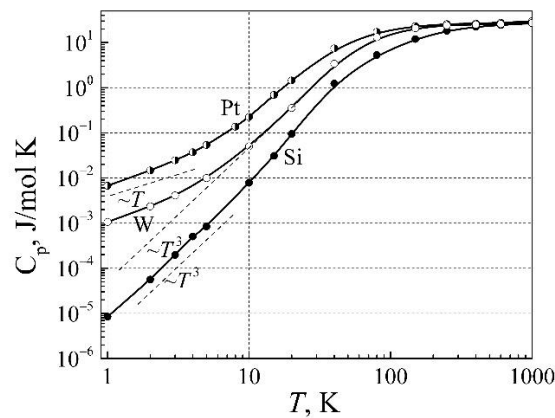


Figure 7. Heat capacity dependence on temperature for conductors (Pt and W), and silicon.

Earlier for estimation of the phonon mediation factor $\eta_{\text{ph}}(T/\Theta)$ (Eq. 14) has been accounted only for the lattice thermal energy based on the Debye model. Additional thermal energy for

conductors is caused by free randomly moving charge carriers, which are scattered by electronic defects. It gives a correction to phonon mediation factor $\eta_{ph}(T/\Theta)$:

$$\eta(T) = (T/\Theta)^4 \int_0^{\Theta/T} \frac{4x^5 dx}{(e^x - 1)(1 - e^{-x})} + \Delta\eta_{ph}(T) \quad (18)$$

where the correction factor

$$\Delta\eta_{ph}(T) = \frac{\Delta E_{ph1}}{E_{e1}} \quad (19)$$

is equal to the ratio between the ratio of the thermal (phonon) energy increase ΔE_{ph1} of the metal atom caused by free RM electron interaction with electronic defects and the average free RM electron energy $E_{e1} = 1.64kT$ at low temperatures. The variation of the thermal energy of the metal atom due to interaction with free randomly moving electrons can be estimated as

$$\Delta E_{ph1} = [(\pi^2/6)g(E_F)(kT)^2/N_A], \quad (20)$$

where N_A is the Avogadro constant. The Eq. (20) can be rewritten as

$$\Delta E_{ph1} = (1/2)\gamma T^2/N_A, \quad (21)$$

where γ is the electronic heat capacity parameter. Then the correction factor can be described in the following form:

$$\Delta\eta_{ph}(T) = \frac{\gamma T^2}{2E_{e1}N_A} = \frac{\gamma T^2}{3.28kTN_A} = \frac{\gamma T}{3.28R}, \quad (22)$$

where $R = kN_A = 8.31 \text{ J/(mol K)}$ is the universal gas constant.

Therefore, the corrected phonon mediation factor can be presented as

$$\eta(T) = \frac{\gamma T}{3.28R} + (T/\Theta)^4 \int_0^{\Theta/T} \frac{4x^5 dx}{(e^x - 1)(1 - e^{-x})}, \quad (23)$$

and the resistivity can be described as

$$\rho(T) = \rho_0 + \rho(T_0) \cdot (T/T_0) \cdot \eta(T), \quad (24)$$

and

$$\rho(T) - \rho_0 = \rho(T_0) \cdot (T/T_0) \cdot \eta(T), \quad (25)$$

Now let us describe the elemental metals from different columns of the periodic table Al, W, Mo, Zn, Li, and Cu resistivity dependences on a very wide temperature range accounting the obtained correction factor. In Figure 8a, there are presented resistivity and corrected phonon mediation factor dependences on temperature for aluminum, and in Figure 8b for wolfram. In Figure 8, the small black dots are the experimental dots of the resistivity [54], and the solid by the black line calculated by Eq. (24); the red dots are experimental data of $(\rho(T) - \rho_0)$, the red solid line is calculated by Eq. (25), the corrected phonon mediation factor $\eta(T)$ is calculated by Eq. (23), the dash line proportional to T^{-4} is estimated by Eq. (14) the phonon mediation factor without correction. From the comparison of the resistivity ($\rho(T)$) curves in Figures 2 and 8a, it is seen that Eq. (24) very well describes the experimental results below 25 K temperature. As can be seen from Figure 8a, at low temperatures the resistivity $(\rho(T) - \rho_0) \approx AT^2$, where $A \approx 6 \text{ f}\Omega\text{m/K}^2$.

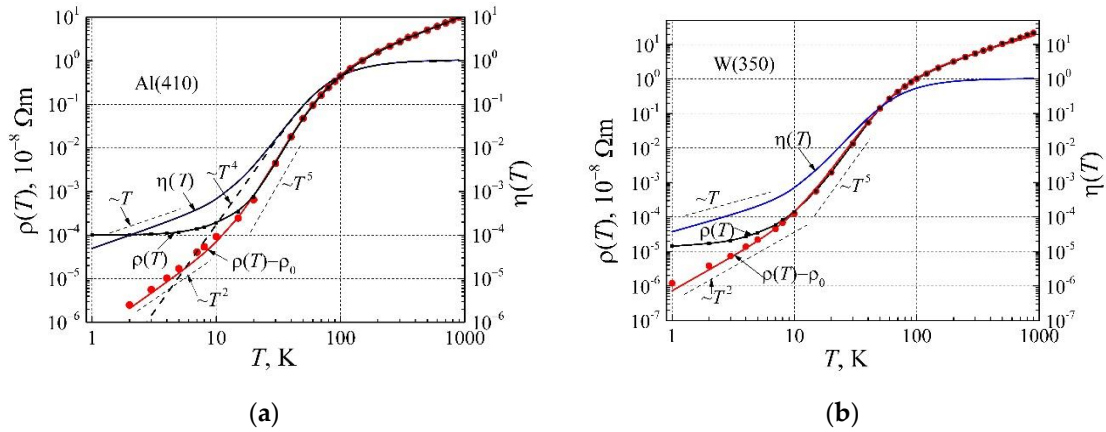


Figure 8. Resistivity (left scale), and corrected phonon mediation factor $\eta(T)$ (right scale) dependences on temperature for aluminum (a), and for wolfram (b).

The similar results to aluminum shown in Figure 8a, in Figure 8b there are presented the investigation results for wolfram. Though the wolfram has a two times larger number of valence electrons than aluminum, the DOS $g(E_F)$ of wolfram is about 1.3 times smaller than that of aluminum. The residual resistivity of a given sample of wolfram is about ten times smaller than that of aluminum. The difference between resistances $\rho(T)$ and $(\rho(T) - \rho_0)$ at low temperatures for aluminum begins at $T < 20$ K, while for wolfram at $T < 8$ K, and the coefficient $A \approx 8$ f $\Omega\text{m}/\text{K}^2$. The phonon mediation factor $\eta(T)$ dependences on temperature are almost the same for both metals.

In Figure 9a, there are presented resistivity and corrected phonon mediation factor dependences on temperature for molybdenum, and in Figure 9b for zinc. The Debye's temperature obtained from resistivity dependence on temperature for Mo and Al is the same 410 K, and corrected phonon mediation factors are close one to other, while the residual resistivity of a given sample of molybdenum is about 10 times larger than that of aluminum. The resistivity $(\rho(T) - \rho_0)$ proportionality to T^2 for molybdenum is obtained at $T < 10$ K. The characteristics of the two-valent zinc (Figure 9b) differ from those of molybdenum (Figure 9a). It is related to the Debye's temperature of zinc 177 K, which is over two times smaller than that of Mo 410 K shifts the resistivity characteristic to the lower temperatures. Besides, the DOS $g(E_F)$ of zinc is about two times smaller than that of molybdenum. As a consequence, the phonon mediation factor linear part begins at a lower temperature, which causes the resistivity $(\rho(T) - \rho_0)$ proportionality to T^2 would be observed only at $T < 4$ K.

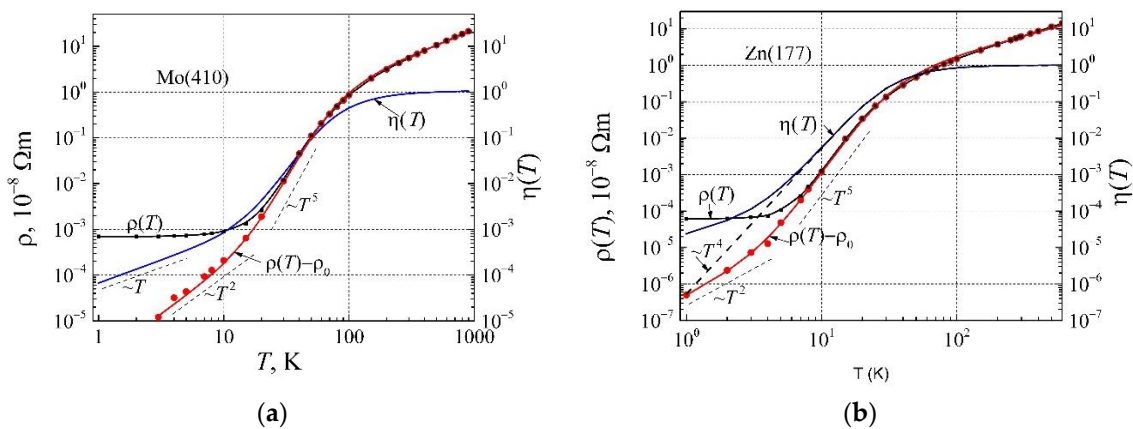


Figure 9. Resistivity (left scale), and corrected phonon mediation factor (right scale) dependences on temperature for molybdenum (a), and for zinc (b).

In Figure 10, there are presented the resistivity and the corrected phonon mediation factor dependences on temperature for copper and lithium. Considering that the presented sample of the copper resistivity data has sufficiently high residual resistivity, there was not capable to observe the T^2 dependence on temperature (Figure 10a). As it is seen from this Figure, the resistivity ($\rho(T) - \rho_0$) changes at low temperatures approximately as T^4 , the similar behavior has been observed for other noble metals by [37,40]. On the other hand, the linear part of the corrected phonon mediation factor begins only below 4 K, from this follows that the resistivity law AT^2 can be obtained at resistivity values smaller than 10 f Ω m. Maybe, it explains that the resistivity law AT^2 for noble metals is difficult to obtain. In Figure 10b, there are presented the resistivity and corrected phonon mediation factor dependences for lithium. As can be seen, the linear part of the corrected phonon mediation factor begins below 7 K, the square resistivity dependence on temperature is observed at a temperature below 7 K, and the proportionality coefficient A is approximately equal to 20 f Ω m/K².

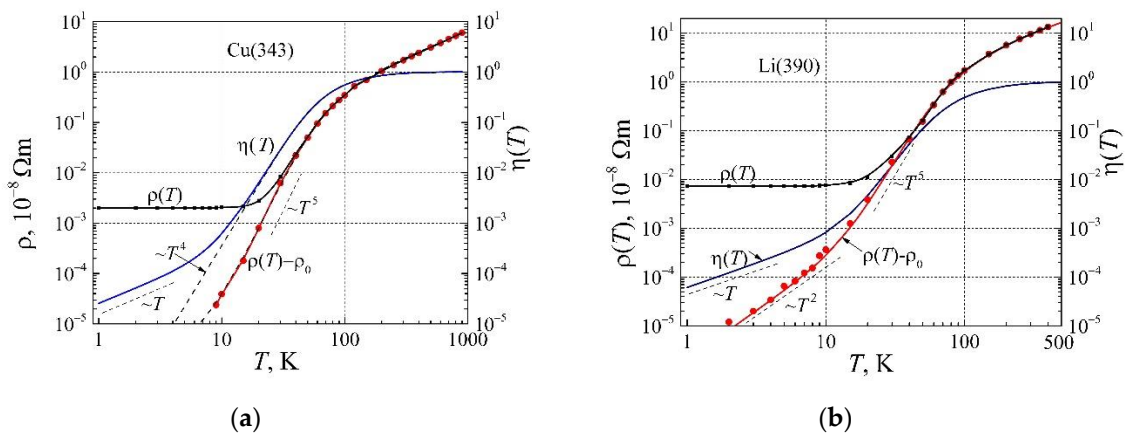


Figure 10. The resistivity (left scale) and corrected phonon mediation factor (right scale) dependences on the temperature for copper (a), and lithium (b).

As can be seen, the resistivity ($\rho(T) - \rho_0$) proportionality to T^2 at low temperatures are observed for alkali, noble, simple polyvalent or transition group metals, and temperatures at which it takes place depend on both the Debye temperature, and on the DOS at the Fermi surface of metals.

In Figure 11, there are presented the effective electron scattering cross-section dependences on temperature for the analyzed metals Al, Cu, Li, Mo, W, and Zn. As can be seen, the effective free electron scattering cross-section dependences on temperature have three characteristic parts: (i) the linear $\sim T$ at $T < 10$ K; (ii) proportional to T^4 in the transition temperature range below the Debye's temperature; and (iii) the constant at $T > 200$ K. The constant part of the effective free electron scattering cross-section shows that earlier explanations that free electron scattering cross-section at room and higher temperature increases proportional to temperature T due to lattice atom vibrations [7–11,19–22] are unfounded. The linear $\eta(T)$ increase at very low temperatures is caused by the linear heat capacity increase with temperature, which causes the quadratic resistivity of elemental metals dependence on temperature. Therefore, it shows that the earlier explanation of metal resistivity by electron-electron scattering at very low temperatures is not acceptable.

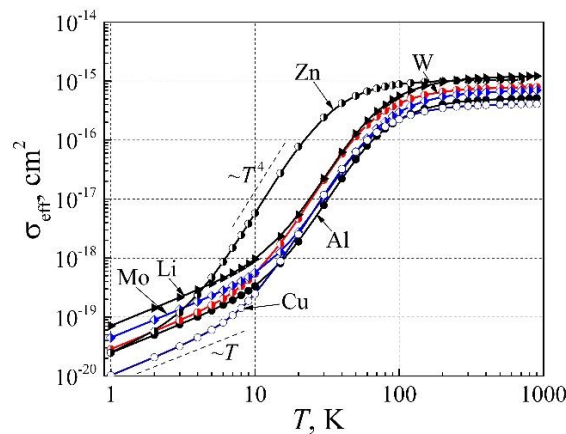


Figure 11. The effective free RM electron scattering cross-section dependences on temperature for metals Al, Cu, Li, Mo, W, and Zn.

From free RM electron effective scattering cross-section dependence on temperature follows such resistivity of elemental metals dependence on temperature characteristic parts: (i) the quadratic term AT^2 at $T < 10$ K (by eliminating the residual resistivity), (ii) the term $\sim T^5$ in the transition temperature range below the Debye's temperature, and (iii) the linear term $\sim T$ at $T > 200$ K. The whole resistivity (including the residual resistivity) temperature dependence is described by Eq. (24) accounting the corrected phonon mediation factor (Eq. (23)), i.e. the elemental metal resistivity in the overall temperature range can be explained by free randomly electron scattering by electronic defects accounting the phonon mediation factor.

4. Conclusion

Accounting that the Einstein relation between the diffusion coefficient and the drift mobility of the free randomly moving electrons in a metal is always valid, the effective electric force in metals is enhanced by a factor $\alpha_\varepsilon = E_F/kT$. Free RM electrons are scattered only by electronic defects (weakly screened ions, which lost the valence electrons), the electron wave vector during a collision with electronic defect changes very remarkably, and after each collision, electron wave vector direction is completely random. The electrical resistivity of almost all elemental metals at very low temperatures has the term AT^2 , which has been explained on the base of the Landau and Pomeranchuk Fermi liquid model by electron-electron scattering. In this work, it is shown that the electrical resistivity term AT^2 at very low temperatures is caused by free RM electron scattering by electronic defects, i.e. it is related to the linear effective scattering cross-section dependence on temperature due to free RM electrons scattering by electronic defects. The value of the parameter A depends on both the DOS at the Fermi surface and the Debye temperature. It is shown that the effective scattering cross-section of the free RM electrons by electronic defects dependence on temperature has three characteristic parts: (i) constant at $T > 200$ K, (ii) proportional to T^4 in the transition range below the Debye temperature, and (iii) the linear $\sim T$ at very low temperatures.

Author Contribution: VP is the author of this manuscript, wrote the text, performed calculations, and prepared all Figures.

Funding: The research received no external funding.

Data Availability Statement: Data are provided in the Figures of the article. All data generated and analyzed during this study are included in the presented manuscript and are available from the author upon reasonable request.

Conflict of Interest: The author declares no conflict of interest.

References

1. Drude, P. Zur Elektronentheorie der Metale. *Ann. Phys.* **1900**, 306.441 – 624.
2. Kaxiras, E. *Atomic and Electronic Structure of Solids*; Cambridge University Press (Online): Cambridge, UK, 2014.
3. Sander, L.M. *Advanced Condensed-matter Physics*; Cambridge University Press (Online): Cambridge, UK, 2014.
4. Blakemore, J. S. *Solid State Physics*; Cambridge University Press, Cambridge, UK 1985
5. Lide, D.E. (Ed.) *Handbook of Chemistry and Physics*, 84th ed.; CRC Press LLC: Boca Raton, FL, USA, 2004.
6. Grigoryev, L.S., Meilikhov, E.Z. (Eds.) *Handbook of the Physical Quantities*; Energoatomizdat: Moscow, Russia, 1991.
7. Mizutani, U. *Introduction to the Electron Theory of Metals*; Cambridge University Press (Online): Cambridge, UK, 2014.
8. Dugdale, J.S. *The Electrical Properties of Disordered Metals*; Cambridge University Press (Online): Cambridge, UK, 1985.
9. Ashcroft, N.W.; Mermin, N.D. *Solid State Physics*; Harcourt College Publ.: New York, NY, USA, 1976.
10. Ziman, J.M. *The Theory of Transport Phenomena in Solids*; Oxford University Press: Oxford, UK, 2001.
11. Kittel, C. *Introduction to Solid State Physics*; John Wiley and Sons, Inc.: New York, NY, USA, 1976.
12. Blokhintsev, D.L. *Quantum Mechanics*; Reidel Publ. Comp.: Dordrecht, The Netherlands, 1964.
13. Palenskis, V.; Žitkevičius, E. Summary of new insight into electron transport in metals. *Crystals* **2021**, 11, 622.
14. Palenskis, V.; Jonkus, V. Study of the free randomly moving electron transport peculiarities in metals. *Metals* **2023**, 13, 1551.
15. Palenskis, V. Drift mobility, diffusion coefficient of randomly moving charge carriers in metals and other materials with degenerate electron gas. *World J. Cond. Matt. Phys.* **2013**, 3, 73–81.
16. Palenskis, V. The effective density of randomly moving electrons and related characteristics of materials with degenerate electron gas. *AIP Adv.* **2014**, 4, 047119.
17. Stirzaker, D.R. *Elementary Probability*; Cambridge University Press (Online): Cambridge, UK, 2014.
18. Grimmet, G.R.; Stirzaker, D.R. *Probability and Random Processes*; Cambridge University Press (Online): Cambridge, UK, 2020.
19. Abrikosov, A.A. *Fundamentals of the Theory of Metals*; North-Holland Publication: Amsterdam, The Netherlands, 1988.
20. Lundstrom, M. *Fundamentals of Carrier Transport*; Cambridge University Press (Online): Cambridge, UK, 2014.
21. Rossiter, P.L. *The Electrical Resistivity of Metals and Alloys*; Cambridge University Press (Online): Cambridge, UK, 2014.
22. Schulze, G.E.R. *Metallphysik*; Akademie-Verlage: Berlin, Germany, 1967.
23. Landau, L.D.; Pomeranchuk, I.Ya. On properties at very low energies. *Zh. Eksp. Teor. Phys.* **1937**, 7, 379 (*Sov. Phys. – JETP* **1937**, 7, 379).
24. Kaveh, M.; Wiser, N. Electron-electron scattering in conducting materials. *Adv. Phys.* **1984**, 33, 257–372.
25. Schulz, W.W.; Allen, P.A. Transport in metals with electron-electron scattering. *Phys. Rev. B*, **1995**, **52**, 7994.
26. Arulsamy, A.D. Low-temperature small-angle electron-electron scattering rate in Fermi metals, *Zeitschrift für Naturforschung A*, **2024**, 79(1), 83-99.
27. Potter, C.; Morgan, G.J. Electron-electron scattering effects in metals. *J. Phys. F: Met. Phys.* **1979**, 9(3), 493.
28. Van Kampen, H.; Neyenhuisen, H.W.; Ribot, J.H.J.M. Semiautomatic bridge for high-precision dc resistance measurements on pure metals at low temperatures. *Rev. Sci. Instrum.* **1979**, 50, 161–164.
29. Zwart, J.W.; Pratt, W.P.; Schroeder, P.A.; Caplic, A.D. High-precision measurements of the resistivity of strained dilute copper-silver alloys below 1 K. *J. Phys.: Met. Phys.* **1983**, 13, 2595-2602.
30. Barnard, B.R.; Caplin, A.D. A simple high-resolution cryogenic resistance bridge using a SQUID null detector. *J. Phys. E: Sci. Instrum.* **1978**, 11, 1117.
31. Kaveh, M.; Wiser, N. General theory of the electrical resistivity of the alkali metals at low temperatures. *Phys. Rev. B* **1974**, 9, 4042.
32. MacDonald, A.H.; Taylor, R.; Geldart, D. J. W., Umklapp electron-electron scattering and the low-temperature electrical resistivity of the alkali metals. *Phys. Rev. B* **1981**, 23, 2718.
33. Kaveh, M.; Wiser, N. Electrical resistivity of potassium at low temperatures. *Phys. Rev. B* **1974**, 9, 4053.

34. Van Kampen, H.; Ribot, J.H.J.M.; Wyder, P. The electrical resistivity of potassium at low temperatures. *J. Phys.: Met. Phys.* **2000**, *11*(3), 597.
35. Awasthi, O.N.; Pundhir, V.K. Electron-electron interactions and the electrical resistivity of lithium at low temperatures. *Pramana J. Phys.* **2007**, *68*, 67–74.
36. Sinvani, M.; Greenfield, A.; Danino, M.; Kaveh, M.; Wiser, N. Anomalous electron-electron scattering contribution to the electrical resistivity of lithium. *J Phys. F: Met. Phys.* **1981**, *11*(4), L73.
37. Sathish, S.; Awasthi, O.N. Electron-electron scattering and low-temperature electrical resistivity in copper and silver. *Phys. Lett. A*, **1984**, *100*(4), 215–217.
38. Thummes, G.; Kötzler, J. Magnetic-field-enhanced electron-electron scattering in the resistivity of copper. *Phys. Rev. B*, **1985**, *31*, 2535.
39. Gasparov, V.A.; Harutunian, m.H. To the question on electron-electron scattering in metals. *Solid State Commun.* **1976**, *19*(3), 189–192.
40. Bergmann, A.; Kaveh, M.; Wiser, N. Explanation of the anomalous T^4 behaviour of the low-temperature electrical resistivity of silver. *J. Phys. F: Met. Phys.* **1980**, *10*, L71.
41. Kaveh, M.; Wiser, N. Evidence for the electron-electron scattering contribution to the electrical resistivity of aluminum. *Phys. Lett. A* **1975**, *51*(2), 89–90.
42. Sambles, J.R.; Elsom, K.C.; Sharp-Dent, G. The effect of sample thickness on the resistivity of aluminum. *J. Phys. F: Met. Phys* **1981**, *11* 1075.
43. Ribot, J.H.J.M.; Bass, J.; Van Kampen, H.; Wyder, P. Further evidence for electron-electron scattering in aluminum. *J. Phys. F: Met. Phys.* **1981**, *9*(6), L117.
44. MacDonald, A.H. Electron-phonon enhancement of electron-electron scattering in Al. *Phys. Rev. Lett.* **1980**, *44*, 489.
45. Volkenshtein, N.V.; Novoselov, V.A.; Startsev, V.E. Role of interelectron collisions in the electric resistance of transition metals. *Sov. Phys. JETP*, **1971**, *33*(3), 58–587.
46. Van der Mass, J.; Huguenin, R.; Gasparov, V.A. Electron-electron scattering in tungsten. *J. Phys. F: Met. Phys.* **1985**, *15*(11), L271.
47. Gasparov, V.A.; Voloshin, I.F.; Fisher, L.M. On electron-electron scattering in molybdenum. *Solid State Commun.* **1979**, *29*(1), 43–46.
48. Gautron, G.J.; Zablocki, J.E.; Hsiang, T.Y.; Weinstock, H.; Schmidt, F.A. Electron-electron scattering in vanadium. *J. Low Temp. Phys.* **1982**, *49*, 185–191.
49. Potter, C.; Morgan, G.J. Electron-electron scattering effects in metals. *J. Phys. F: Met. Phys.* **1979**, *9*(3), 493.
50. Elefant, D. Limits of Ohm's law in ultra-pure metals with almost ideal crystal lattices. *J. Appl. Phys.* **2023**, *134*, 205102.
51. Nyquist, H. Thermal agitation of electric charge in conductors. *Phys. Rev.* **1928**, *32*, 110–113.
52. Kubo, R. Statistical-mechanical theory of irreversible processes. I. General theory and simple applications to magnetic and conduction problems. *J. Phys. Soc. Jap.* **1957**, *12*, 570–586.
53. Kubo, R.A. General expression for the conductivity tensor. *J. Phys. (Canada)* **1956**, *34*, 1274–1277
54. Lide, D.R. (Ed.) *Handbook of Chemistry and Physics*, 84th ed.; CRC Press LLC: Boca Raton, FL, USA, 2004.

Disclaimer/Publisher's Note: The statements, opinions and data contained in all publications are solely those of the individual author(s) and contributor(s) and not of MDPI and/or the editor(s). MDPI and/or the editor(s) disclaim responsibility for any injury to people or property resulting from any ideas, methods, instructions or products referred to in the content.