

## Supplementary Material

# Detection of tert-butylhydroquinone in edible oils using electrochemical sensor based on nickel-aluminium layered double hydroxide@carbon spheres derived carbon composite

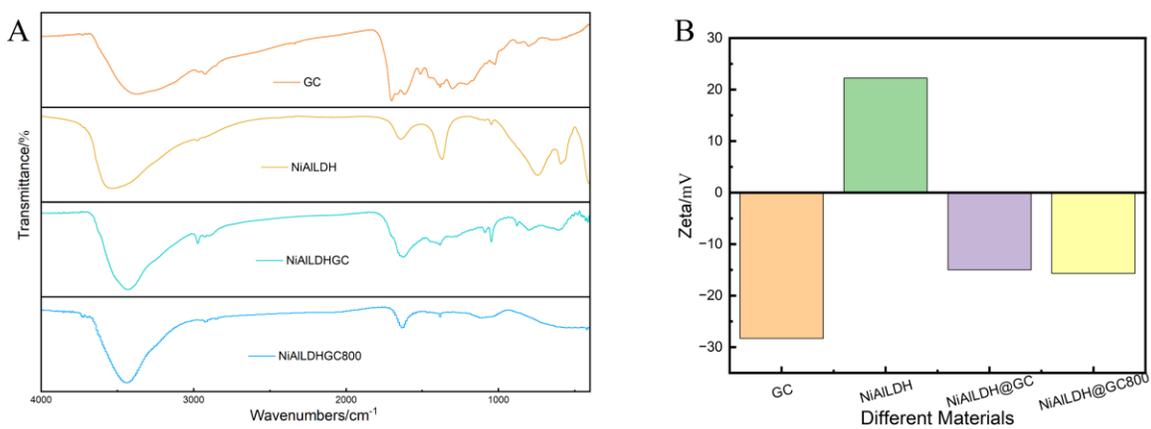
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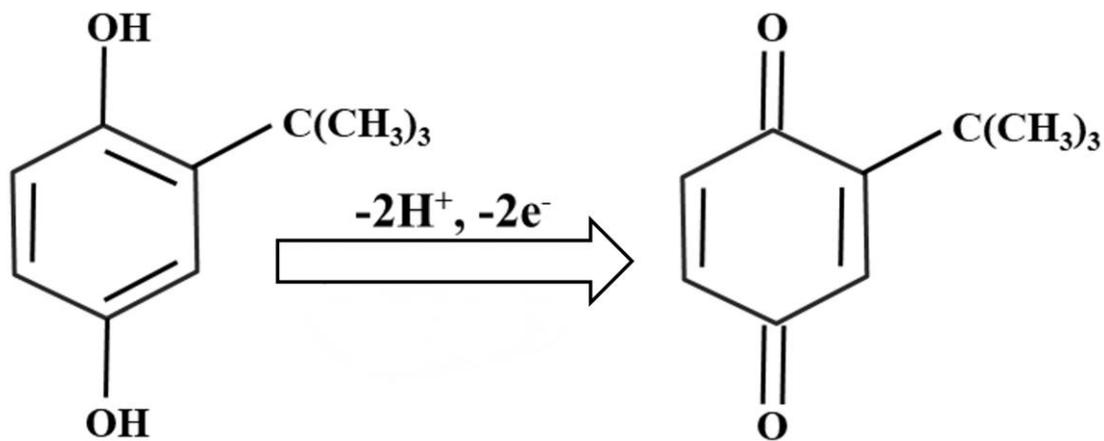
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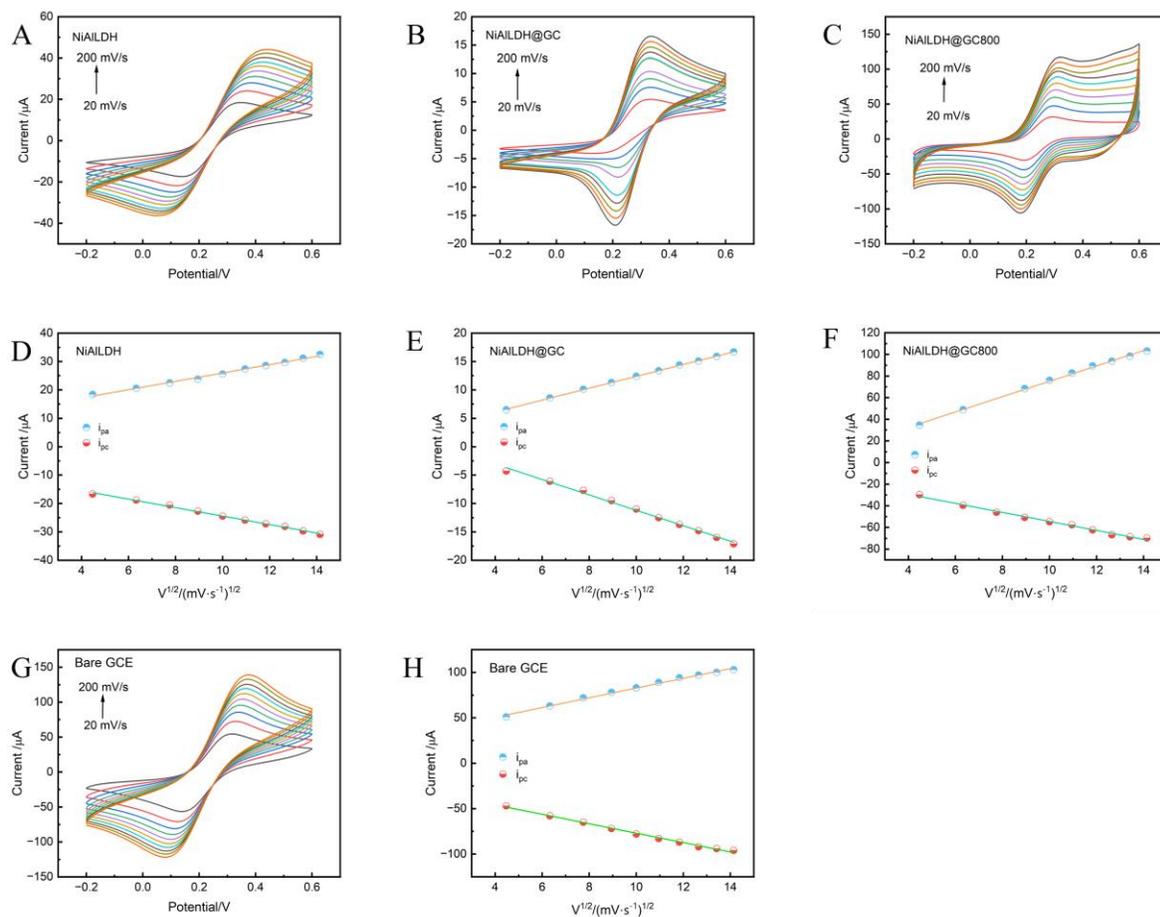
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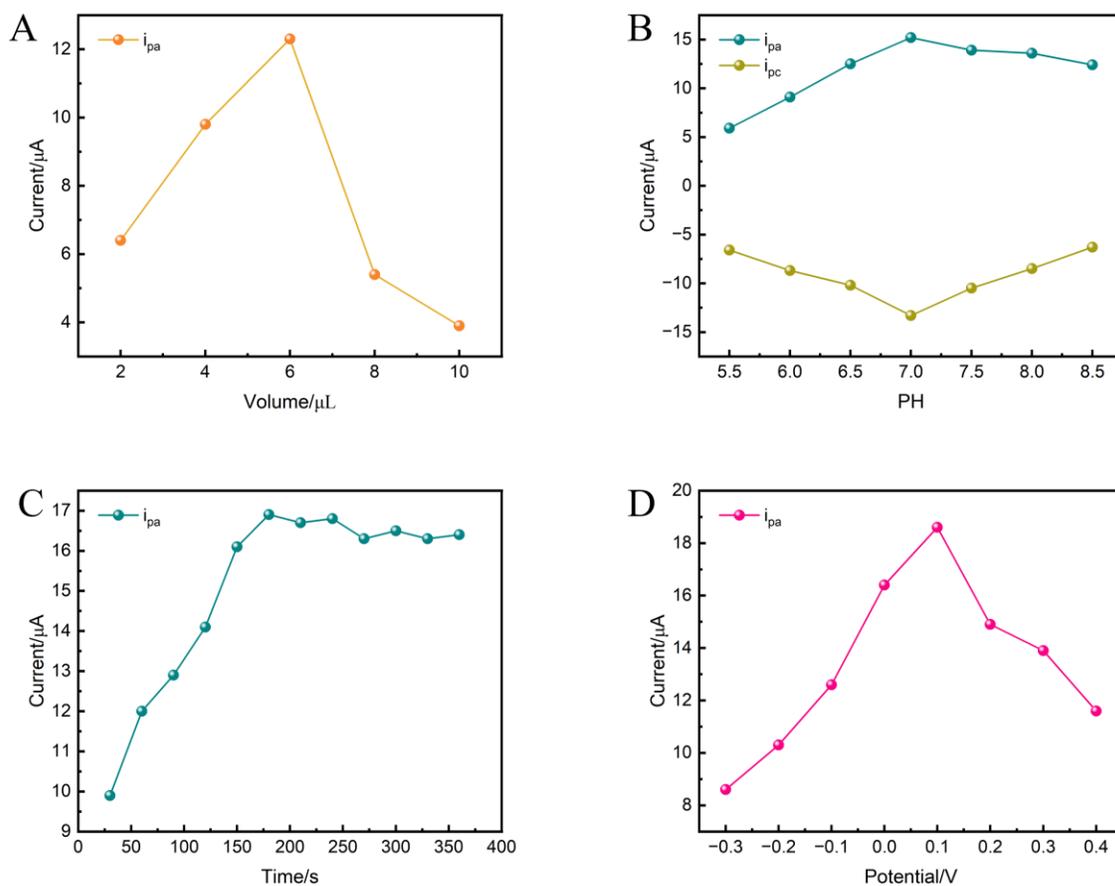
**Figure S1.** (A) FTIR spectra of GC, NiAl-LDH, NiAl-LDH@GC, and NiAl-LDH@GC800. (B) Zeta data of GC, NiAl-LDH, NiAl-LDH@GC, and NiAl-LDH@GC800.



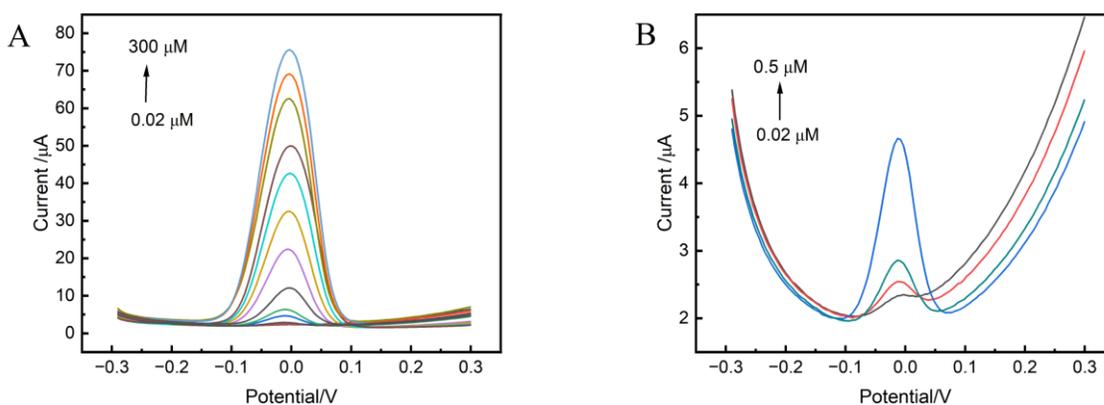
**Figure S2.** (A) Possible redox processes of TBHQ on the NiAl-LDH@GC800 sensor.



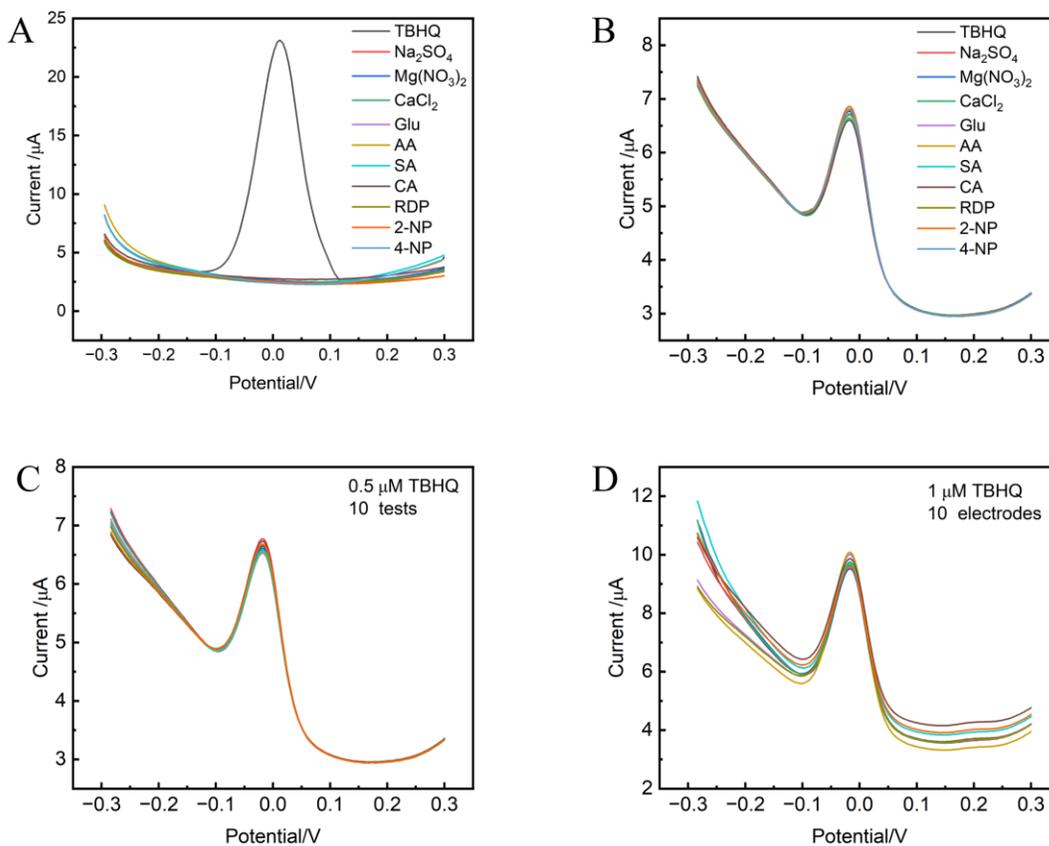
**Figure S3.** CVs of NiAl-LDH/GCE(A), NiAl-LDH@GC/GCE(B), NiAl-LDH @ GC800/GCE (C) and GCE (G) in a 5 mM  $[\text{Fe}(\text{CN})_6]^{3-/4-}$  probe containing 0.1 M KCl at different scan rates. (D, E, F, H) Oxidation peak current versus scan rate for each modified electrode.



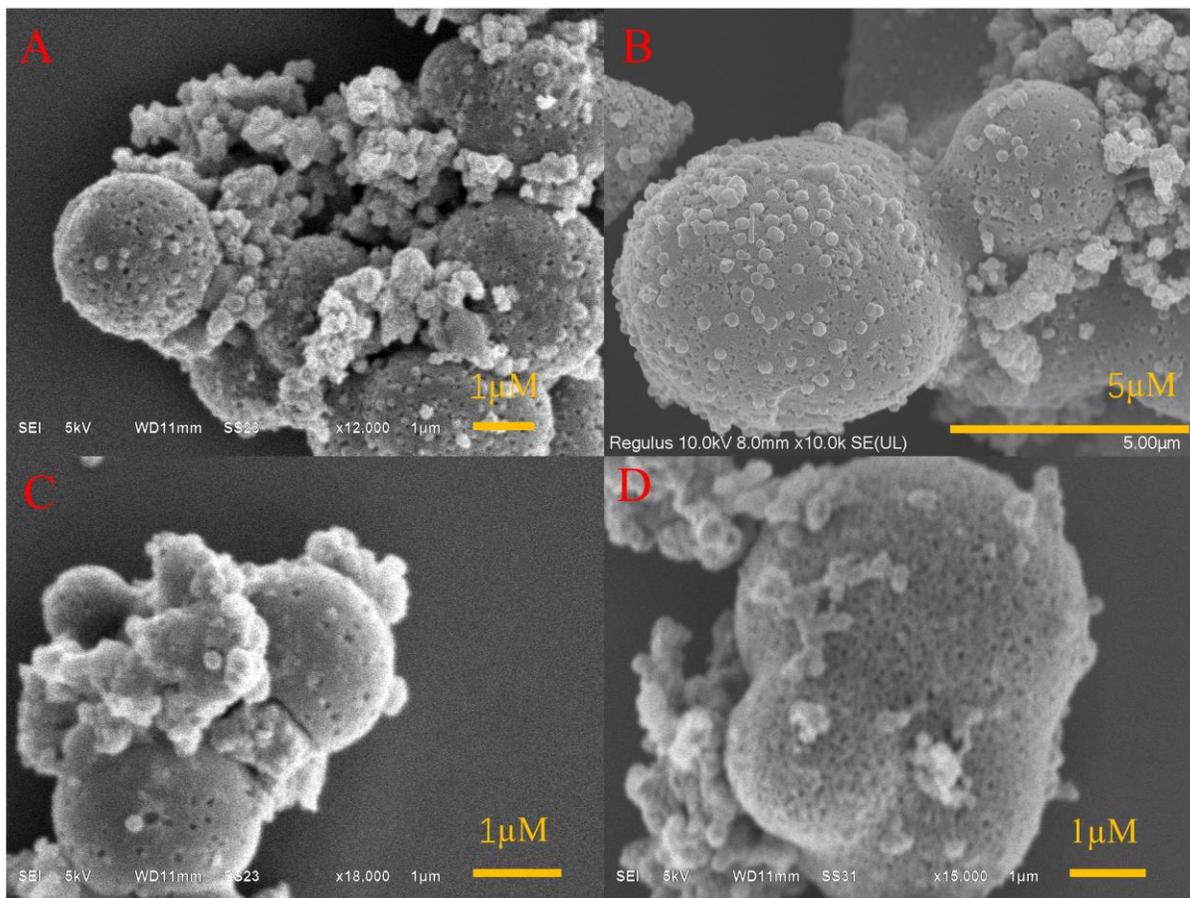
**Figure S4.** (A) Peak oxidation currents of NiAl-LDH@GC800/GCE in 0.1 M PBS (pH=7) containing  $5 \times 10^{-6}$  M TBHQ at different droplet amounts. (B) the oxidation peak currents of 0.1 M PBS with different pH values. (C) the influence of enrichment time on oxidation peak current. (D) the influence of enrichment potential on oxidation peak current.



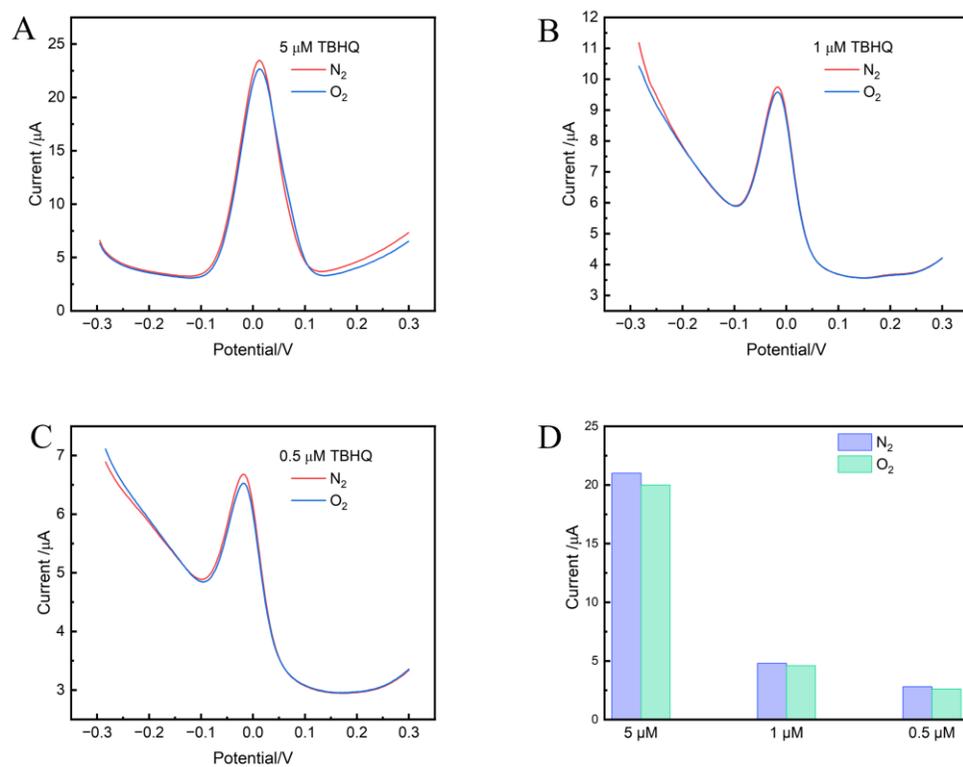
**Figure S5.** DPV responses of different TBHQ concentrations (0.02  $\mu\text{M}$ , 0.05  $\mu\text{M}$ , 0.1  $\mu\text{M}$ , 0.5  $\mu\text{M}$ , 1  $\mu\text{M}$ , 3  $\mu\text{M}$ , 5  $\mu\text{M}$ , 50  $\mu\text{M}$ , 100  $\mu\text{M}$ , 150  $\mu\text{M}$ , 200  $\mu\text{M}$ , 250  $\mu\text{M}$ , 300  $\mu\text{M}$ ).



**Figure S6.** (A) DPV curves of different interfering substances in 0.1 M PBS solution (without TBHQ). (B) DPV curves of TBHQ (0.5  $\mu\text{M}$ ) in PBS containing different interfering substances. (C) DPV curves of TBHQ (0.5  $\mu\text{M}$ ) in PBS (10 tests). (D) DPV curves of TBHQ (1  $\mu\text{M}$ ) in PBS (10 electrodes).



**Figure S7.** (A-B) SEM images of NiAl-LDH@GC800 before electrochemical measurements. (C-D) SEM images of NiAl-LDH@GC800 after electrochemical measurements.



**Fig.S8.** Comparison of detection in nitrogen and oxygen environments (A) 5 μM TBHQ. (B) 1 μM TBHQ. (C) 0.5 μM TBHQ. (D) Oxidation peak current comparison.

**Table S1.** CV data for modified electrodes in a 5.0 mM [Fe(CN)<sub>6</sub>]<sup>3-/4-</sup>.

Modified electrode	I <sub>pa</sub> (μA)	I <sub>pc</sub> (μA)	E <sub>pa</sub> (V)	E <sub>pc</sub> (V)	ΔE <sub>p</sub> (V)
NiAl-LDH	25.63	-23.71	0.372	0.091	0.281
NiAl-LDH@GC	12.51	-12.19	0.313	0.186	0.127
NiAl-LDH@GC800	76.73	-75.84	0.304	0.191	0.113
GCE	58.71	-53.16	0.276	0.151	0.125

**Table S2.** Comparative evaluation of the performance of NiAl-LDH@GC800/GCE with other modified electrodes in TBHQ detection.

Modified electrode	Detection technique	LOD ( $\mu\text{M}$ )	Linear range ( $\mu\text{M}$ )	References
MIP-MWCNT/GCE	DPV	0.85	2.84-150	(Santos Moretti et al., 2016)
MIP/MoS <sub>2</sub> /EACC	DPV	0.00072	0.001-120	(Chi et al., 2024)
MIP/AuNPs/EGP	DPV	0.07	0.08-100	(Fan et al., 2018)
FeNi <sub>3</sub> /rGO/HMPF <sub>6</sub> /CPE	SWV	0.01	0.05-900	(Tahernejad-Javazmi et al., 2019)
PVP-CTAB/Au-PVP-Gr/GCE	DPV	0.009	0.02-0.1, 0.1-100	(Wang et al., 2016)
ZnCuMg TMO/ $\beta$ -CD-CB/SPCE	DPV	0.001	0.031-12.56, 12.56-118.8	(Sebastian, et al., 2022)
ZnO TPHS@GO/GCE	SWV	0.137	0.8-65	(Gan et al., 2016)
Co NC/CNT/MB/ GCE	DPV	0.054	0.1-20, 20-100	(Zhang et al., 2024)
NiAl-LDH@GC800/GCE	DPV	0.0082	0.02-5, 5-300	This work

**Table S3.** Detection of TBHQ in Different Edible Oils (n=3)

sample	recruitment (nM)	Discovery quantity (nM)	recovery rate (%)	RSD (%)	UV-Vis (test)	UV-Vis (dilution)
chili oil	0	105	—	—	21.85 $\mu$ M	109.2 nM
	40	143.8	97.0	1.7		
	80	188	103.7	2.3		
	120	223	102.5	2.1		
peanut oil	0	174	—	—	34.26 $\mu$ M	171.3 nM
	50	225.7	103.4	2.6		
	100	276.2	102.2	1.9		
	150	329.6	103.6	2.3		
rap oil	0	210.3	—	—	41.30 $\mu$ M	206.5 nM
	50	259.4	98.2	2.2		
	100	311.3	101.0	1.5		
	150	363.6	102.2	2.0		

Relative standard deviation (RSD) is as follows:

$$RSD = \frac{SD}{\bar{x}} \times 100\% = \frac{\sqrt{\frac{\sum_{i=1}^n (x_i - \bar{x})^2}{n-1}}}{\bar{x}} \times 100\%, \text{ where SD is the standard deviation and } \bar{x} \text{ is the mean.}$$